

# Efficiency of short-lived halogens at influencing climate through depletion of stratospheric ozone

R. Hossaini<sup>1\*</sup>, M. P. Chipperfield<sup>1</sup>, S. A. Montzka<sup>2</sup>, A. Rap<sup>1</sup>, S. Dhomse<sup>1</sup>, and W. Feng<sup>1,3</sup>

1. School of Earth and Environment, University of Leeds, Leeds, UK
2. National Oceanic and Atmospheric Administration, Boulder, USA
3. National Centre for Atmospheric Science, University of Leeds, Leeds, UK

\* Corresponding author ([r.hossaini@leeds.ac.uk](mailto:r.hossaini@leeds.ac.uk))

This supplement contains 4 tables and 3 figures. Table S1 gives a summary of model experiments and their design to examine the impact of halogens from VSLs in the stratosphere. Table S2 gives the calculated global mean radiative effect caused by VSLs-driven ozone perturbations. Table S3 gives a summary of additional model experiments that were performed to examine the impact of VSLs on ozone in the troposphere. Table S4 presents observed CH<sub>2</sub>Cl<sub>2</sub> mixing ratios at 13 surface locations from the ongoing monitoring program of the National Oceanic and Atmospheric Administration (NOAA) Earth System Research Laboratory (ESRL). Table S4 also contains the site-wise CH<sub>2</sub>Cl<sub>2</sub> growth rate over the 2010-2013 period along with hemispheric and global averages. Figure S1 shows the 2011 mean simulated ozone change due to VSLs in the troposphere. Figure S2 shows a comparison between long-term ozone anomalies from the TOMCAT model and TOMS/SBUV satellite data between 1985 and 2013. Finally, Figure S3 shows column O<sub>3</sub> changes due to bromine VSLs in 2011 and also in an atmosphere with a pre-industrial stratospheric halogen load.

**Table S1. Summary of stratospheric model experiments and halogen load from VSLS.** Experiments were designed to examine the individual and combined impact of halogens from VSLS on ozone.

Experiment*	VSLS Loading [ppt] **			Comment
	Bromine (Br)	Chlorine (Cl)	Iodine (I)	
STRAT1 <sup>+</sup>	0	0	0	No VSLS, control run
STRAT2 <sup>+</sup>	6	0	0	Br - best
STRAT3 <sup>+</sup>	3	0	0	Br - lower
STRAT4 <sup>+</sup>	8	0	0	Br - upper
STRAT5	6	40	0	Br - best Cl - lower
STRAT6	6	80	0	Br - best Cl - best
STRAT7	6	130	0	Br - best Cl - upper
STRAT8	6	80	0.15	Br - best Cl - best I - upper
STRAT9 <sup>^</sup>	3	40	0	All - lower
STRAT10 <sup>^</sup>	8	130	0.15	All - upper

Notes:

\* The model was run for the 1979-2013 period at a resolution of ~5.6° longitude by ~5.6° latitude and with 32 levels from the surface to ~60 km. Meteorological forcing data was taken from the European Centre for Medium-Range Weather Forecasts (ECMWF) ERA-Interim (6-hourly) reanalysis.

\*\* Range and best estimates based on the World Meteorological Organization Scientific Assessment of Ozone Depletion 2010. Upper limit of Cl encompasses the recent CH<sub>2</sub>Cl<sub>2</sub> trend.

+ In addition, also performed with a fixed pre-industrial stratospheric halogen load comprising background CH<sub>3</sub>Br and CH<sub>3</sub>Cl only.

<sup>^</sup> Extreme ranges used to determine the uncertainty on O<sub>3</sub> changes and RE due to VSLS.

**Table S2. Radiative effect (RE) due to VSLS-driven O<sub>3</sub> perturbations.** Net RE (longwave + shortwave) reported as global mean area-weighted averages for 2011 (Wm<sup>-2</sup>). Range shown in brackets is due to uncertainty in VSLS loading.

	VSLS halogen	Global Mean RE [Wm <sup>-2</sup> ]
Stratosphere	Bromine	-0.07 (-0.035 to -0.096)
	Chlorine	-0.007 (-0.004 to -0.011)
	Iodine	< -0.003
	<b>Combined stratosphere</b>	<b>-0.08 (-0.04 to -0.11)</b>
Troposphere	Chlorine	Negligible
	Bromine & Iodine	-0.12
	<i>Bromine &amp; Iodine [ref 23]</i>	~ -0.1
	<b>Combined troposphere</b>	<b>-0.12</b>
	<b>Whole atmosphere</b>	<b>-0.20 (-0.16 to -0.23)</b>

**Table S3. Summary of tropospheric model experiments.** Experiments were designed to examine the individual and combined impact of halogens from VSLS on ozone.

Experiment*	Inclusion of VSLS**			Comment
	Bromine (Br)	Chlorine (Cl)	Iodine (I)	
<b>TROP1</b>	No	No	No	No VSLS
<b>TROP2</b>	Yes	No	No	Br only
<b>TROP3</b>	Yes	Yes	No	Br and Cl
<b>TROP4</b>	Yes	Yes	Yes	Br, Cl and I

Notes:

\* The model was run for the 2009 to 2013 period at a resolution of ~2.8° longitude by ~2.8° latitude and with 31 levels from the surface to ~30 km. Meteorological forcing data was taken from the ECMWF ERA-Interim (6-hourly) reanalysis.

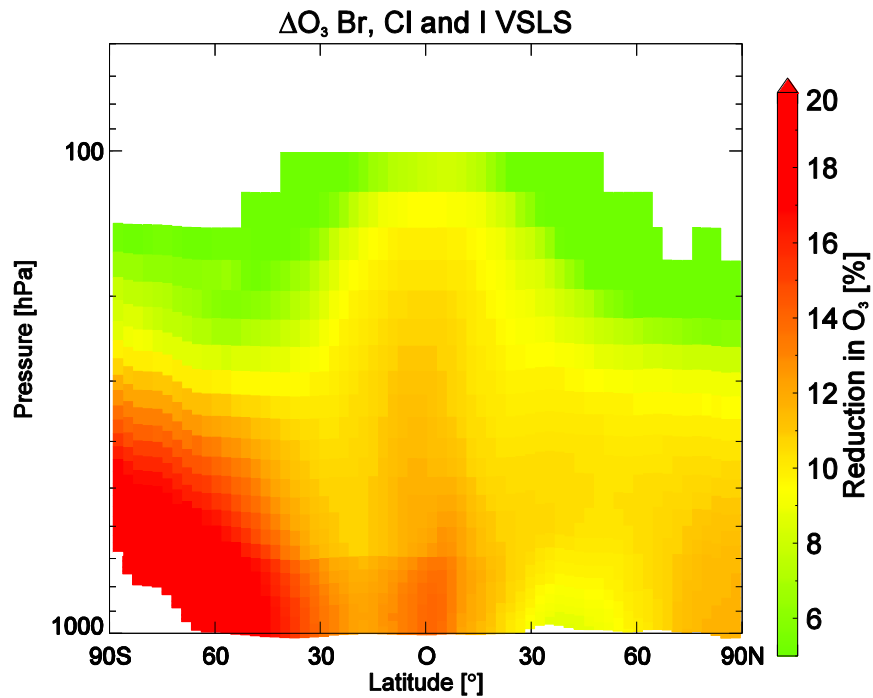
\*\* The tropospheric configuration of the TOMCAT model considers explicit emissions<sup>S4-5</sup> of the following VSLS: CHBr<sub>3</sub>, CH<sub>2</sub>Br<sub>2</sub>, CHBr<sub>2</sub>Cl, CH<sub>2</sub>BrCl, CHBrCl<sub>2</sub>, CH<sub>3</sub>I, CH<sub>2</sub>I<sub>2</sub>, CH<sub>2</sub>ICl, CH<sub>2</sub>IBr, C<sub>2</sub>H<sub>5</sub>I and C<sub>3</sub>H<sub>7</sub>I. A latitude-dependent mixing ratio boundary condition, derived from available global surface observations, was used to constrain the abundance of CHCl<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>2</sub>ClCH<sub>2</sub>Cl, C<sub>2</sub>HCl<sub>3</sub> and C<sub>2</sub>Cl<sub>4</sub> in the model. TOMCAT has been used extensively for previous studies of tropospheric halogen chemistry<sup>S1</sup> and studies examining the emission<sup>S2</sup>, transport and chemistry of VSLS<sup>S3</sup>. The above model configuration has been shown previously to perform well in reproducing atmospheric observations of a range of VSLS in the troposphere<sup>S2-3</sup>.

**Table S4. Observed surface mixing ratio (ppt) and growth rate of CH<sub>2</sub>Cl<sub>2</sub>.** Observations made as part of the ongoing National Oceanic and Atmospheric Administration Earth System Research Laboratory (NOAA/ESRL) monitoring program. Average growth rates at all sites calculated between 2010-2013.

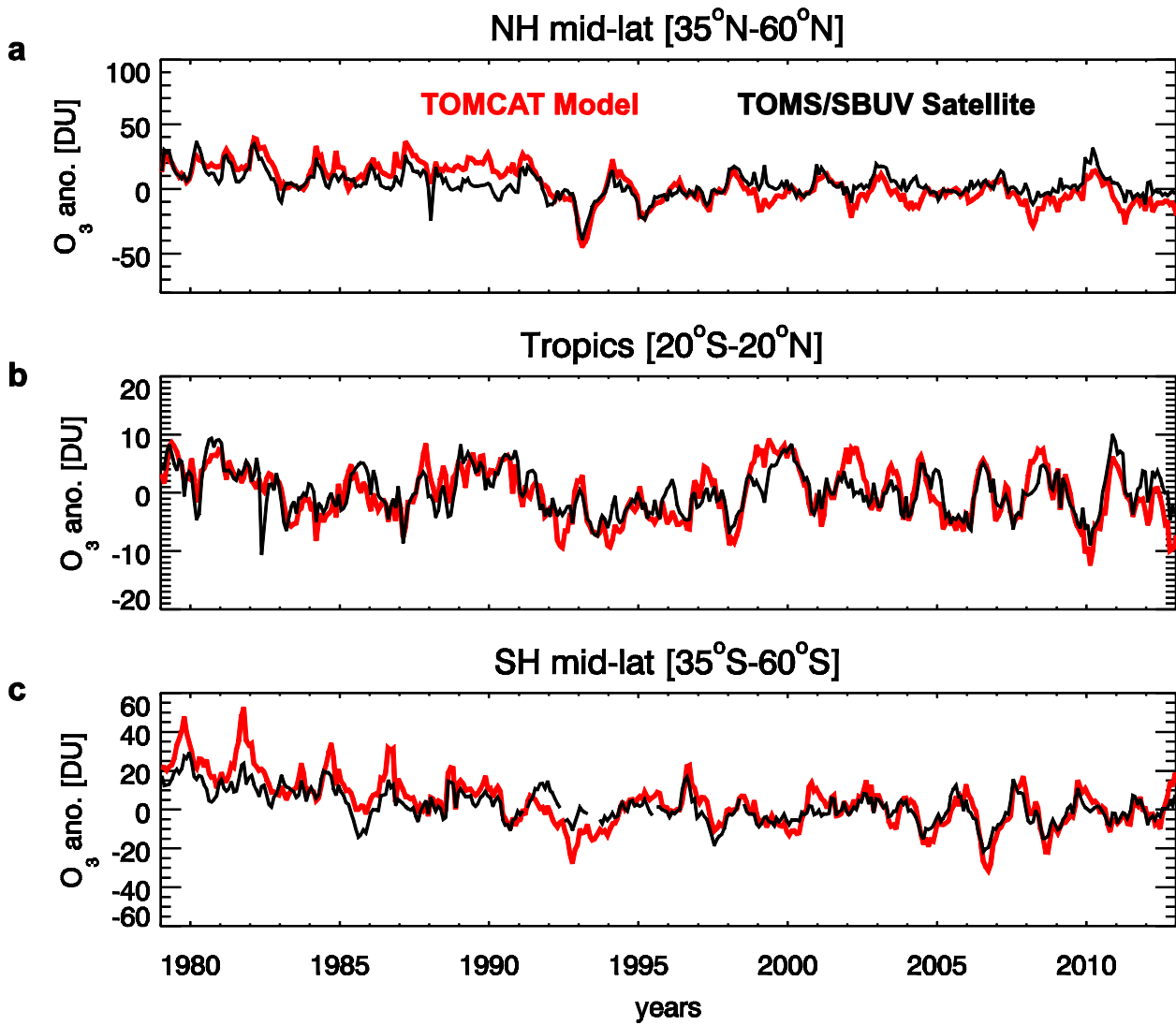
Monitoring Site	Annual Mean Mixing Ratio [ppt]				Growth Rate	
	2010	2011	2012	2013	ppt yr <sup>-1</sup>	% yr <sup>-1</sup>
Alert, NW Territories, Canada*	46.0	47.0	48.1	59.6	4.5	8.6
Summit, Greenland	46.8	45.7	47.5	60.2	4.5	8.4
Pt. Barrow, Alaska, USA*	46.5	46.3	48.2	59.8	4.4	8.4
Mace Head, Ireland	46.7	45.7	48.1	58.9	4.1	7.8
Wisconsin, USA	48.8	48.4	52.0	62.0	4.4	8.0
Trinidad Head, USA	48.6	48.6	49.7	61.5	4.3	7.9
Niwot Ridge, Colorado, USA*	44.0	45.7	50.7	60.3	5.4	10.5
Cape Kumukahi, Hawaii, USA*	42.9	42.3	45.0	53.9	3.7	7.6
Mauna Loa, Hawaii, USA*	39.2	37.4	41.9	52.0	4.3	9.4
Cape Matatula, American Samoa*	14.9	15.6	16.4	19.3	1.4	8.5
Cape Grim, Tasmania, Australia*	13.1	13.7	14.3	15.6	0.9	6.0
Palmer Station, Antarctica	12.7	13.4	13.4	15.3	0.9	6.3
South Pole*	12.1	13.3	13.3	14.2	0.7	5.4
Northern Hemisphere	43.7	43.7	46.8	57.1	4.5	8.9
Southern Hemisphere	13.4	14.2	14.7	16.4	1.0	6.8
<b>All site average</b>	<b>35.5</b>	<b>35.6</b>	<b>37.6</b>	<b>45.6</b>	<b>3.3</b>	<b>8.3</b>

Notes:

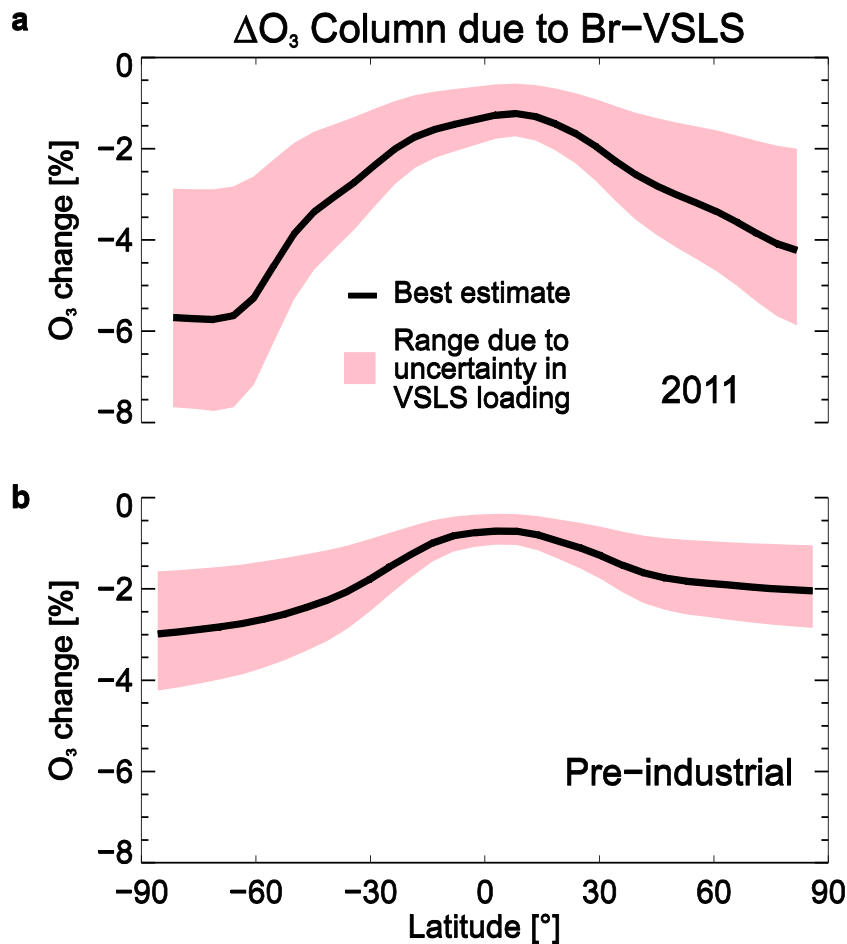
\* Sites used in estimating hemispheric mean mixing ratios in Table S3 and in Figure 3.



**Figure S1. Impact of VLSL on tropospheric ozone.** Simulated  $O_3$  reduction (%) due to combined influence of bromine, chlorine and iodine from VLSL in 2011 (relative to a control run with no VLSL). The impact of chlorine VLSL on tropospheric ozone is here negligible (<0.5%).



**Figure S2. Long-term ozone anomalies from observations and model.** Comparison of total ozone anomalies (Dobson Units) for (a) northern hemisphere mid-latitudes (35°N-60°N), (b) tropics (20°S-20°N) and (c) southern hemisphere mid-latitudes (35°S-60°S). Anomalies are calculated by subtracting climatological monthly mean column ozone values (1990-2005) from monthly mean values.



**Figure S3. Present day and pre-industrial impact of bromine VSLs on column ozone.** Simulated column ozone change (%) due to a best estimate of 6 parts per trillion (ppt) of bromine VSLs in the stratosphere relative to a run with no VSLs in (a) 2011 and (b) the pre-industrial stratosphere (background  $CH_3Br$  and  $CH_3Cl$  only). Globally averaged, the influence of bromine VSLs on column ozone is  $\sim 30\%$  smaller in the pre-industrial stratosphere. The shaded regions denote the range due to uncertainty in the stratospheric loading of bromine VSLs; simulations were also performed with 3 ppt and 8 ppt.

## Supplementary References

S1 Breider, T. J. et al. Impact of BrO on dimethylsulfide in the remote marine boundary layer. *Geophys. Res. Lett.*, **37**, L02807 (2010).

S2 Hossaini, R. et al. Evaluating global emission inventories of biogenic bromocarbons. *Atmos. Chem. Phys.* **13**, 11819-11838 (2013).

S3 Hossaini, R. et al. The contribution of natural and anthropogenic very short-lived species to stratospheric bromine. *Atmos. Chem. Phys.* **12**, 371-380 (2012).

S4 Ordóñez, C. et al. Bromine and iodine chemistry in a global chemistry-climate model: description and evaluation of very short-lived oceanic sources. *Atmos. Chem. Phys.* **12**, 1423-1447 (2012).

S5 Ziska, F. et al. Global sea-to-air flux climatology for bromoform, dibromomethane and methyl iodide. *Atmos. Chem. Phys.* **13**, 8915-8934 (2013).