

# Climate Change 2007

## The Physical Science Basis

The Working Group I contribution to the IPCC Fourth Assessment Report  
Summary for Policymakers, Technical Summary and Frequently Asked Questions

### ERRATA

Page	Item	Correction
33	<b>Technical Summary</b>	Table TS.2 A number of species were inadvertently omitted that should have been included in the original table {Table 2.14}. Minor typographical errors (unit, superscripts, and footnotes) have also been corrected. Please see end of this Errata for the complete table {Table 2.14 Errata}.
38	<b>Technical Summary</b>	Figure TS.7, Panel A The bottom left label of the y-axis should be -1.0 and not -0.1 as given.
100	<b>Frequently Asked Question 2.1</b>	FAQ 2.1, Figure 1 The red label on the graph should read: "Carbon Dioxide"
100	<b>Frequently Asked Question 2.1</b>	FAQ 2.1, Figure 1 The CH <sub>4</sub> curve was incorrectly plotted. Please see end of this Errata for the revised figure {FAQ 2.1, Figure 1 Errata}
116	<b>Frequently Asked Question 7.1</b>	FAQ 7.1, Figure 1 In Panel (a) the "Land-Based Sink" and "Net Oceanic Sink" labels should be reversed.
116	<b>Frequently Asked Question 7.1</b>	FAQ 7.1, Figure 1 In Panel (d) the "Human-caused Sources" bar should indicate a value of 6.7 and an uncertainty range of 3.0-11.7.
117	<b>Frequently Asked Question 8.1</b>	FAQ 8.1, Figure 1 The caption should read: "Global mean near-surface temperature anomalies..."
132	<b>Glossary</b>	Dobson unit (DU) The definition of Dobson unit should read: "... $2.69 \times 10^{20}$ molecules per square metre"
133	<b>Glossary</b>	Emission scenario: The definition of Emission scenario should read: "(Nakićenović and Swart, 2000)".
140	<b>Glossary</b>	SRES scenarios, Illustrative Scenario, Marker Scenario: The definitions should read: "(Nakićenović and Swart, 2000)"

**Table 2.14 (Errata).** Lifetimes, radiative efficiencies and direct (except for  $\text{CH}_4$ ) GWPs relative to  $\text{CO}_2$ . For ozone-depleting substances and their replacements, data are taken from IPCC/TEAP (2005) unless otherwise stated. See IPCC AR4 (Forster et al., 2007; Section 2.10.2 and Table 2.14) for details. A number of species were inadvertently omitted that should have been included in the list, and the complete table appears below. Information on the GWPs of these species were included in IPCC TAR (Ramaswamy et al., 2001; Tables 6.7 and 6.8). These species are now included in this Errata to Table 2.14 of IPCC AR4 (Forster et al., 2007), following established procedures and precedents.  $\text{CO}_2$  AGWP values from IPCC AR4 (Forster et al., 2007; Section 2.10.2), and estimates of the lifetimes and radiative efficiency of these species (based on TAR and updates from WMO (2002, Chapter 1)), are employed to obtain their GWPs. Estimates of GWPs from SAR<sup>‡</sup> are also listed for reference. Minor typographical errors (unit, parenthesis, superscripts, and footnotes) have also been corrected in this Errata.

Industrial Designation or Common Name	Chemical Formula	Lifetime (years)	Radiative Efficiency ( $\text{W m}^{-2} \text{ ppb}^{-1}$ )	Global Warming Potential for Given Time Horizon			
				SAR <sup>‡</sup> (100-yr)	20-yr	100-yr	500-yr
Carbon dioxide	$\text{CO}_2$	See below <sup>a</sup>	<sup>b</sup> $1.4 \times 10^{-5}$	1	1	1	1
Methane <sup>c</sup>	$\text{CH}_4$	12 <sup>c</sup>	$3.7 \times 10^{-4}$	21	72	25	7.6
Nitrous oxide	$\text{N}_2\text{O}$	114	$3.03 \times 10^{-3}$	310	289	298	153
<b>Substances controlled by the Montreal Protocol</b>							
CFC-11	$\text{CCl}_3\text{F}$	45	0.25	3,800	6,730	4,750	1,620
CFC-12	$\text{CCl}_2\text{F}_2$	100	0.32	8,100	11,000	10,900	5,200
CFC-13	$\text{CClF}_3$	640	0.25		10,800	14,400	16,400
CFC-113	$\text{CCl}_2\text{FCClF}_2$	85	0.3	4,800	6,540	6,130	2,700
CFC-114	$\text{CClF}_2\text{CClF}_2$	300	0.31		8,040	10,000	8,730
CFC-115	$\text{CClF}_2\text{CF}_3$	1,700	0.18		5,310	7,370	9,990
Halon-1301	$\text{CBrF}_3$	65	0.32	5,400	8,480	7,140	2,760
Halon-1211	$\text{CBrClF}_2$	16	0.3		4,750	1,890	575
Halon-2402	$\text{CBrF}_2\text{CBrF}_2$	20	0.33		3,680	1,640	503
Carbon tetrachloride	$\text{CCl}_4$	26	0.13	1,400	2,700	1,400	435
Methyl bromide	$\text{CH}_3\text{Br}$	0.7	0.01		17	5	1
Methyl chloroform	$\text{CH}_3\text{CCl}_3$	5	0.06	100*	506	146	45
HCFC-21	$\text{CHCl}_2\text{F}$	1.7	0.14		530	151	46
HCFC-22	$\text{CHClF}_2$	12	0.2	1,500	5,160	1,810	549
HCFC-123	$\text{CHCl}_2\text{CF}_3$	1.3	0.14	90	273	77	24
HCFC-124	$\text{CHClFCF}_3$	5.8	0.22	470	2,070	609	185
HCFC-141b	$\text{CH}_3\text{CCl}_2\text{F}$	9.3	0.14	600	2,250	725	220
HCFC-142b	$\text{CH}_3\text{CClF}_2$	17.9	0.2	1,800	5,490	2,310	705
HCFC-225ca	$\text{CHCl}_2\text{CF}_2\text{CF}_3$	1.9	0.2		429	122	37
HCFC-225cb	$\text{CHClFCF}_2\text{CClF}_2$	5.8	0.32		2,030	595	181
<b>Hydrofluorocarbons</b>							
HFC-23	$\text{CHF}_3$	270	0.19	11,700	12,000	14,800	12,200
HFC-32	$\text{CH}_2\text{F}_2$	4.9	0.11	650	2,330	675	205
HFC-41	$\text{CH}_3\text{F}$	2.4	0.02	150	323	92	28
HFC-125	$\text{CHF}_2\text{CF}_3$	29	0.23	2,800	6,350	3,500	1,100
HFC-134	$\text{CHF}_2\text{CHF}_2$	9.6	0.18	1000	3,400	1,100	335
HFC-134a	$\text{CH}_2\text{FCF}_3$	14	0.16	1,300	3,830	1,430	435
HFC-143	$\text{CH}_2\text{FCHF}_2$	3.5	0.13	300	1,240	353	107
HFC-143a	$\text{CH}_3\text{CF}_3$	52	0.13	3,800	5,890	4,470	1,590
HFC-152	$\text{CH}_2\text{FCH}_2\text{F}$	0.60	0.09		187	53	16
HFC-152a	$\text{CH}_3\text{CHF}_2$	1.4	0.09	140	437	124	38
HFC-161	$\text{CH}_3\text{CH}_2\text{F}$	0.3	0.03		43	12	3.7
HFC-227ea	$\text{CF}_3\text{CHFCF}_3$	34.2	0.26	2,900	5,310	3,220	1,040
HFC-236cb	$\text{CH}_2\text{FCF}_2\text{CF}_3$	13.6	0.23		3,630	1,340	407
HFC-236ea	$\text{CHF}_2\text{CHFCF}_3$	10.7	0.3		4,090	1,370	418
HFC-236fa	$\text{CF}_3\text{CH}_2\text{CF}_3$	240	0.28	6,300	8,100	9,810	7,660
HFC-245ca	$\text{CH}_2\text{FCF}_2\text{CHF}_2$	6.2	0.23	560	2,340	693	211
HFC-245fa	$\text{CHF}_2\text{CH}_2\text{CF}_3$	7.6	0.28		3,380	1,030	314
HFC-365mfc	$\text{CH}_3\text{CF}_2\text{CH}_2\text{CF}_3$	8.6	0.21		2,520	794	241
HFC-43-10mee	$\text{CF}_3\text{CHFCHFCF}_2\text{CF}_3$	15.9	0.4	1,300	4,140	1,640	500

Table 2.14 (continued)

Industrial Designation or Common Name	Chemical Formula	Lifetime (years)	Radiative Efficiency ( $\text{W m}^{-2} \text{ ppb}^{-1}$ )	Global Warming Potential for Given Time Horizon			
				SAR <sup>‡</sup> (100-yr)	20-yr	100-yr	500-yr
<b>Perfluorinated compounds</b>							
Sulphur hexafluoride	$\text{SF}_6$	3,200	0.52	23,900	16,300	22,800	32,600
Nitrogen trifluoride	$\text{NF}_3$	740	<sup>d</sup> 0.21		12,300	17,200	20,700
PFC-14	$\text{CF}_4$	50,000	<sup>e</sup> 0.10	6,500	5,210	7,390	11,200
PFC-116	$\text{C}_2\text{F}_6$	10,000	0.26	9,200	8,630	12,200	18,200
PFC-218	$\text{C}_3\text{F}_8$	2,600	0.26	7,000	6,310	8,830	12,500
PFC-318	c-C <sub>4</sub> F <sub>8</sub>	3,200	0.32	8,700	7,310	10,300	14,700
PFC-3-1-10	$\text{C}_4\text{F}_{10}$	2,600	0.33	7,000	6,330	8,860	12,500
PFC-4-1-12	$\text{C}_5\text{F}_{12}$	4,100	0.41	7,500	6,510	9,160	13,300
PFC-5-1-14	$\text{C}_6\text{F}_{14}$	3,200	0.49	7,400	6,600	9,300	13,300
PFC-9-1-18	$\text{C}_{10}\text{F}_{18}$	>1,000 <sup>f</sup>	0.56		>5,500	>7,500	>9,500
trifluoromethyl sulphur pentafluoride	$\text{SF}_5\text{CF}_3$	800	0.57		13,200	17,700	21,200
Perfluorocyclopropane	c-C <sub>3</sub> F <sub>6</sub>	>1000	0.42		>12,700	>17,340	>21,800
<b>Fluorinated ethers</b>							
HFE-125	$\text{CHF}_2\text{OCF}_3$	136	0.44		13,800	14,900	8,490
HFE-134	$\text{CHF}_2\text{OCHF}_2$	26	0.45		12,200	6,320	1,960
HFE-143a	$\text{CH}_3\text{OCF}_3$	4.3	0.27		2,630	756	230
HCFC-235da2	$\text{CHF}_2\text{OCHClCF}_3$	2.6	0.38		1,230	350	106
HFE-245cb2	$\text{CH}_3\text{OCF}_2\text{CF}_3$	5.1	0.32		2,440	708	215
HFE-245fa2	$\text{CHF}_2\text{OCH}_2\text{CF}_3$	4.9	0.31		2,280	659	200
HFE-254cb2	$\text{CH}_3\text{OCF}_2\text{CHF}_2$	2.6	0.28		1,260	359	109
HFE-347mcc3	$\text{CH}_3\text{OCF}_2\text{CF}_2\text{CF}_3$	5.2	0.34		1,980	575	175
HFE-347pcf2	$\text{CHF}_2\text{CF}_2\text{OCH}_2\text{CF}_3$	7.1	0.25		1,900	580	175
HFE-356pcc3	$\text{CH}_3\text{OCF}_2\text{CF}_2\text{CHF}_2$	0.33	0.93		386	110	33
HFE-449sl (HFE-7100)	$\text{C}_4\text{F}_9\text{OCH}_3$	3.8	0.31		1,040	297	90
HFE-569sf2 (HFE-7200)	$\text{C}_4\text{F}_9\text{OC}_2\text{H}_5$	0.77	0.3		207	59	18
HFE-43-10pccc124 (H-Galden 1040x)	$\text{CHF}_2\text{OCF}_2\text{OC}_2\text{F}_4\text{OCHF}_2$	6.3	1.37		6,320	1,870	569
HFE-236ca12 (HG-10)	$\text{CHF}_2\text{OCF}_2\text{OCHF}_2$	12.1	0.66		8,000	2,800	860
HFE-338pcc13 (HG-01)	$\text{CHF}_2\text{OCF}_2\text{CF}_2\text{OCHF}_2$	6.2	0.87		5,100	1,500	460
	$(\text{CF}_3)_2\text{CFOCH}_3$	3.4	0.31		1204	343	104
	$\text{CF}_3\text{CF}_2\text{CH}_2\text{OH}$	0.4	0.24		147	42	13
	$(\text{CF}_3)_2\text{CHOH}$	1.8	0.28		687	195	59
HFE-227ea	$\text{CF}_3\text{CHFOCF}_3$	11	0.40		4,540	1,540	468
HFE-236ea2	$\text{CHF}_2\text{OCHFCF}_3$	5.8	0.44		3,370	989	301
HFE-236fa	$\text{CF}_3\text{CH}_2\text{OCF}_3$	3.7	0.34		1,710	487	148
HFE-245fa1	$\text{CHF}_2\text{CH}_2\text{OCF}_3$	2.2	0.30		1,010	286	87
HFE-263fb2	$\text{CF}_3\text{CH}_2\text{OCH}_3$	0.2	0.1		38	11	3
HFE-329mcc2	$\text{CHF}_2\text{CF}_2\text{OCF}_2\text{CF}_3$	6.8	0.49		3,060	919	279
HFE-338mcf2	$\text{CF}_3\text{CH}_2\text{OCF}_2\text{CF}_3$	4.3	0.43		1,920	552	168
HFE-347mcf2	$\text{CHF}_2\text{CH}_2\text{OCF}_2\text{CF}_3$	2.8	0.41		1,310	374	114
HFE-356mec3	$\text{CH}_3\text{OCF}_2\text{CHFCF}_3$	0.94	0.30		355	101	31
HFE-356pcf2	$\text{CHF}_2\text{CH}_2\text{OCF}_2\text{CHF}_2$	2.0	0.37		931	265	80
HFE-356pcf3	$\text{CHF}_2\text{OCH}_2\text{CF}_2\text{CHF}_2$	3.6	0.39		1,760	502	153
HFE 365mcf3	$\text{CF}_3\text{CF}_2\text{CH}_2\text{OCH}_3$	0.27	0.11		41	11	4

Table 2.14 (continued)

Industrial Designation or Common Name	Chemical Formula	Lifetime (years)	Radiative Efficiency ( $\text{W m}^{-2} \text{ ppb}^{-1}$ )	Global Warming Potential for Given Time Horizon			
				SAR <sup>a</sup> (100-yr)	20-yr	100-yr	500-yr
<b>Fluorinated ethers (continued)</b>							
HFE-374pc2	$\text{CHF}_2\text{CF}_2\text{OCH}_2\text{CH}_3$	5.0	0.25		1,930	557	169
	- $(\text{CF}_2)_4\text{CH(OH)}$ -	0.3	0.85		258	73	23
	$(\text{CF}_3)_2\text{CHOCHF}_2$	3.1	0.41		1,330	380	115
	$(\text{CF}_3)_2\text{CHOCH}_3$	0.25	0.30		94	27	8.2
<b>Perfluoropolyethers</b>							
PFPMIE	$\text{CF}_3\text{OCF}(\text{CF}_3)\text{CF}_2\text{OCF}_2\text{OCF}_3$	800	0.65		7,620	10,300	12,400
<b>Hydrocarbons and other compounds – Direct Effects</b>							
Dimethylether	$\text{CH}_3\text{OCH}_3$	0.015	0.02		1	1	<<1
Chloroform	$\text{CHCl}_3$	0.51	0.11	4	108	31	9.3
Methylene chloride	$\text{CH}_2\text{Cl}_2$	0.38	0.03	9	31	8.7	2.7
Methyl chloride	$\text{CH}_3\text{Cl}$	1.0	0.01		45	13	4
	$\text{CH}_2\text{Br}_2$	0.41	0.01		5.4	1.54	0.47
Halon-1201	$\text{CHBrF}_2$	5.8	0.14		1,380	404	123
Trifluoroiodomethane	$\text{CF}_3\text{I}$	0.005	0.23	<1	1	0.4	0.1

Notes:

<sup>a</sup> The CO<sub>2</sub> response function used in this report is based on the revised version of the Bern Carbon cycle model used in Chapter 10 of this report (Bern2.5CC; Joos et al. 2001) using a background CO<sub>2</sub> concentration value of 378 ppm. The decay of a pulse of CO<sub>2</sub> with time it is given by

$$a_0 + \sum_{i=1}^3 a_i \cdot e^{-t/\tau_i}$$

Where  $a_0 = 0.217$ ,  $a_1 = 0.259$ ,  $a_2 = 0.338$ ,  $a_3 = 0.186$ ,  $\tau_1 = 172.9$  years,  $\tau_2 = 18.51$  years, and  $\tau_3 = 1.186$  years

<sup>b</sup> The radiative efficiency of CO<sub>2</sub> is calculated using the IPCC (1990) simplified expression as revised in the TAR, with an updated background concentration value of 378 ppm and a perturbation of +1 ppm (see Section 2.10.2).

<sup>c</sup> The perturbation lifetime for methane is 12 years as in the TAR (see also Section 7.4). The GWP for methane includes indirect effects from enhancements of ozone and stratospheric water vapour (see Section 2.10.3.1).

<sup>d</sup> Robson et al. (2006)

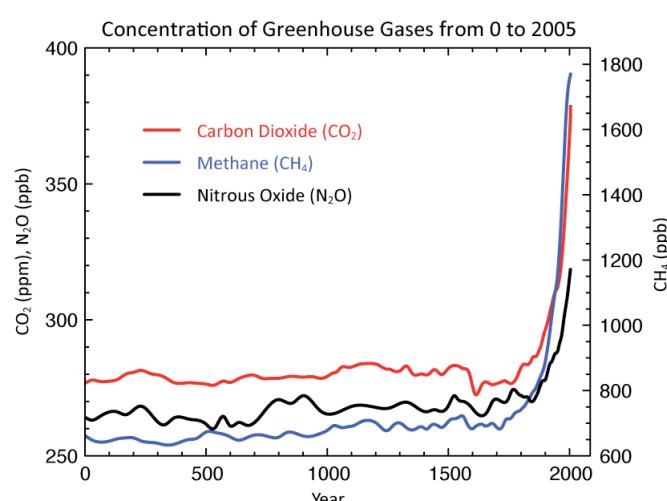
<sup>e</sup> Hurley et al. (2005)

<sup>f</sup> Shine et al. (2005c), updated by the revised AGWP for CO<sub>2</sub>. The assumed lifetime of 1,000 years is a lower limit.

<sup>g</sup> Second Assessment Report (IPCC, 1996)

<sup>\*</sup> Compound in SAR (Table 2.8) was erroneously listed as CH<sub>3</sub>Cl<sub>3</sub>.

Frequently Asked Question 2.1, Figure 1 (Errata)



**FAQ 2.1, Figure 1 (Errata).** Revised figure showing atmospheric concentrations of important long-lived greenhouse gases over the last 2,000 years. Using the combined and simplified data from Chapters 6 and 2, the original figure displayed the CH<sub>4</sub> curve incorrectly. The revised figure shows the same data correctly plotted. For further details please refer to the original figure caption.