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(54) Title: PHENOLIC COMPOUNDS WITH ANTIOXIDANT AND ANTI-CANCER PROPERTIES, ANALOGS AND SYNTHESIS THEREOF

(57) Abstract: The present document describes a phytochemical isolated from maple syrup and composition comprising the same. More specifically, the document describes an antioxidant phytochemical compound, derivatives thereof, and composition comprising the same. The document also describes a process of synthesizing the antioxidant phytochemical compound.



WO 2012/167364 A1

File No. P1777PC00

Title: PHENOLIC COMPOUNDS WITH ANTIOXIDANT AND ANTI-CANCER PROPERTIES, ANALOGS AND SYNTHESIS THEREOF

CROSS-REFERENCE TO RELATED APPLICATIONS

This application claims priority of US provisional patent application 61/495,574 filed on June 10, 2011.

BACKGROUND

(a) Field

[0001] The subject matter disclosed generally relates to a phytochemical isolated from maple syrup and composition comprising the same. More specifically, the subject matter relates to an antioxidant phytochemical compound, derivatives thereof, and composition comprising the same. The subject matter also relates to a process of synthesizing the antioxidant phytochemical compound.

(b) Related Prior Art

[0002] Maple syrup (MS) is a natural product obtained by thermal evaporation of sap collected from certain maple (*Acer*) species including the sugar maple (*A. saccharum*) tree. The province of Quebec in Canada is the largest producer of MS, and this premium natural sweetener is popularly consumed worldwide. Thus, identification of the chemical constituents, beyond the natural sugars (sucrose), of MS is of great interest from a human health perspective. MS contains a diverse range of phenolic compounds which are naturally present in the xylem sap and concentrated in syrup. The identification of these new phenolic compounds may lead to the foundation of a new class of compounds with health beneficial effects.

[0003] Therefore, there is a need for the identification of new constituent compounds of maple syrup that could have beneficial effects on health.

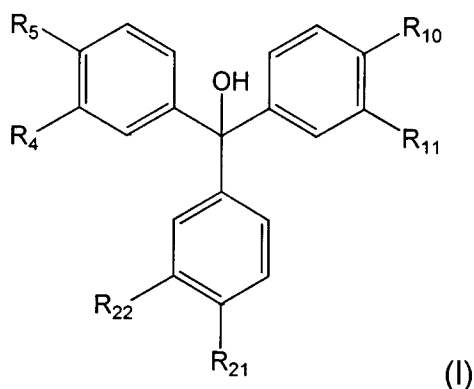
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[0004] There is a need for analogs of compounds from maple syrup that could have beneficial effects on health.

[0005] There is also a need for compositions containing new constituent compounds of maple syrup, and their analogs, that could have beneficial effects on health.

SUMMARY

[0006] According to an embodiment, there is provided a compound of formula (I):



[0007] wherein

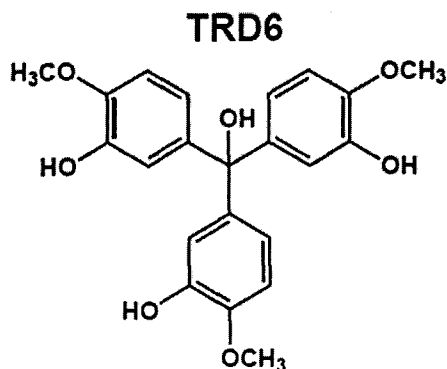
[0008] R_5 , R_{10} , and R_{21} may be OCH_3 ,

[0009] R_4 , R_{11} , and R_{22} may be independently chosen from OH, Cl, Br, and CHO,

[0010] pharmaceutically acceptable salt, racemic mixture, enantiomer, diastereoisomer, isomer, and tautomer thereof.

[0011] According to another embodiment, there is provided a compound of formula TRD6:

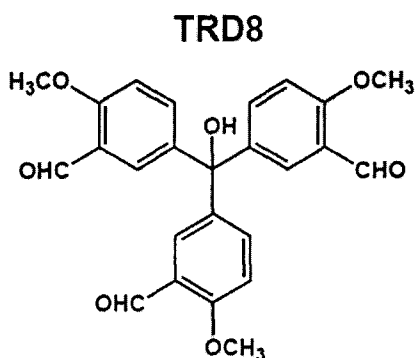
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[0012] 5,5',5''-(hydroxymethanetriyl)tris(2-methoxyphenol) ,

[0013] pharmaceutically acceptable salt, racemic mixture, enantiomer, diastereoisomer, isomer, and tautomer thereof.

[0014] According to another embodiment, there is provided a compound of formula TRD8:

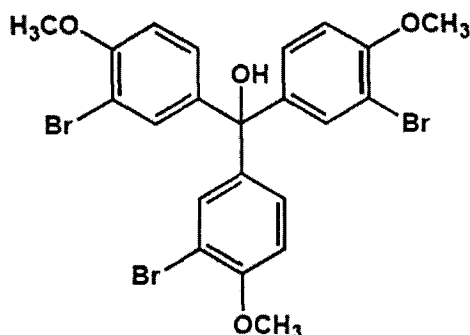


[0015] 5,5',5''-(hydroxymethanetriyl)tris(2-methoxybenzaldehyde) ,

[0016] pharmaceutically acceptable salt, racemic mixture, enantiomer, diastereoisomer, isomer, and tautomer thereof.

[0017] According to another embodiment, there is provided a compound of formula TRD9:

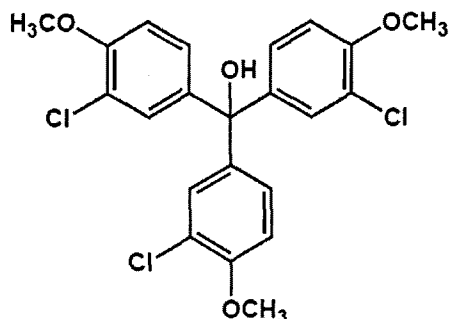
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TRD9

[0018] tris(3-bromo-4-methoxyphenyl)methanol ,

[0019] pharmaceutically acceptable salt, racemic mixture, enantiomer, diastereoisomer, isomer, and tautomer thereof.

[0020] According to another embodiment, there is provided a compound of formula TRD10:

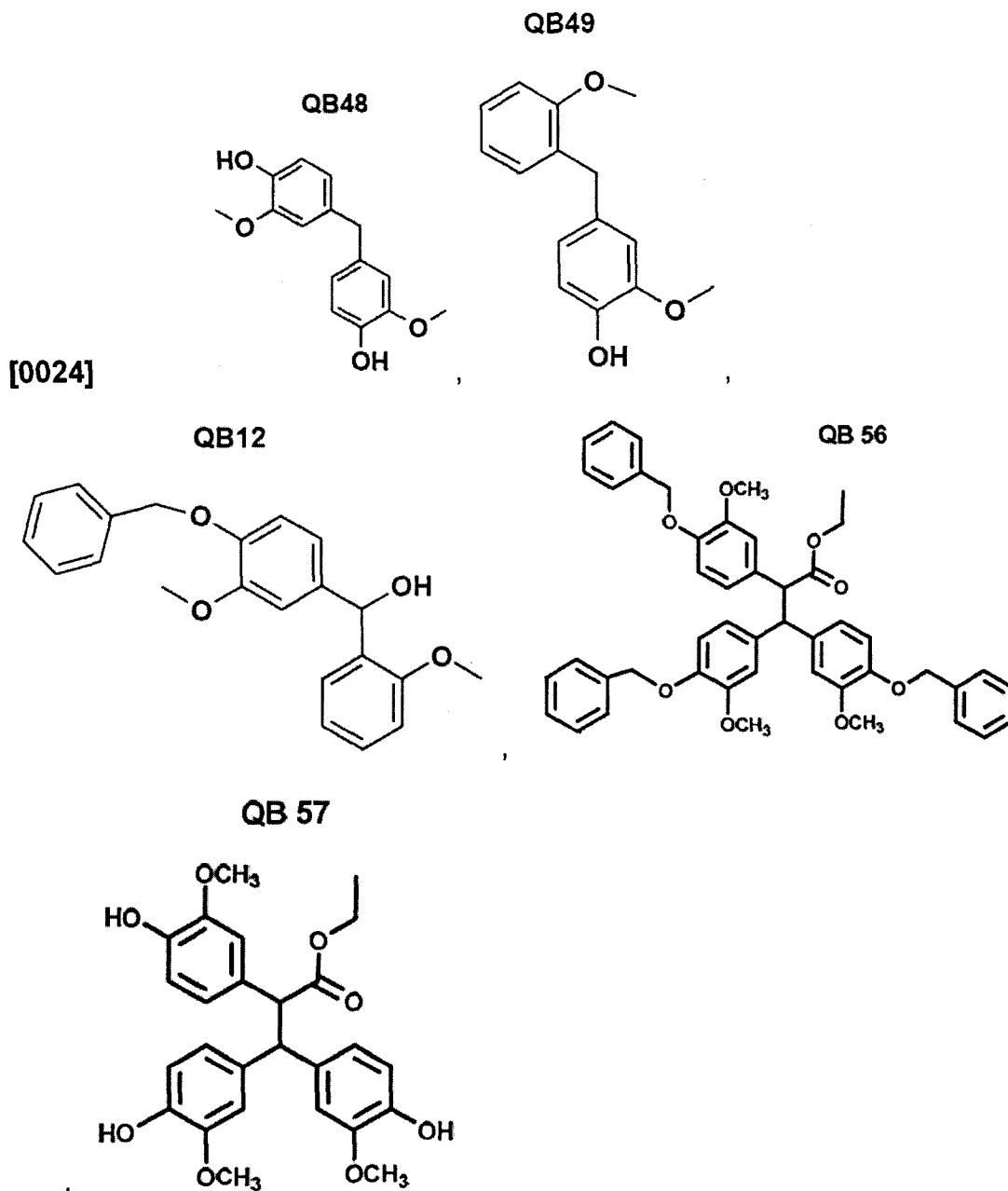
TRD10

[0021] tris(3-chloro-4-methoxyphenyl)methanol ,

[0022] pharmaceutically acceptable salt, racemic mixture, enantiomer, diastereoisomer, isomer, and tautomer thereof.

[0023] According to another embodiment, there is provided a compound of formula QB12, QB48, QB49, QB56, and QB57:

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,and

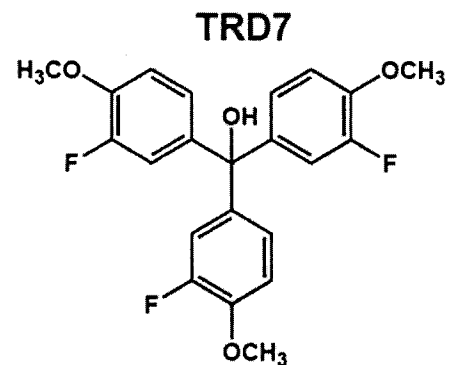
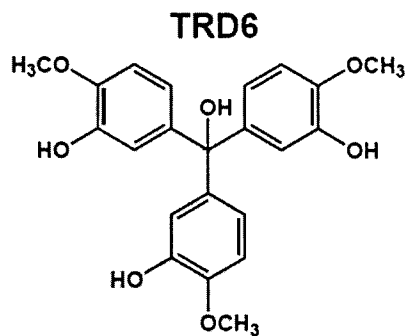
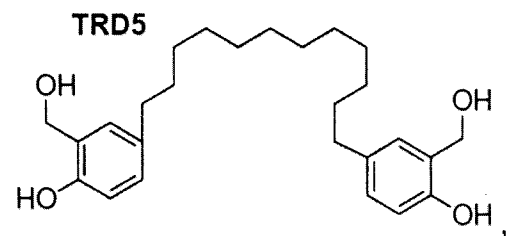
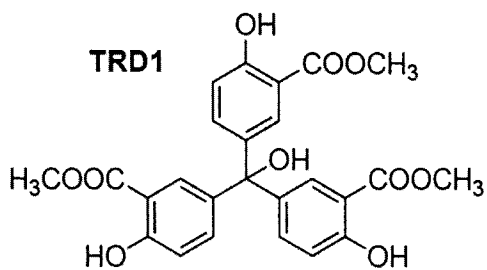
[0025] pharmaceutically acceptable salt, racemic mixture, enantiomer, diastereoisomer, isomer, and tautomer thereof.

[0026] According to another embodiment, there is provided a pharmaceutical composition comprising a therapeutically effective amount of a compound according to the present invention.

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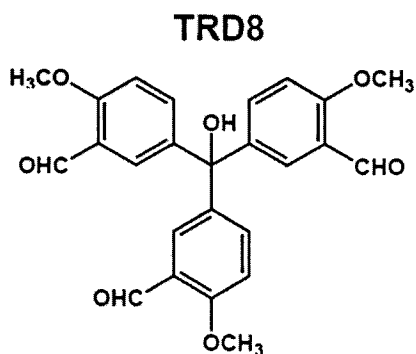
[0027] According to another embodiment, there is provided a method to inhibit tumor growth in a subject, which comprises administering a composition according to the present invention.

[0028] According to another embodiment, there is provided a method to inhibit tumor growth in a subject, which comprises administering an anticancer amount of a compound TRD1, RD5, TRD6, TRD7, TRD8, TRD9, TRD10, QB12, QB39, QB46, QB56, and QB57:

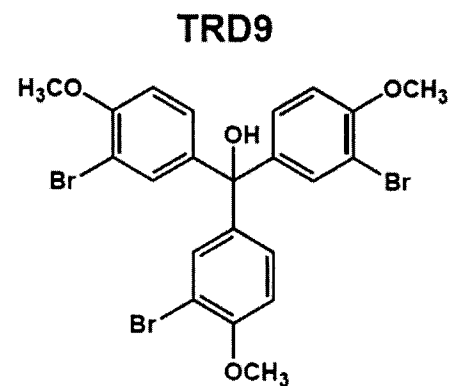


5,5',5''-(hydroxymethanetriyl)tris(2-methoxyphenol) ,

tris(3-fluoro-4-methoxyphenyl)methanol ,



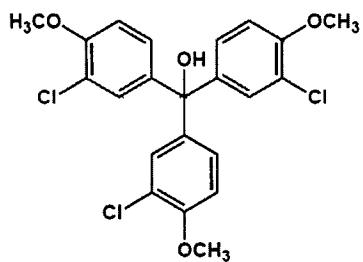
5,5',5''-(hydroxymethanetriyl)tris(2-methoxybenzaldehyde) ,



tris(3-bromo-4-methoxyphenyl)methanol ,

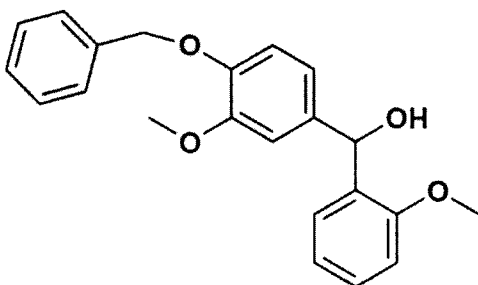
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TRD10 100 mg

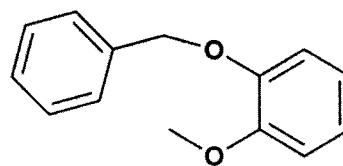


tris(3-chloro-4-methoxyphenyl)methanol ,

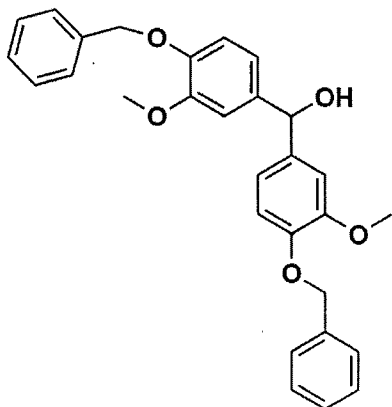
QB12



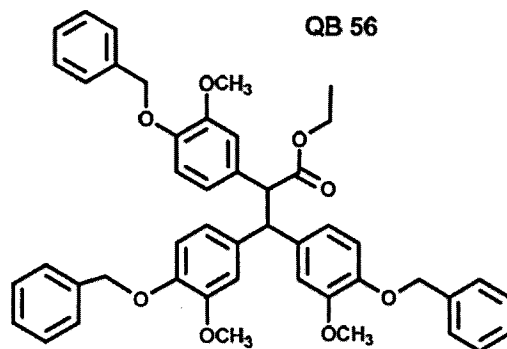
QB39



QB46

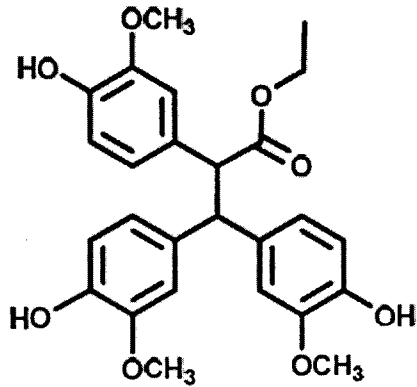


QB 56



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QB 57

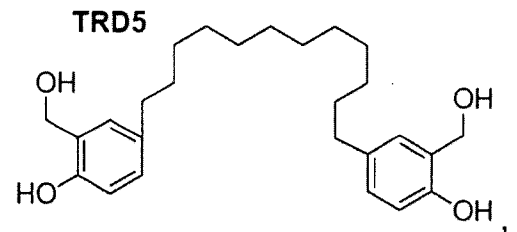
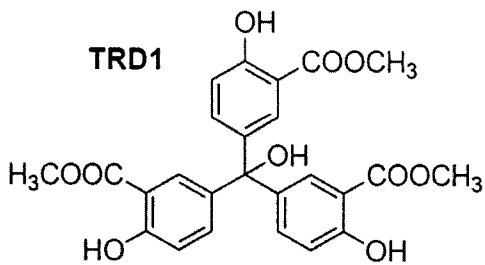


and

[0029] According to another embodiment, there is provided a use of a compound of the present invention for the preparation of a medicament for the treatment of cancer.

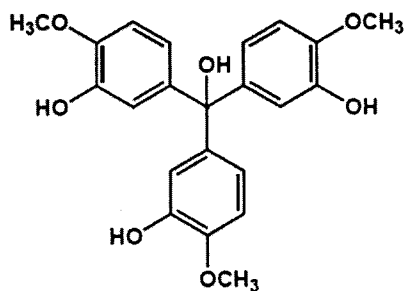
[0030] According to another embodiment, there is provided a use of a compound of the present invention for the treatment of cancer.

[0031] According to another embodiment, there is provided a of a compound TRD1, TRD5, TRD6, TRD7, TRD8, TRD9, TRD10, QB12, QB39, QB46, QB56, and QB57 to inhibit tumor growth in a subject:



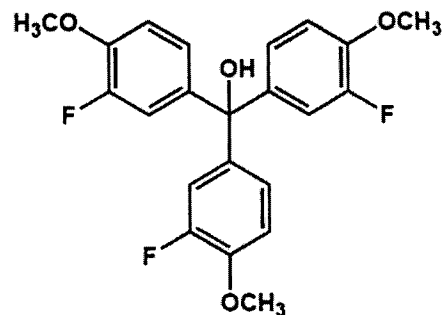
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TRD6



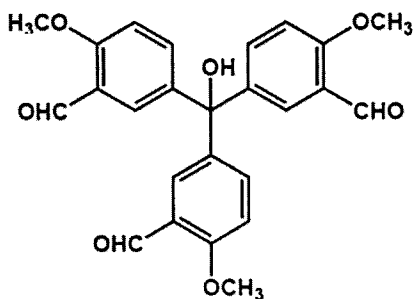
5,5',5''-(hydroxymethanetriyl)tris(2-methoxyphenol) ,

TRD7



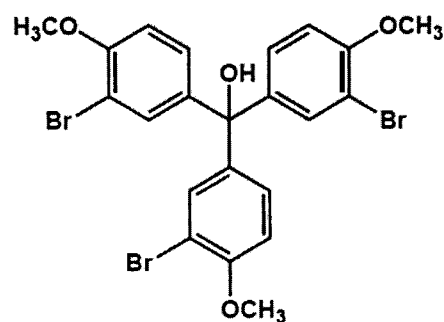
tris(3-fluoro-4-methoxyphenyl)methanol ,

TRD8



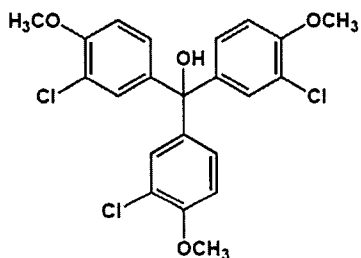
5,5',5''-(hydroxymethanetriyl)tris(2-methoxybenzaldehyde) ,

TRD9



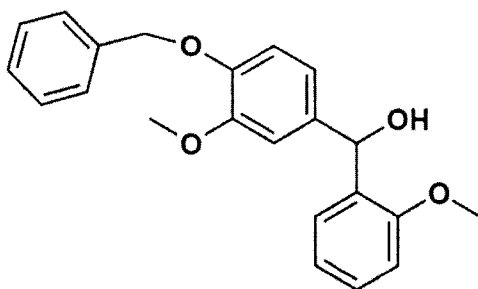
tris(3-bromo-4-methoxyphenyl)methanol ,

TRD10

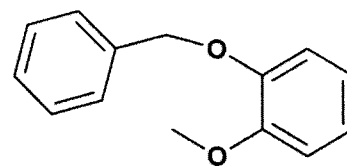


tris(3-chloro-4-methoxyphenyl)methanol ,

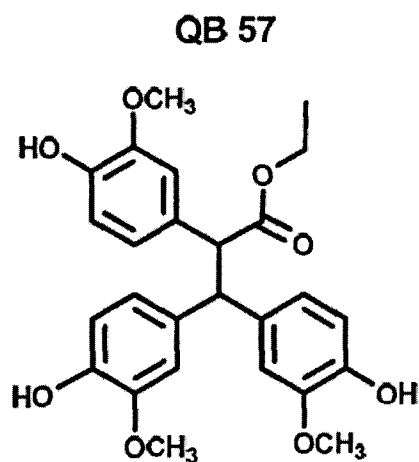
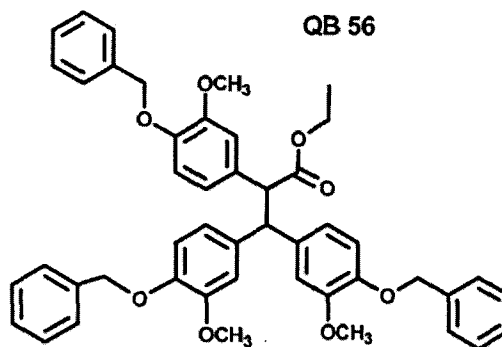
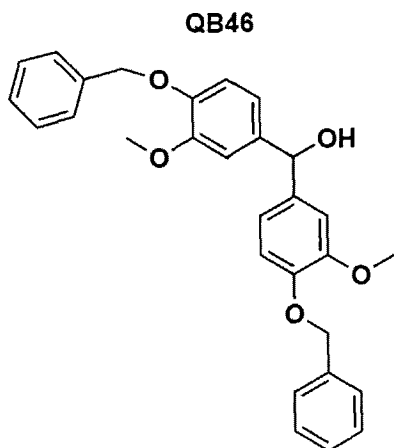
QB12



QB39

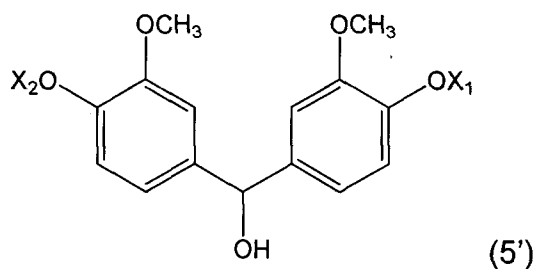


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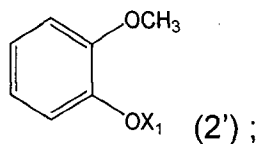
and

[0032] According to another embodiment, there is disclosed a process for the synthesis of a compound of formula (5'):



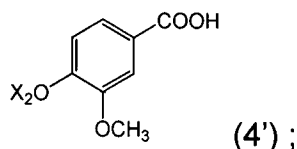
[0033] comprising the step of :

- i. reacting a compound of formula (2')



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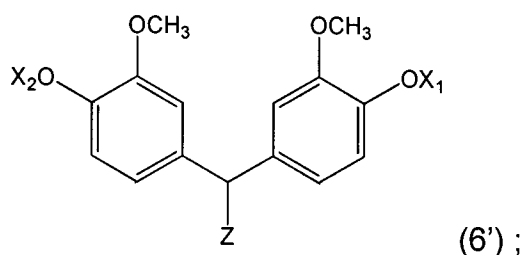
[0034] with a compound of formula (4')



[0035] to obtain the compound of formula (5')

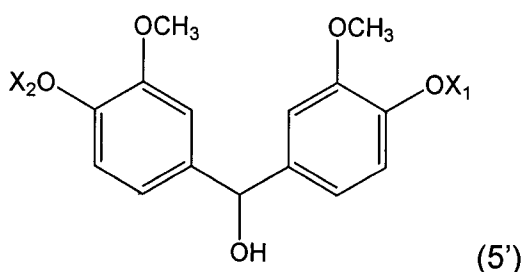
[0036] wherein X_1 , and X_2 may be a suitable protecting group for a hydroxyl group.

[0037] According to another embodiment, there is disclosed a process for the synthesis of a compound of formula (6')



[0038] comprising the step of :

i. reacting a compound of formula (5')

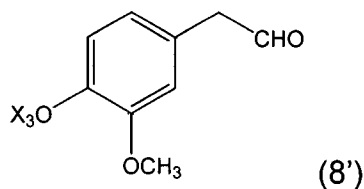


[0039] with a halogenating agent to obtain the compound of formula (6'),

[0040] wherein X_1 , and X_2 may be a suitable protecting group for a hydroxyl group, and wherein Z may be a halogen atom.

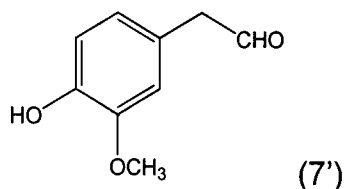
[0041] According to another embodiment, there is disclosed a process for the synthesis of a compound of formula (8')

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[0042] the process comprising the step of :

i. reacting a compound of formula (7')

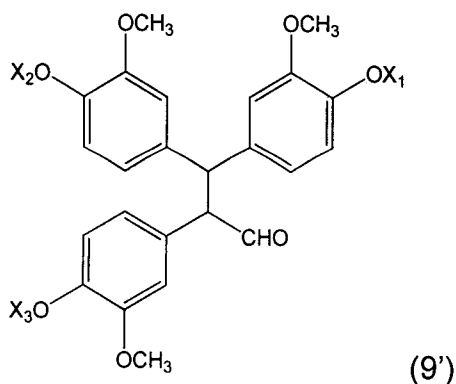


[0043] with a suitable hydroxyl protecting group, to obtain the compound of formula (8')

[0044] wherein X_3 may be a suitable protecting group for a hydroxyl group; and

[0045] wherein Z may be a halogen atom.

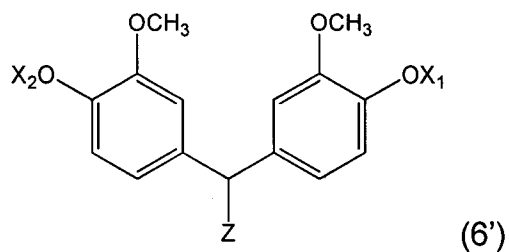
[0046] According to another embodiment, there is disclosed a process for the synthesis of a compound of formula (9')



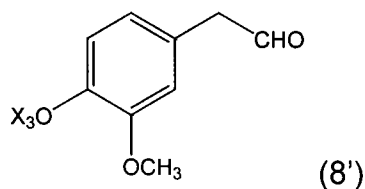
[0047] the process comprising the step of :

i. reacting a compound of formula (6')

File No. P1777PC00



[0048] with a compound of formula (8')

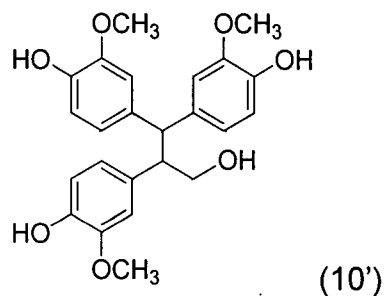


[0049] to obtain the compound of formula (9'),

[0050] wherein X_1 , X_2 and X_3 may be a suitable protecting group for a hydroxyl group; and

[0051] wherein Z may be a halogen atom.

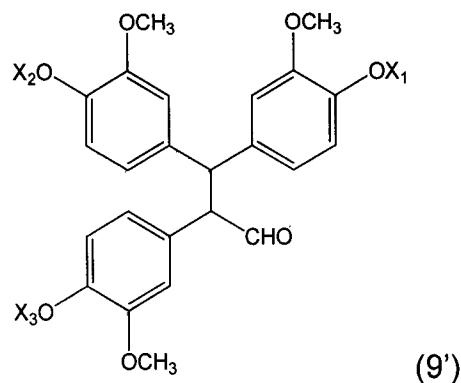
[0052] According to another embodiment, there is disclosed a process for the synthesis of a compound of formula (10') (Quebecol)



[0053] the process comprising the steps of:

- i. reducing and deprotecting a compound of formula (9')

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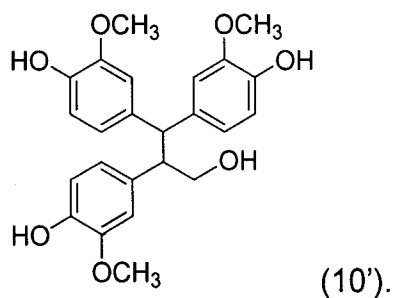


[0054] to obtain the compound of formula (10') (Quebecol);

[0055] wherein X_1 , X_2 and X_3 may be a suitable protecting group for a hydroxyl group; and

[0056] wherein Z may be a halogen atom

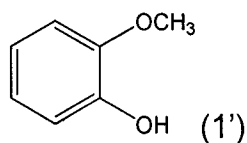
[0057] According to another embodiment, there is provided a process for the synthesis of a compound of formula (10') (Quebecol)



[0058]

[0059] The process may be comprising the steps of:

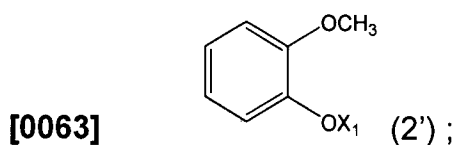
[0060] i. reacting a compound of formula (1')



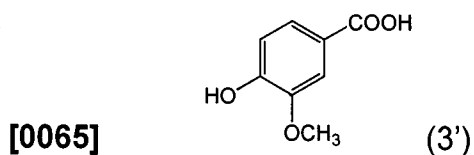
[0061]

[0062] with a suitable hydroxyl protecting group, to obtain a compound of formula (2')

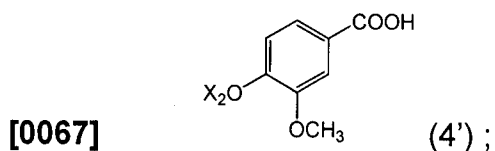
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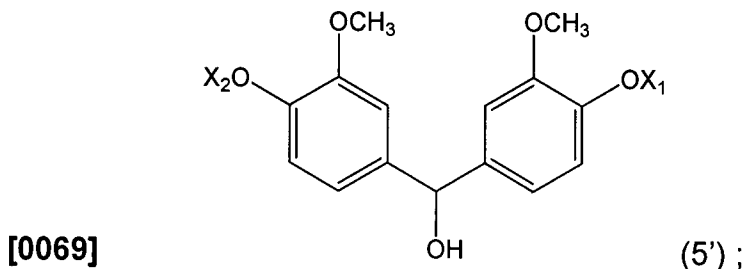
[0064] ii. reacting a compound of formula (3')



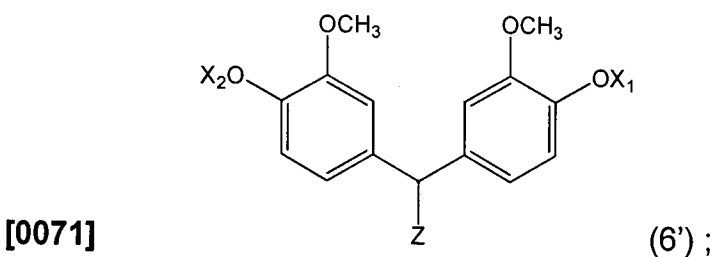
[0066] with a suitable hydroxyl protecting group, to obtain a compound of formula (4')



[0068] iii. reacting the compound of formula (2') with the compound of formula (4') to obtain a compound of formula (5')

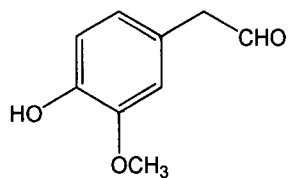


[0070] iv. reacting the compound of formula (5') with a halogenating agent to obtain a compound of formula (6')

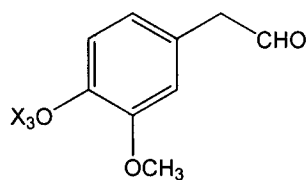


[0072] v. reacting a compound of formula (7')

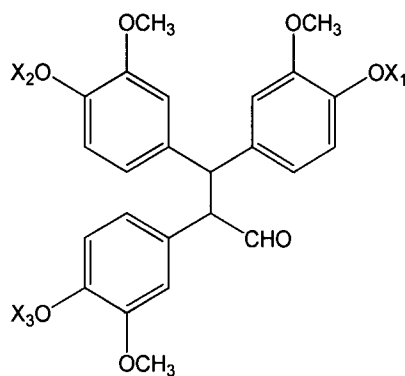
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**[0073]** (7')

[0074] with a suitable hydroxyl protecting group, to obtain a compound of formula (8')

**[0075]** (8') ;

[0076] vi. reacting the compound of formula (6') with the compound of formula (8') to obtain a compound of formula (9')

**[0077]** (9')

[0078] vii. reducing and deprotecting the compound of formula (9') to obtain a compound of formula (10') (Quebecol);

[0079] wherein X_1 , X_2 and X_3 may be a suitable protecting group for a hydroxyl group; and wherein Z may be a halogen atom.

[0080] The X_3 may be chosen from Fluorenylmethyloxycarbonyl chloride (FMOC), Triphenylmethyl chloride, and a silyl ether.

[0081] The X_3 may be a silyl ether.

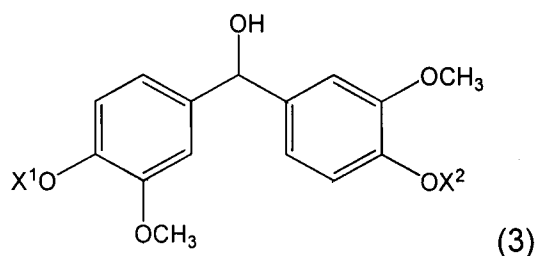
[0082] In the process according to the present invention, in step vii, the reducing may be by reacting the compound (9') with NaBH_4 .

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[0083] In the process according to the present invention, the deprotection may be by reacting the compound of formula (9') with one of tetra-*n*-butylammonium fluoride (TBAF) or trifluoroacetic acid (TFA).

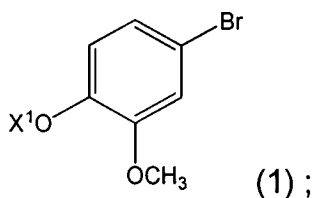
[0084] In the process according to the present invention, the halogenating agent may be a trihalide of phosphorous and the trihalide of phosphorous may be chosen from PBr₃, and PCl₃.

[0085] According to another embodiment, there is provided a process for the synthesis of a compound of formula (3)

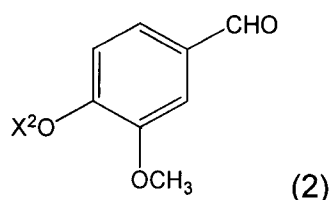


[0086] comprising the step of:

- i. reacting a compound of formula (1)



[0087] with a compound of formula (2)



[0088] in presence of a strong base, to obtain a compound of formula (3)

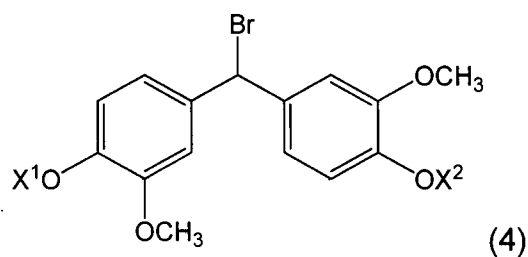
[0089] wherein X¹ and X² may be a suitable protecting group for a hydroxyl group.

[0090] The strong base may be *n*-butyllithium (*n*-BuLi).

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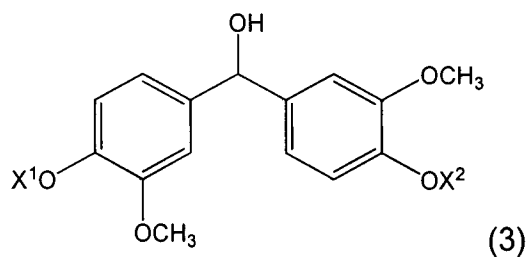
[0091] The reaction may be in tetrahydrofuran (THF) at -78°C.

[0092] According to another embodiment, there is provided a process for the synthesis of a compound of formula (4)



[0093] comprising the step of:

- i. brominating a compound of formula (3)



[0094] to obtain the compound of formula (4)

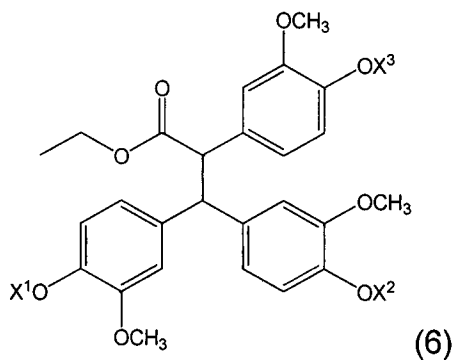
[0095] wherein X¹ and X² may be a suitable protecting group for a hydroxyl group.

[0096] The bromination may be with acetyl bromide (CH₃COBr).

[0097] The bromination may be with acetyl bromide (CH₃COBr) in benzene.

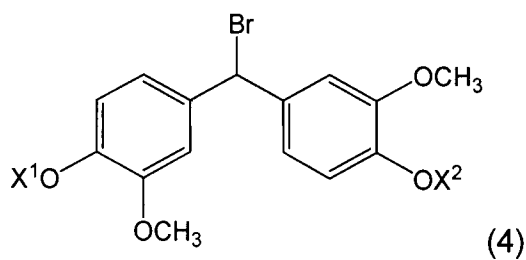
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[0098] According to another embodiment, there is provided a process for the synthesis of a compound of formula (6)

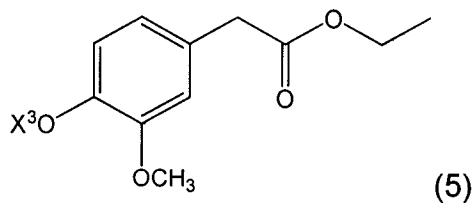


[0099] comprising the step of:

i. reacting a compound of formula (4)



[00100] with a compound of formula (5)



[00101] in the presence of a strong base, to obtain the compound of formula (6),

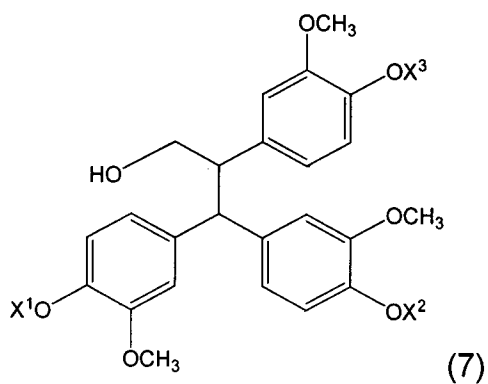
[00102] wherein X¹, X², and X³ may be a suitable protecting group for a hydroxyl group.

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[00103] The strong base may be Lithium diisopropylamide (LDA).

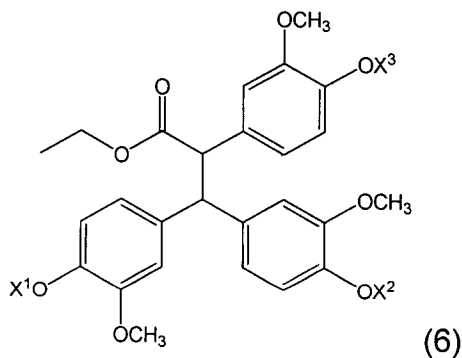
[00104] The strong base may be Lithium diisopropylamide (LDA) in tetrahydrofuran at tetrahydrofuran (THF) at -78°C .

[00105] According to another embodiment, there is provided a process for the synthesis of a compound of formula (7)



[00106] comprising the step of:

- i. reducing a compound of formula (6)



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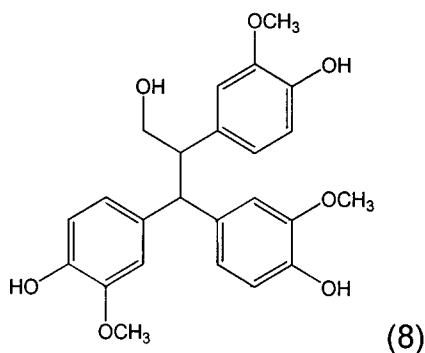
[00107] to obtain the compound of formula (7),

[00108] wherein X^1 , X^2 , and X^3 may be a suitable protecting group for a hydroxyl group.

[00109] The reduction may be with lithium aluminum hydride ($LiAlH_4$).

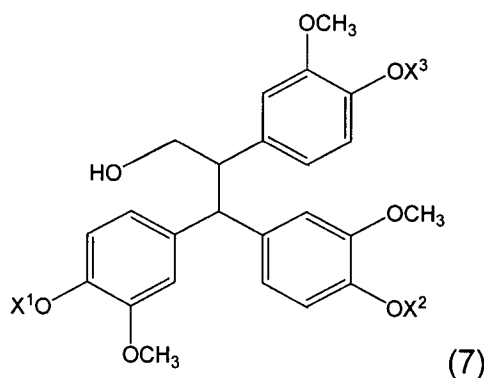
[00110] The reduction may be with lithium aluminum hydride ($LiAlH_4$) in tetrahydrofuran (THF).

[00111] According to another embodiment, there is provided a process for the synthesis of a compound of formula (8) (Quebecol)



[00112] comprising the step of:

- i. deprotecting a compound of formula (7)



[00113] to obtain the compound of formula (8),

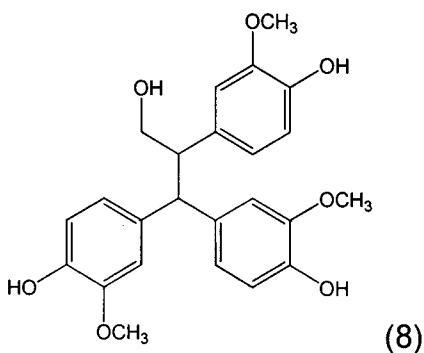
[00114] wherein X^1 , X^2 , and X^3 may be a suitable protecting group for a hydroxyl group.

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[00115] The deprotection may be with ammonium formate (HCO_2NH_4) and palladium on carbon (Pd/C).

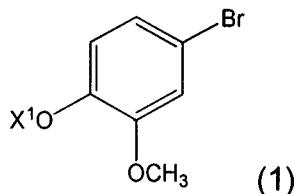
[00116] The deprotection may be with ammonium formate (HCO_2NH_4) and palladium on carbon (Pd/C) in methanol (MeOH).

[00117] According to another embodiment, there is provided a process for the synthesis of a compound of formula (8) (Quebecol)

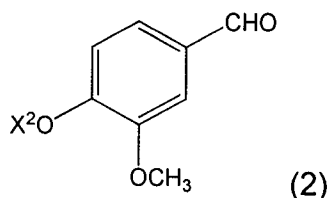


[00118] comprising the step of:

- i. reacting a compound of formula (1)

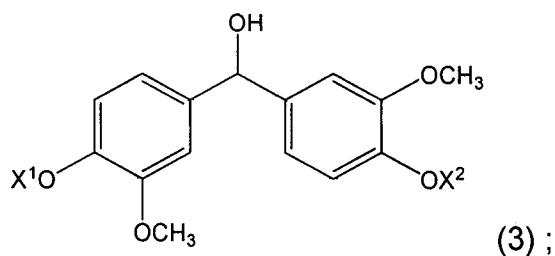


[00119] with a compound of formula (2)

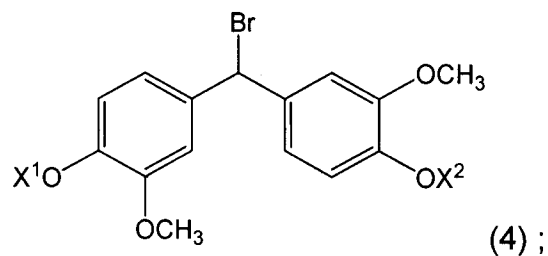


in presence of a strong base, to obtain a compound of formula (3)

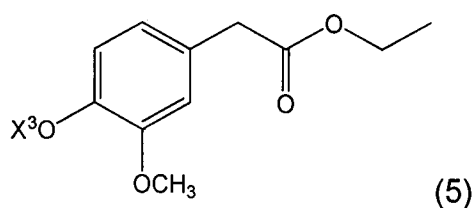
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- ii. brominating the compound of formula (3), to obtain a compound of formula (4)

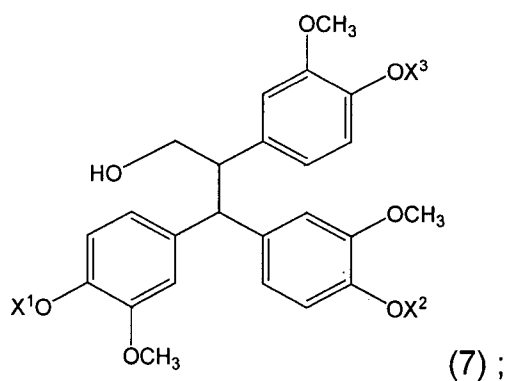


- iii. reacting the compound of formula (4) with a compound of formula (5)



in the presence of a strong base, to obtain a compound of formula (6) ;

- iv. reducing the compound of formula (6) to obtain a compound of formula (7)



- v. deprotecting the compound of formula (7) to obtain the compound of formula (8) (Quebecol),

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[00120] wherein X^1 , X^2 , and X^3 may be a suitable protecting group for a hydroxyl group.

[00121] The suitable protecting group for a hydroxyl group may be chosen from C_1 - C_{25} ethers, C_1 - C_{25} substituted methyl ethers, C_1 - C_{25} substituted ethyl ethers, C_1 - C_{25} acyl groups, C_1 - C_{25} halogenated acyl groups, C_1 - C_{25} substituted benzyl ethers, C_1 - C_{25} silyl ethers, C_1 - C_{25} esters, C_1 - C_{25} carbonates, and C_1 - C_{25} sulfonates.

[00122] The suitable protecting group for a hydroxyl group may be chosen from diphenylmethylchlorosilane (DPMS), Tosyl, methyl, methoxymethyl, benzyloxymethyl, tetrahydropyranyl, tetrahydrofuranyl, 2-(trimethylsilyl)ethoxymethyl, dioxanyl, 1-ethoxyethyl, 1-(2-chloroethoxy)ethyl, 2,2,2-trichloroethyl, t-butyl, allyl, propargyl, benzyl, p-methoxybenzyl, diphenylmethyl, triphenylmethyl, trimethylsilyl, triethylsilyl, triisopropylsilyl, dimethylisopropylsilyl, diethylisopropylsilyl, dimethylhexylsilyl, t-butyl dimethylsilyl, t-butyl diphenylsilyl, tribenzylsilyl, triphenylsilyl, triisopropylsilyl, diphenylmethylsilyl, benzylformate, methylcarbonyl, ethylcarbonyl, methoxymethyl carbonyl, trichloroethoxycarbonyl, benzylcarbonyl, benzyloxycarbonyl, allylsulfonyl, methanesulfonyl, and p-toluenesulfonyl.

[00123] The suitable protecting group for a hydroxyl group may be benzyl (Bn).

[00124] In the process of the present invention, in step i), the strong base may be n-butyllithium (n-BuLi).

[00125] The reaction may be in tetrahydrofuran (THF) at -78°C .

[00126] In the process of the present invention, in step ii) brominating may be with acetyl bromide (CH_3COBr).

[00127] In the process of the present invention, in step ii) brominating may be with acetyl bromide (CH_3COBr) in benzene.

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[00128] In the process of the present invention, in step iii) the strong base is Lithium diisopropylamide (LDA).

[00129] In the process of the present invention, in step iii) the strong base may be Lithium diisopropylamide (LDA) in tetrahydrofuran at tetrahydrofuran (THF) at -78°C.

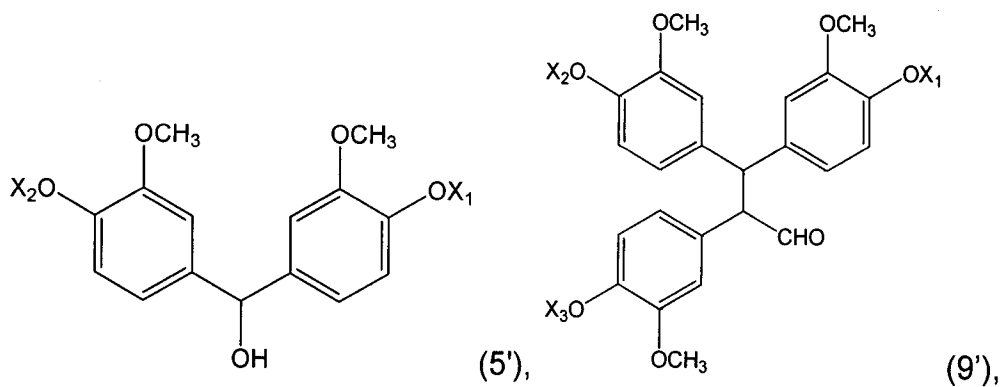
[00130] In the process of the present invention, in step iv) reducing may be with lithium aluminum hydride (LiAlH₄).

[00131] In the process of the present invention, in step iv) reducing may be with lithium aluminum hydride (LiAlH₄) in tetrahydrofuran (THF).

[00132] In the process of the present invention, in step v) deprotecting may be with ammonium formate (HCO₂NH₄) and palladium on carbon (Pd/C).

[00133] The deprotection may be with ammonium formate (HCO₂NH₄) and palladium on carbon (Pd/C) in methanol (MeOH).

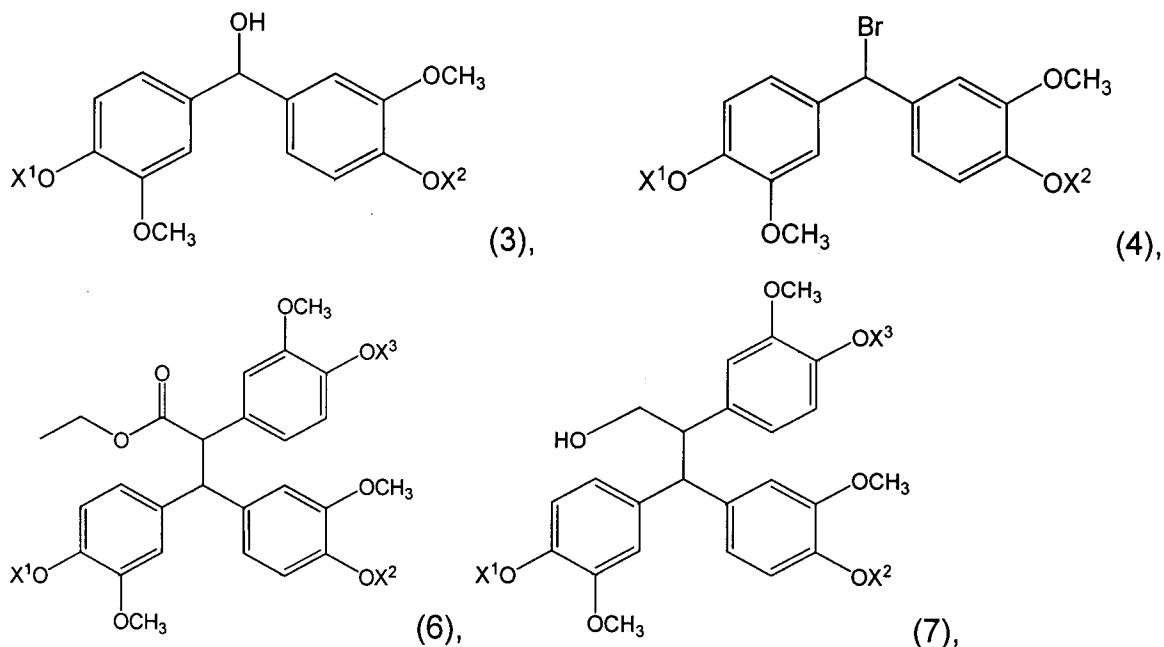
[00134] According to another embodiment, there is disclosed a compound of formula (5') and (9'):



[00135] wherein X₁, X₂ and X₃ are as defined above.

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[00136] According to another embodiment, there is disclosed a compound of formula (3), (4), (6) and (7):



[00137] wherein X^1 , X^2 and X^3 are as defined above.

[00138] Features and advantages of the subject matter hereof will become more apparent in light of the following detailed description of selected embodiments, as illustrated in the accompanying figures. As will be realized, the subject matter disclosed and claimed is capable of modifications in various respects, all without departing from the scope of the claims. Accordingly, the drawings and the description are to be regarded as illustrative in nature, and not as restrictive and the full scope of the subject matter is set forth in the claims.

BRIEF DESCRIPTION OF THE DRAWINGS

[00139] Further features and advantages of the present disclosure will become apparent from the following detailed description, taken in combination with the appended drawings, in which:

[00140] Fig. 1 illustrates the structural similarity between Tamoxifen and Quebecol.

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[00141] Fig. 2 illustrates a reaction scheme for the synthesis of Quebecol according to an embodiment of the present invention.

[00142] Fig. 3 illustrates a reaction scheme for the synthesis of Quebecol according to an embodiment of the present invention

[00143] Fig. 4A illustrates a ^1H NMR spectrum of the compound 3.

[00144] Fig. 4B illustrates a MS spectrum of the compound 3.

[00145] Fig. 5A illustrates a ^1H NMR spectrum of the compound 6.

[00146] Fig. 5B illustrates a ^1H NMR spectrum of the compound 6.

[00147] Fig. 5C illustrates a ^1H NMR spectrum of the compound 6.

[00148] Fig. 5D illustrates a MS spectrum of the compound 6.

[00149] Fig. 6A illustrates a ^1H NMR spectrum of the compound 7.

[00150] Fig. 7B illustrates a ^1H NMR spectrum of the compound 7.

[00151] Fig. 6C illustrates a ^1H NMR spectrum of the compound 7.

[00152] Fig. 6D illustrates a MS spectrum of the compound 7.

[00153] Fig. 7A illustrates a ^1H NMR spectrum of the compound 8 - Quebecol.

[00154] Fig. 7B illustrates a ^1H NMR spectrum of the compound 8 - Quebecol.

[00155] Fig. 7C illustrates a HPLC Chromatogram of natural Quebecol (bottom trace) vs. Synthetic Quebecol (top trace).

[00156] Fig. 8 illustrates the chemical structure of phenolic compound named Quebecol.

[00157] Fig. 9A illustrates the chemical structure of phenolic compounds that are derivatives of Quebecol.

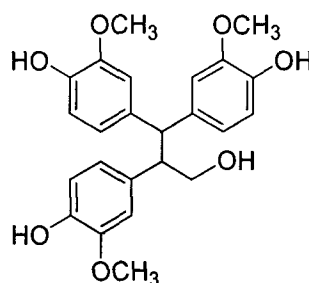
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[00158] Fig. 9B illustrates the chemical structure of phenolic compounds that are derivatives of Quebecol.

[00159] Fig. 9C illustrates the chemical structure of phenolic compounds that are derivatives of Quebecol.

DETAILED DESCRIPTION OF THE PREFERRED EMBODIMENTS

[00160] In embodiments there is disclosed a new polyphenolic compound isolated from Canadian maple syrup. The compound, which is obtained as a pale yellow amorphous powder has been named Quebecol.

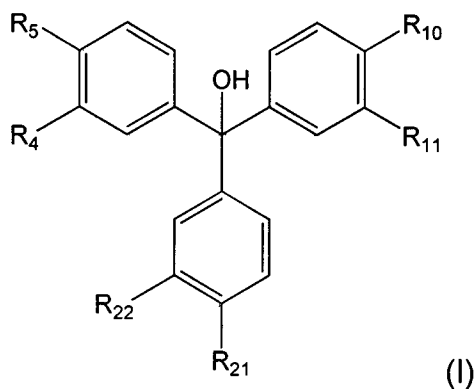


(10) - Quebecol

[00161] Now referring to Fig. 1, Quebecol displays some similarity to the known drug Tamoxifen. Tamoxifen is a widely used chemotherapy agent for hormonally dependent cancers such as breast cancer. However, Tamoxifen has severe side effects. Quebecol is a phytochemical derived compound present in maple syrup which has been consumed for centuries without toxicity. Thus, based on structural similarities to Tamoxifen and current laboratory assays, it is believed that Quebecol and analogs may exert greater anticancer effects than Tamoxifen without the adverse side effects.

[00162] According to another embodiment, the compounds of formula (I) are also represented by the compounds of formula (I):

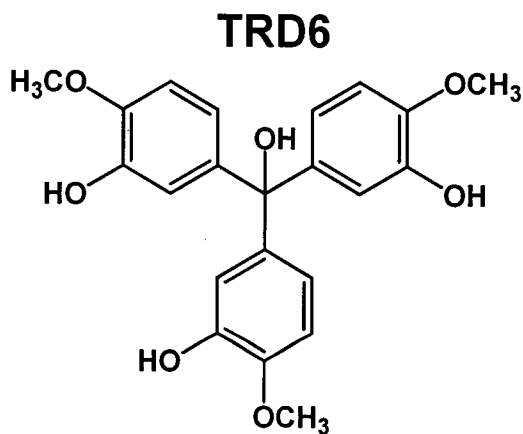
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[00163] where R₅, R₁₀, and R₂₁ are OCH₃,

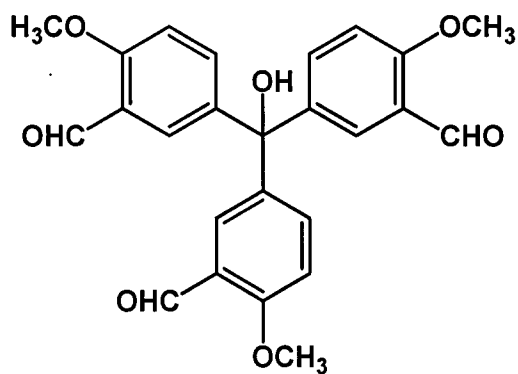
[00164] where R₄, R₁₁, and R₂₂ are independently chosen from OH, Cl, F, CF₃, CH₃, Br, and CHO, and their pharmaceutically acceptable salts, racemic mixture, enantiomer, diastereoisomer, isomer, and tautomer thereof.

[00165] In embodiments, there is also disclosed compounds of formulae TRD6, TRD8, TRD9, and TRD10:

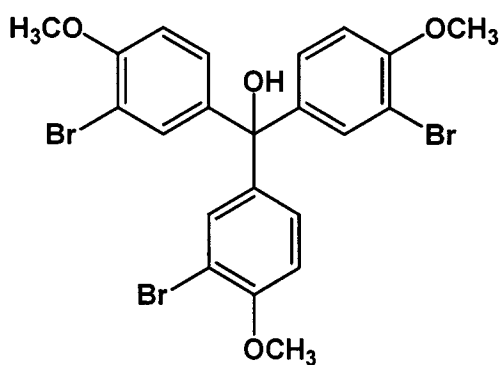


5,5',5''-(hydroxymethanetriyl)tris(2-methoxyphenol),

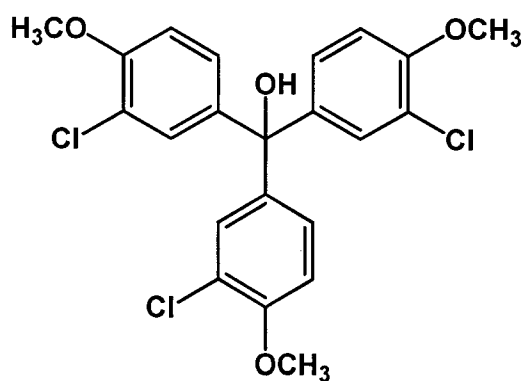
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TRD8

5,5',5''-(hydroxymethanetriyl)tris(2-methoxybenzaldehyde),

TRD9

tris(3-bromo-4-methoxyphenyl)methanol, and

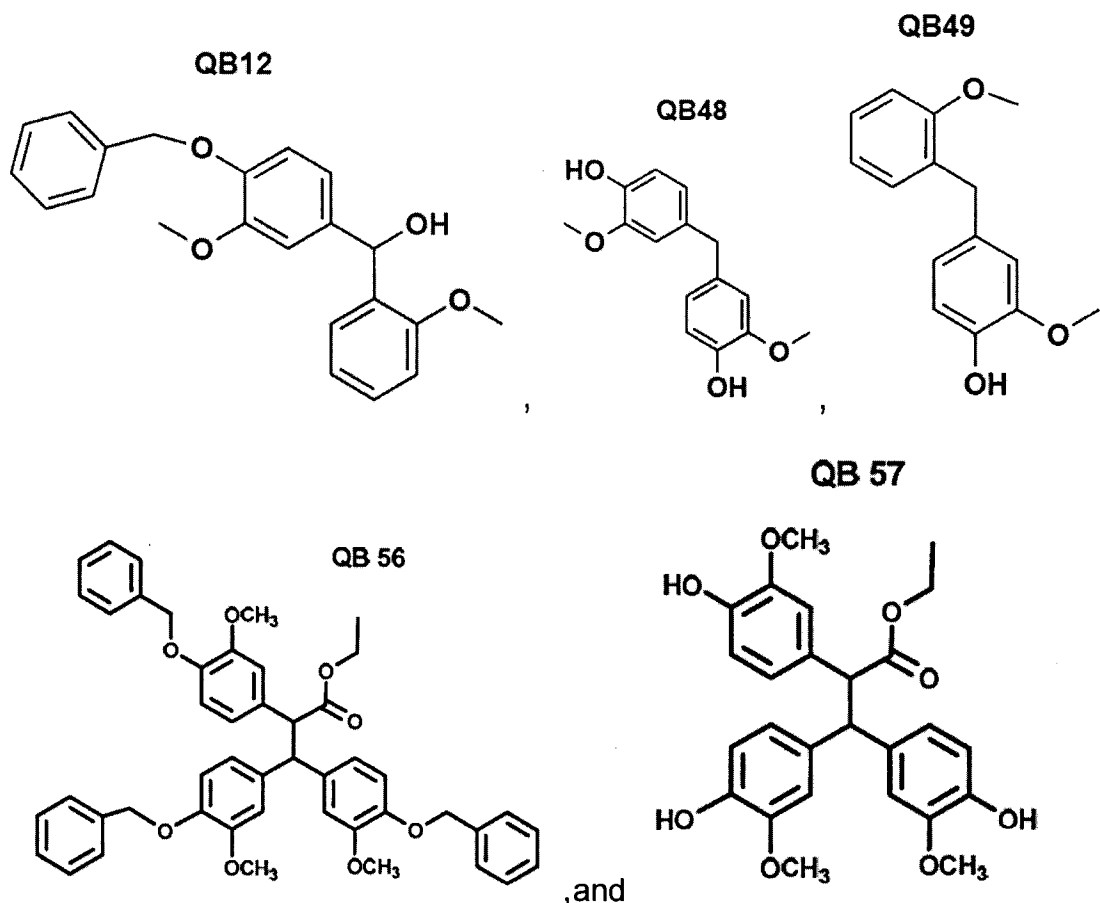
TRD10

tris(3-chloro-4-methoxyphenyl)methanol

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[00166] their pharmaceutically acceptable salt, racemic mixture, enantiomer, diastereoisomer, isomer, and tautomer thereof.

[00167] According to another embodiment, there is disclosed compounds of formulae QB12, QB48, QB49, QB56, and QB57:



[00168] their pharmaceutically acceptable salt, racemic mixture, enantiomer, diastereoisomer, isomer, and tautomer thereof.

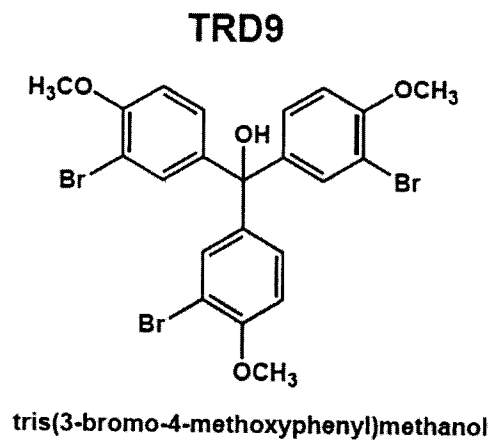
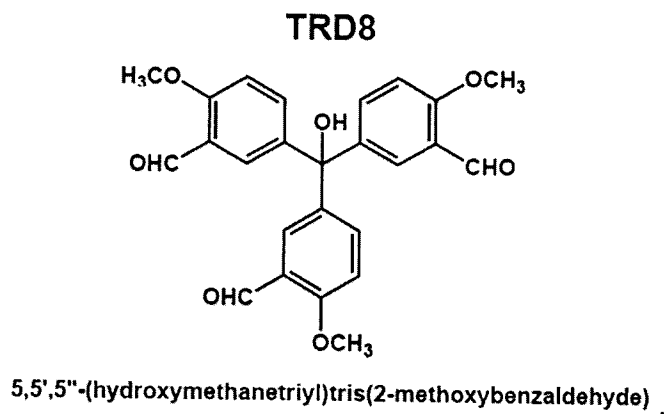
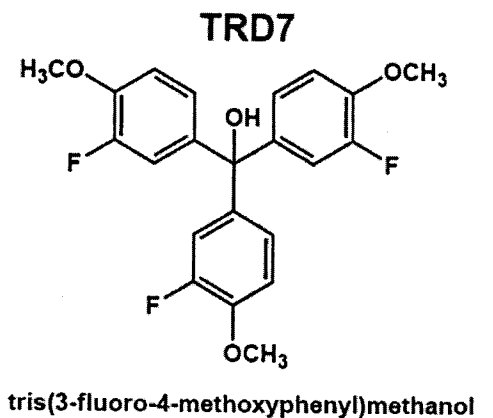
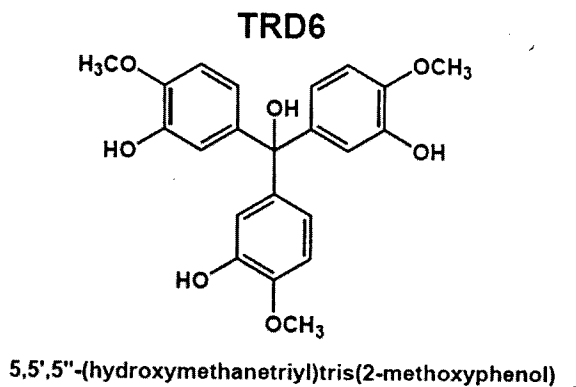
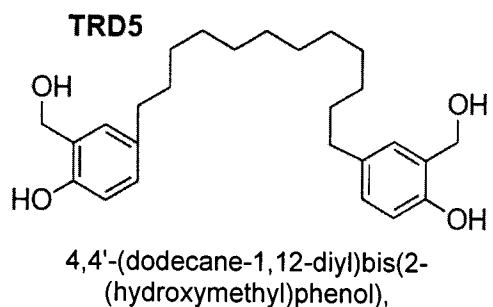
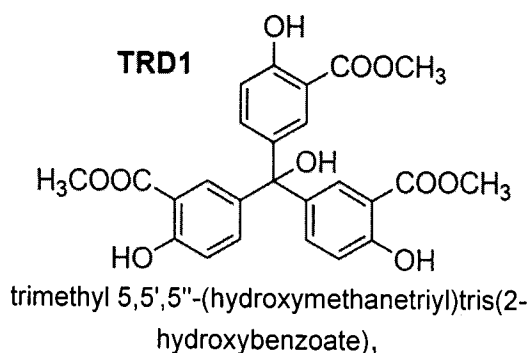
[00169] In embodiments, there is also disclosed a pharmaceutical composition comprising a therapeutically effective amount of a compound according to the present invention.

[00170] In embodiments, there is also disclosed a method to inhibit tumor growth in a subject, which comprises administering an anticancer amount of a

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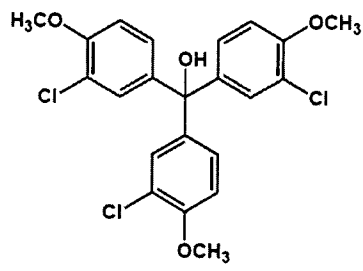
compound of the present invention, or composition according to the present invention.

[00171] In embodiments, there is also disclosed method to inhibit tumor growth in a subject, which comprises administering an anticancer amount of a compound TRD1, TRD5, TRD6, TRD7, TRD8, TRD9, TRD10, QB12, QB39, QB46, QB56, and QB57:



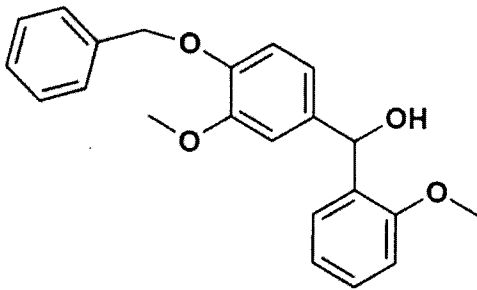
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TRD10 100 mg

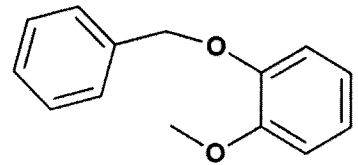


tris(3-chloro-4-methoxyphenyl)methanol ,

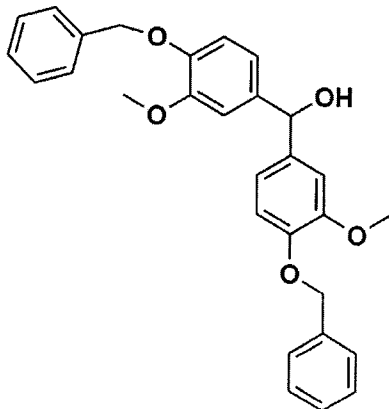
QB12



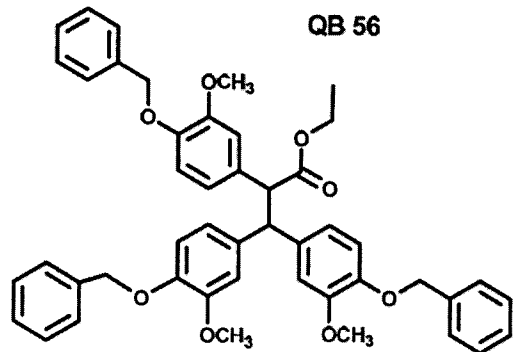
QB39



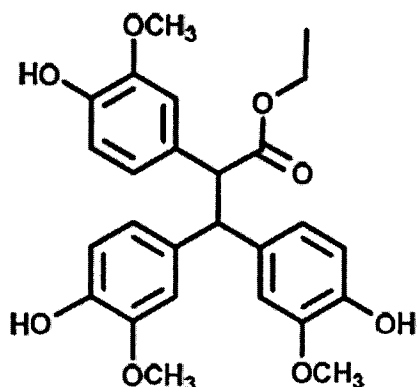
QB46



QB 56



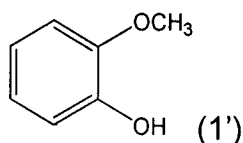
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QB 57

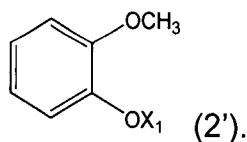
and

[00172] In embodiments the tumor may be a breast tumor, a prostate tumor, a lung tumor, a colon tumor, a liver tumor and a testes tumor.

[00173] In embodiments, there is also disclosed a process for the synthesis of a compound of formula (10') (Quebecol). Now referring to Fig. 2, the process comprises a first step of reacting a compound of formula (1')

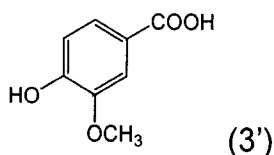


[00174] with a suitable hydroxyl protecting group, to obtain a compound of formula (2')



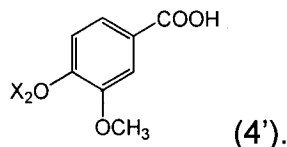
[00175] X₁ is a suitable protecting group for a hydroxyl group.

[00176] The process also comprises a second step of reacting a compound of formula (3')



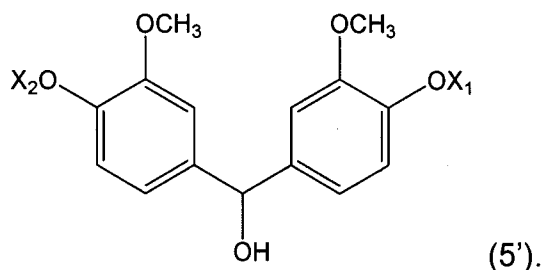
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[00177] with a suitable hydroxyl protecting group, to obtain a compound of formula (4')

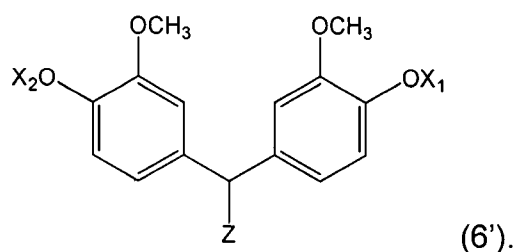


[00178] X_2 is a suitable protecting group for a hydroxyl group.

[00179] The third step of the process comprises the reaction of the compound of formula (2') with the compound of formula (4') to obtain a compound of formula (5')



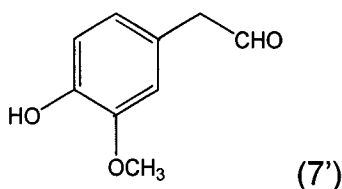
[00180] The fourth step of the process comprises reacting the compound of formula (5') with a trihalide of phosphorus, such as phosphorus tribromide (PBr₃), phosphorus trichloride (PCl₃), for example to obtain a compound of formula (6')



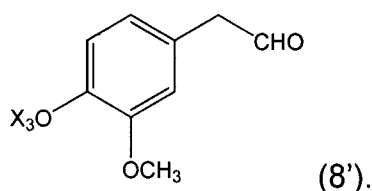
[00181] Z represents a halogen atom. Preferably, the halogen atom is Br.

[00182] The fifth step of the process comprises reacting the compound of formula (7')

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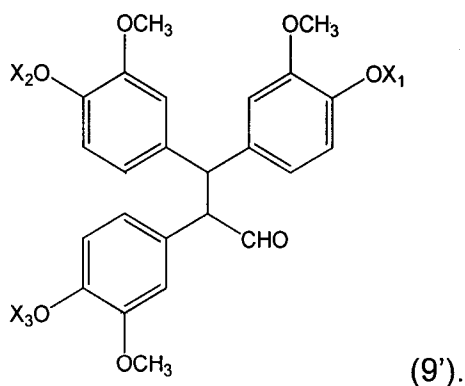


[00183] with a suitable hydroxyl protecting group, to obtain a compound of formula (8')



[00184] X_3 is a suitable protecting group for a hydroxyl group.

[00185] The sixth step of the process comprises the reaction of the compound of formula (6') with the compound of formula (8') to obtain a compound of formula (9')



[00186] Finally, the seventh step of the process comprises reducing the CHO group to a CH_2OH group, and deprotecting the compound of formula (9') to obtain a compound of formula (10') (Quebecol). X_1 , X_2 and X_3 represent suitable protecting groups for a hydroxyl groups.

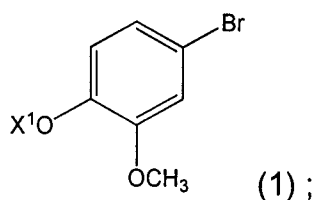
[00187] The suitable protecting groups for hydroxyl groups for X_3 may be chosen from Fmoc, triphenylmethyl chloride, and a silyl ether. Preferably, the protecting group is a silyl ether protecting group.

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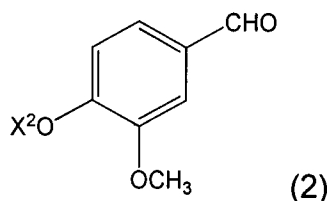
[00188] According to an embodiment of the present invention, the reduction reaction of the compound of formula (9') may be effected with NaBH_4 .

[00189] The deprotection of the compound of formula (9') may be achieved with one of tetra-*n*-butylammonium fluoride (TBAF) or trifluoroacetic acid (TFA), depending on the protecting group for a hydroxyl group chosen.

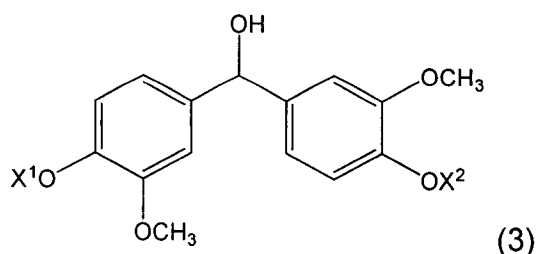
[00190] In embodiments, there is also disclosed an alternative process for the synthesis of a compound of formula (8) (Quebecol). Now referring to Fig. 3, the process comprises a first step of reacting a compound of formula (1)



[00191] with a compound of formula (2)



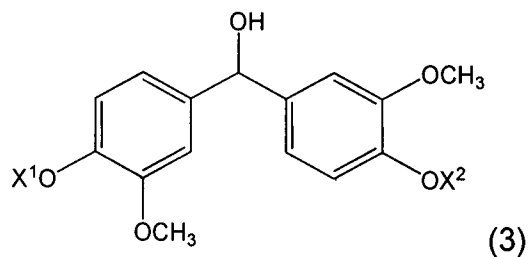
[00192] in presence of a strong base, to obtain a compound of formula (3)



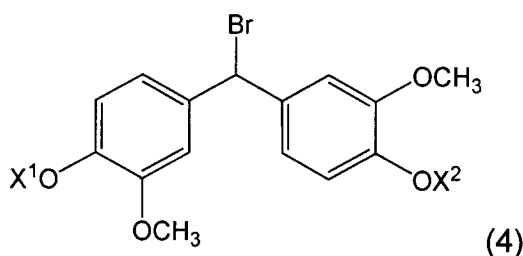
[00193] wherein X^1 and X^2 is a suitable protecting group for a hydroxyl group. According to an embodiment, the strong base may be for example *n*-butyllithium (*n*-BuLi). The reaction may take place for example in tetrahydrofuran (THF) at -78°C .

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[00194] The second step of the process involves brominating a compound of formula (3)

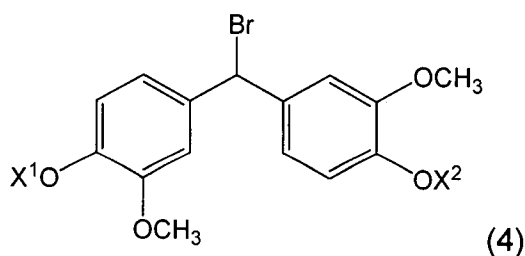


[00195] to obtain the compound of formula (4)

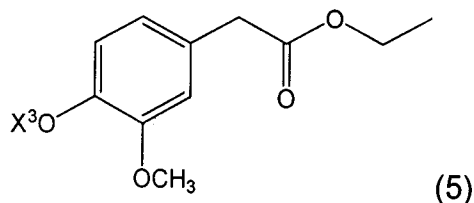


[00196] where X^1 and X^2 is a suitable protecting group for a hydroxyl group. Bromination is preferably done with acetyl bromide (CH_3COBr). The reaction may be carried out for example in benzene.

[00197] The third step involves reacting a compound of formula (4)

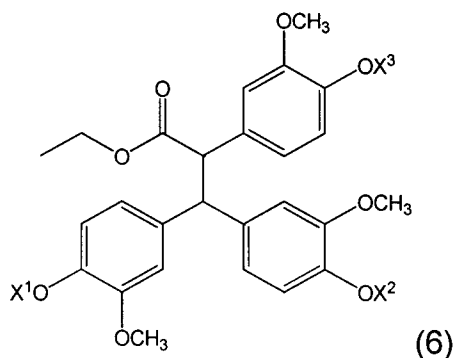


with a compound of formula (5)



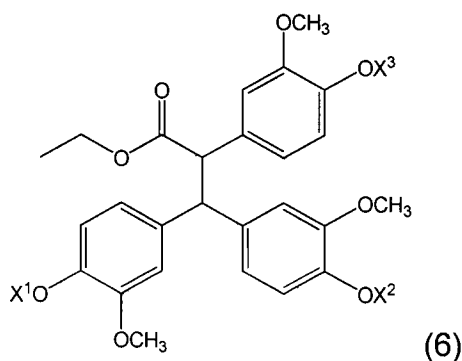
[00198] in the presence of a strong base, to obtain a compound of formula (6),

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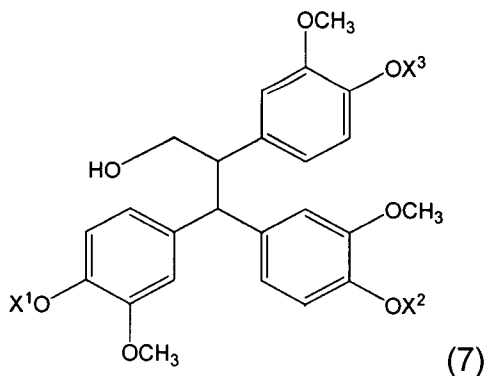


[00199] where X^1 , X^2 , and X^3 is a suitable protecting group for a hydroxyl group. The strong base may be for example lithium diisopropylamide (LDA). The reaction may be carried out in tetrahydrofuran (THF) at -78°C for example.

[00200] The fourth step involves reducing a compound of formula (6)



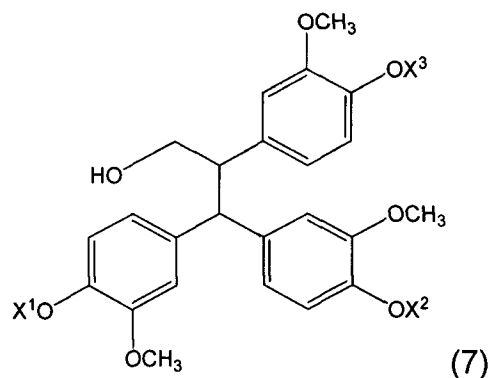
to obtain the compound of formula (7),



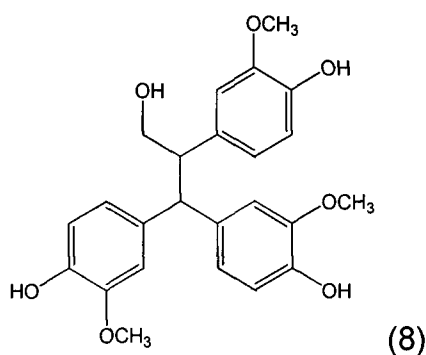
[00201] where X^1 , X^2 , and X^3 is a suitable protecting group for a hydroxyl group. The reduction may be achieved for example with lithium aluminum hydride (LiAlH_4). The reaction may be carried out in tetrahydrofuran (THF).

[00202] The fifth step involves deprotecting a compound of formula (7)

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[00203] to obtain the compound of formula (8),



[00204] where X^1 , X^2 , and X^3 is a suitable protecting group for a hydroxyl group. The deprotection may be achieved for example with ammonium formate (HCO_2NH_4) and palladium on carbon (Pd/C). The reaction may be carried out in methanol, for example.

[00205] Suitable protecting group for a hydroxyl group include but are not limited to $\text{C}_1\text{-C}_{25}$ ethers, $\text{C}_1\text{-C}_{25}$ substituted methyl ethers, $\text{C}_1\text{-C}_{25}$ substituted ethyl ethers, $\text{C}_1\text{-C}_{25}$ acyl groups, $\text{C}_1\text{-C}_{25}$ halogenated acyl groups, $\text{C}_1\text{-C}_{25}$ substituted benzyl ethers, $\text{C}_1\text{-C}_{25}$ silyl ethers, $\text{C}_1\text{-C}_{25}$ esters, $\text{C}_1\text{-C}_{25}$ carbonates, and $\text{C}_1\text{-C}_{25}$ sulfonates. Other suitable protecting group for a hydroxyl group include but are not limited to diphenylmethylchlorosilane (DPMS), tosyl, methyl, methoxymethyl, benzyloxymethyl, tetrahydropyranyl, tetrahydrofuranyl, 2-(trimethylsilyl)ethoxymethyl, dioxanyl, 1-ethoxyethyl, 1-(2-chloroethoxy)ethyl, 2,2,2-trichloroethyl, t-butyl, allyl, propargyl, benzyl, p-methoxybenzyl, diphenylmethyl, triphenylmethyl, trimethylsilyl, triethylsilyl, triisopropylsilyl, dimethylisopropylsilyl, diethylisopropylsilyl, dimethylhexylsilyl, t-

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butyldimethylsilyl, t-butyldiphenylsilyl, tribenzylsilyl, triphenylsilyl, triisopropylsilyl, diphenylmethylsilyl, benzylformate, methylcarbonyl, ethylcarbonyl, methoxymethyl carbonyl, trichloroethoxycarbonyl, benzylcarbonyl, benzyloxycarbonyl, allylsulfonyl, methanesulfonyl, and p-toluenesulfonyl.

[00206] Preferably, the suitable protecting group for a hydroxyl group is benzyl (Bn).

[00207] The present invention will be more readily understood by referring to the following examples which are given to illustrate the invention rather than to limit its scope.

EXAMPLE 1

Identification of a new compound from the process of preparation of Maple Syrup

[00208] Reagents & Materials: All solvents are either analytical grade or HPLC grade and purchased from Wilkem Scientific Co. (Pawtucket, RI). Maple syrup (grade C, 20 L) is provided by the Federation of Maple Syrup Producers of Quebec (Canada). The syrup is kept frozen until extraction when it is subjected to liquid-liquid partitioning with ethyl acetate (10 L x 3) followed by n-butanol (10 L x 3) solvents, to yield ethyl acetate (4.7 g) and butanol (108 g) extracts, respectively, after solvent removal in vacuo.

[00209] Isolation: A portion of the butanol extract (87g) is reconstituted in methanol to afford methanol soluble (36 g) and insoluble (57 g) fractions. The methanol soluble fraction is selected for further purification by repeated Sephadex-LH20 column chromatography followed by C 18 semi-preparative HPLC. First, the extract is chromatographed on 65 x 4 cm Sephadex-LH-20 column eluted with a CH₃OH-H₂O gradient system (3:7 to 1:0, v/v) to afford twelve subfractions, A1-A12. Subfraction A4 (1.6 g) is re-chromatographed on a 65 x 4 cm Sephadex-LH-20 column eluted with same gradient system (3:7 to 1:0, v/v) to afford twelve subfractions, B1-B12. Subfraction B5 (137.2 mg) is purified

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by semi-preparative HPLC (Neckman Coulter) using a Waters Sunfire C18 column (250 × 10 mm i.d., 5 μm, flow = 2 mL/min) with a gradient elution system of CH₃OH-H₂O (0.1% trifluoroacetic acid) (1:4, v/v to 1:0, v/v in 60 min) to afford compound **1** (4 mg).

[00210] **NMR:** Data is collected on a Varian 500 MHz Biospin instrument using CD₃OD as solvent.

[00211] Compound (**10**) – Quebecol, (Figs. 1 and 8) is obtained as pale yellow amorphous powder. The positive ESIMS exhibits a molecular peak at *m/z* 449.1571 [M+Na]⁺, and negative ESI shows at *m/z* 425.1979 [M-H]⁻. The ¹H NMR (in DMSO-d₆) spectrum exhibits the signals for three pairs of ABX aromatic system at δ_H 6.81 (1H, *J*=8.0 Hz, H-6), 6.67 (1H, *J*=8.0 Hz, H-5), 6.98 (1H, s, H-2); 6.56 (1H, *J*=8.0 Hz, H-6'), 6.41 (1H, *J*=8.0 Hz, H-5'), 6.78 (1H, s, H-2'); 6.60 (1H, *J*=8.0 Hz, H-6''), 6.50 (1H, *J*=8.0 Hz, H-5''), 6.56 (1H, s, H-2'') respectively, suggesting the presence of three benzene rings, which is supported by the ¹³C NMR (in DMSO-d₆) data (Table 4) and ¹H-¹H COSY spectrum analysis (Fig. 4). Three singlet signals at δ_H 3.76, 3.66 and 3.63 with three-proton density for each reveal the presence of three methoxyl groups. Additionally, one doublet signal at δ_H 4.02 (1H, *J*=10.5 Hz, H-7), two multiplet signals at δ_H 3.41 (1H, m, H-8) and 3.40 (2H, m, H-10) can be observed in the ¹H spectrum. All the proton signals are assigned to the corresponding carbons through direct ¹H-¹³C correlations in the HSQC (Table 4) spectrum, with exception of the two singlets at δ_H 8.67(1 H) and 8.43 (2 H) which are in good accordance with proton of hydroxyl group. Furthermore a CH-CH-CH₂ substructure can be deduced from COSY correlations (Fig. 4) analysis. In the HMBC spectrum, the correlations signals (Fig. 4) from δ_H 6.67 (H-5) and 3.76 (3-OCH₃) to C-3 (δ 147.72), δ_H 6.41 (H-5') and 3.66 (3'-OCH₃) to C-3' (δ 147.17), δ_H 6.50 (H-5'') and 3.63 (3''-OCH₃) to C-3'' (δ 147.08), reveals three methoxyl groups substituted on the C-3, 3' and 3'' individually. In the same HMBC experiment, correlation signals show from δ_H

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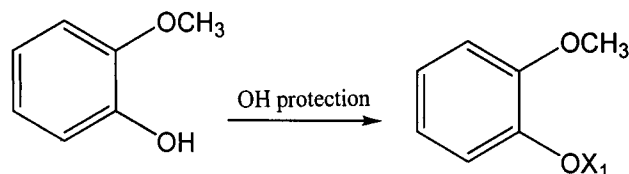
4.02 (H-7) to C-2 (112.56), C-6 (120.33) and C-1' (136.26), and from δ_{H} 6.78 (H-2') to C-8 (51.42) suggest three benzene rings are attached to the CH-CH-CH₂OH chain on C-7, C-7 and C-8 position respectively.

Table 1: ¹H and ¹³C NMR data (in DMSO-d₆, 500 and 125MHz) of compound 10

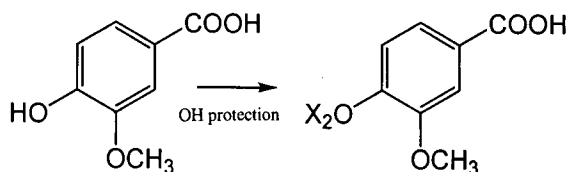
No	δ_{C}	δ_{H} (J in Hz)	No	δ_{C}	δ_{H} (J in Hz)
1	136.70	-	1'	136.26	-
2	112.56	6.98 (s)	2'	113.15	6.78 (s)
3	147.72	-	3'	147.17	-
4	144.92	-	4'	144.26	-
5	115.72	6.67 (d, 8.0)	5'	115.23	6.41 (d, 8.0)
6	120.33	6.81 (d, 8.0)	6'	121.04	6.56 (d, 8.0)
7	52.67	4.02 (d, 10.5)	1"	134.65	-
8	51.42	3.41 (m)	2"	113.90	6.78 (s)
9	64.92	3.40 (m)	3"	147.08	-
3-OCH ₃	56.14	3.76 (s)	4"	144.48	-
3'-OCH ₃	56.01	3.66 (s)	5"	115.09	6.50 (d, 8.0)
3''-OCH ₃	55.94	3.63 (s)	6"	121.77	6.60 (d, 8.0)
4-OH	-	8.64 (s)	4''-OH	-	8.43 (s)
4'-OH	-	8.43 (s)			

[00212] The absolute configuration of compound (10) is elucidated by combination of ¹H NMR analysis and computer modelling. The coupling constant of H-7 is 10.5 Hz, suggesting H-7 and H-8 are both at the axial positions, which is in accordance with S configuration. Thus, based on above findings, the structure of compound (10) is elucidated as shown in Fig. 4 to which the common name, Quebecol, has been assigned.

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EXAMPLE 2**Preparation of a compound of formula (2')***Reaction scheme of Example 2*

[00213] Compounds of formula (2') may be synthesized, for example, by using the following conditions. To a stirred solution of the corresponding commercially available phenolic compound (1.00 mmol) in acetone is added potassium carbonate (1.50 mmol) and benzyl bromide (1.10 mmol). The solution was then stirred at ambient temperature (~ 30 °C) for 16 h. The organic solvent was evaporated under reduced pressure. The residue was diluted with water and extracted with ethyl acetate. The organic layer was dried over anhydrous Na₂SO₄ and evaporated under reduced pressure. The crude compound was purified by column chromatography (ethyl acetate/hexane) to afford corresponding benzylated compound in 80-95% yields.

EXAMPLE 3**Preparation of a compound of formula (4')***Reaction scheme of Example 3*

[00214] Compounds of formula (4') may be synthesized, for example, by using the following conditions. To a stirred solution of the corresponding commercially available phenolic compound (1.00 mmol) in acetone is added potassium carbonate (1.50 mmol) and benzyl bromide (1.10 mmol). The solution

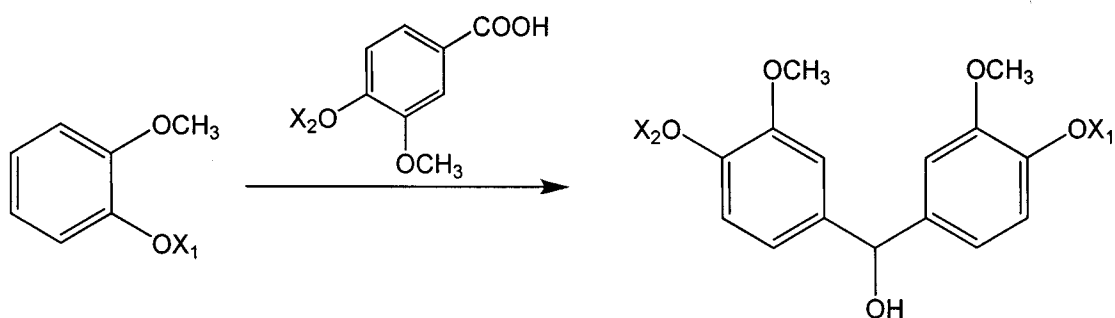
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was then stirred at ambient temperature (~ 30 °C) for 16 h. The organic solvent was evaporated under reduced pressure. The residue was diluted with water and extracted with ethyl acetate. The organic layer was dried over anhydrous Na₂SO₄ and evaporated under reduced pressure. The crude compound was purified by column chromatography (ethyl acetate/hexane) to afford corresponding benzylated compound in 80-95% yields.

EXAMPLE 4

Preparation of a compound of formula (5')

Reaction scheme of Example 4

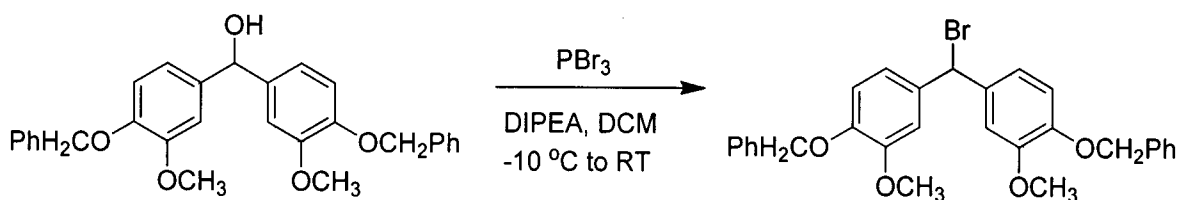


[00215] Compound (2') (80 mmol) was reacted with 4' (80 mmol) in the presence of polyphosphoric acid at 80 °C for 30 min according to the previously reported procedure (Harig et al., Eur. J. Org. Chem. 2004, 2381-2397). The crude product (80 mmol) was suspended in methanol (400 mL), which had been deacidified by passing it through basic alumina. Pyridine (0.5 mL) and palladium on charcoal (10% Pd, oxidic form) were then added and the mixture was shaken under hydrogen in a hydrogen generator. The suspension was then filtered through silica gel and the filter was washed with deacidified methanol. Removal of the solvent under reduced pressure afforded 5'.

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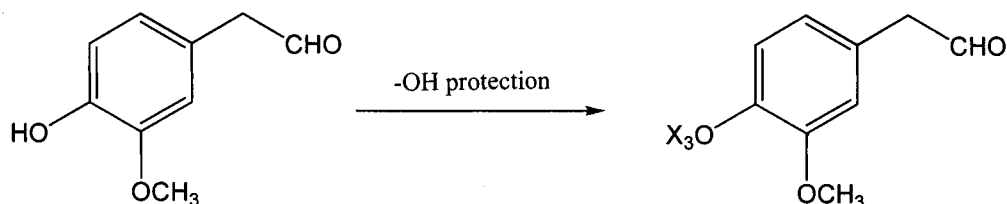
EXAMPLE 5

Preparation of a compound of formula (6') – bis(4-(benzyloxy)-3-methoxyphenyl)bromomethane

Reaction scheme of Example 5

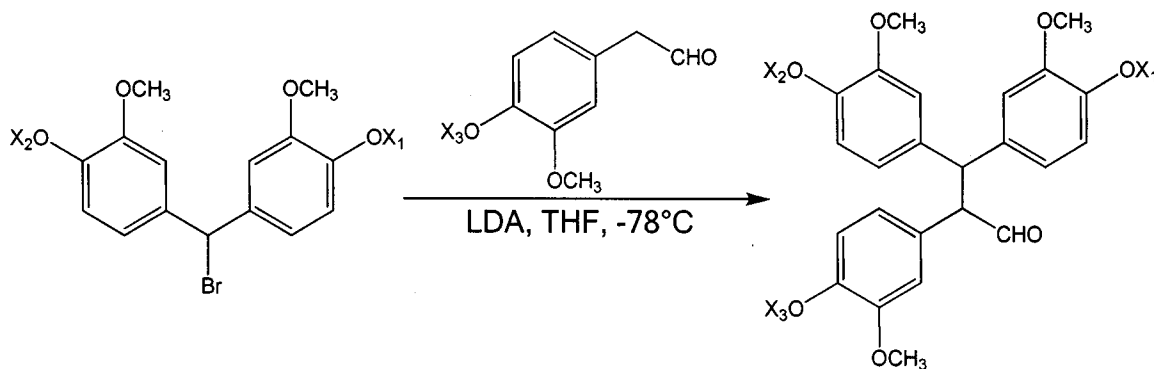
[00216] Bis(4-(benzyloxy)-3-methoxyphenyl)methanol (0.68 g, 1.5 mmol) was dissolved in dry DCM (20 mL) and then to the solution was added *N,N*-diisopropylethylamine (347 μ l, 2.0 mmol). The mixture was cooled to -10 °C. PBr₃ (176 μ l, 1.1 eq.) in dry DCM (10 mL) was added dropwise in the dark over 15 min. The reaction mixture was brought to 0 °C and was stirred for 1 h and then it was stirred at room temperature for additional 6 h. After completion of the reaction as indicated by TLC, the reaction mixture was concentrated on rotatory evaporator under reduced pressure. The residue was washed with water (10 mL) and extracted with ethyl acetate (2 \times 10 mL). The combined organic phases were dried over anhydrous sodium sulfate and concentrated. The product was purified by column chromatography over silica gel as a white solid showing a mixture of brominated compound and ketone product.

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EXAMPLE 6**Preparation of a compound of formula (8')***Reaction scheme of Example 6*

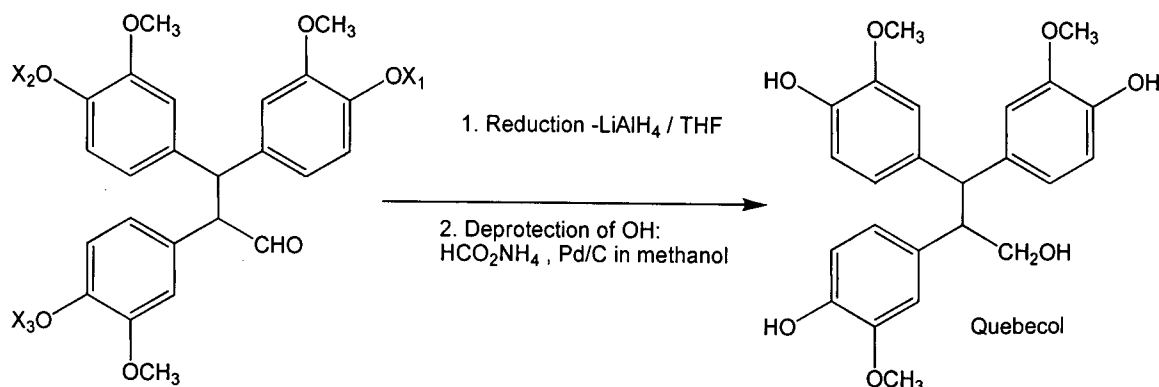
[00217] Compounds of formula (8') may be synthesized, for example, by using the following conditions. To a stirred solution of the corresponding commercially available phenolic compound (1.00 mmol) in acetone is added potassium carbonate (1.50 mmol) and benzyl bromide (1.10 mmol). The solution was then stirred at ambient temperature (~ 30 °C) for 16 h. The organic solvent was evaporated under reduced pressure. The residue was diluted with water and extracted with ethyl acetate. The organic layer was dried over anhydrous Na₂SO₄ and evaporated under reduced pressure. The crude compound was purified by column chromatography (ethyl acetate/hexane) to afford corresponding benzylated compound in 80-95% yields..

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EXAMPLE 7**Preparation of a compound of formula (9')***Reaction scheme of Example 7*

[00218] To an oven dried 50 mL RB flask, was added lithium diisopropylamine (3.85 mmol) and THF (10 mL) under N₂ and then was cooled to 0 °C. *n*-BuLi (3.85 mmol, 1.6 M solution in hexane) was added slowly to the above solution under N₂ atmosphere. The solution was then stirred for 30 min at the same temperature. The solution was cooled to -78 °C. Compound (7') in THF (5 mL) was slowly added to the reaction mixture. The stirring was continued for 15 min at the same temperature. Then, freshly prepared brominated compound (6') (0.77 mmol) in THF (5 mL) was added to the reaction mixture and the solution was stirred at the same temperature for 30 min. TLC analysis indicated complete conversion of compound (6'). The reaction mixture was allowed to reach 0 °C, quenched with cold water, and extracted into ethyl acetate. The organic layer was dried over anhydrous Na₂SO₄ and evaporated under reduced pressure. The crude compound was purified by column chromatography (ethyl acetate:hexane 30/70 v/v) to afford compound (9').

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EXAMPLE 8**Preparation of a compound of formula (10') Quebecol***Reaction scheme of Example 8*

[00219] To an oven dried 10 mL RB flask, was added lithium aluminum hydride (0.26 mmol) under N₂ atmosphere. The flask was cooled to 0 °C. THF (2 mL) was slowly added followed by compound (9') in THF (2 mL) to the flask. The solution was stirred at ambient temperature (~30°C) for 1 h. TLC analysis indicated complete conversion of compound (9'). The reaction mixture was quenched with saturated NH₄Cl solution and extracted with ethyl acetate. The organic layer was dried over anhydrous Na₂SO₄ and evaporated the volatiles. The crude compound was purified by column chromatography (ethyl acetate:hexane 50/50 v/v). To a stirred solution of the crude compound in methanol was added ammonium formate and Pd/C. The reaction mixture was stirred for 16h at ambient temperature (~30°C). TLC analysis indicated complete conversion of the crude compound. The reaction mixture was filtered through celite pad and the bed was washed with ethyl acetate. The filtrate was evaporated to dryness under reduced pressure. The crude compound was purified by column chromatography (ethyl acetate:hexane 70:30 v/v) to produce Quebecol (10') as off-white solid (yield 67%). These reactions will be conducted under conditions that will be optimized for maximum yield of product.

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EXAMPLE 9**Alternative Synthesis of Quebecol**

[00220] Fig. 3 illustrates the general procedure for the synthesis of Quebecol. Bis(4-(benzyloxy)-3-methoxyphenyl)methanol (**3**) is synthesized from the reaction of 1-(benzyloxy)-4-bromo-2-methoxybenzene (**1**) with 4-(benzyloxy)-3-methoxybenzaldehyde (**2**) in the presence of *n*-butyllithium in THF. Bromination of compound **3** in the presence of acetyl bromide in benzene generated the crude building block 4,4'-(bromomethylene)bis(1-(benzyloxy)-2-methoxybenzene) (**4**) that is used immediately for the reaction with ethyl 2-(4-(benzyloxy)-3-methoxyphenyl)acetate (**5**) in the presence of LDA to afford tribenzylated compound **6**. Subsequent reduction of ethyl ester to alcohol **7** in the presence of lithium aluminum hydride followed by debenzylation with ammonium formate and Pd/C afforded Quebecol (**8**).

[00221] Experimental Procedures

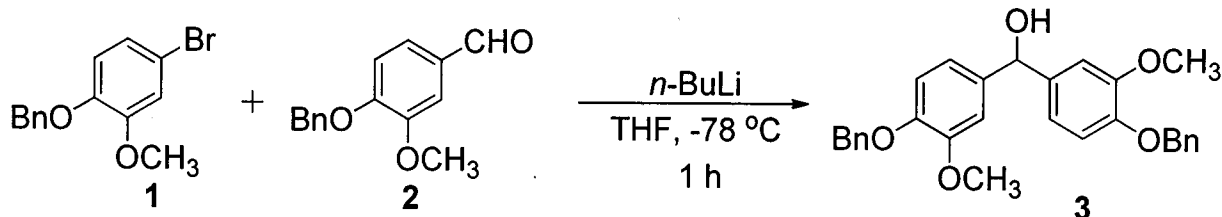
[00222] Generalized procedure for the synthesis of compounds 1-(benzyloxy)-4-bromo-2-methoxybenzene (1), 4-(benzyloxy)-3-methoxybenzaldehyde (2), and ethyl 2-(4-(benzyloxy)-3-methoxyphenyl)acetate (5).

[00223] To a stirred solution of the corresponding commercially available phenol (1.00 mmol) in acetone is added potassium carbonate (1.50 mmol) and benzyl bromide (1.10 mmol). The solution is then stirred at ambient temperature (~ 30 °C) for 16 h. The organic solvent is evaporated under reduced pressure. The residue is diluted with water and extracted with ethyl acetate. The organic layer is dried over anhydrous Na₂SO₄ and evaporated under reduced pressure. The crude compound is purified by column chromatography (ethyl acetate/hexane) to afford corresponding benzylated compound in 80-95% yields.

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[00224] Synthesis of Bis(4-(benzyloxy)-3-methoxyphenyl)methanol (3)

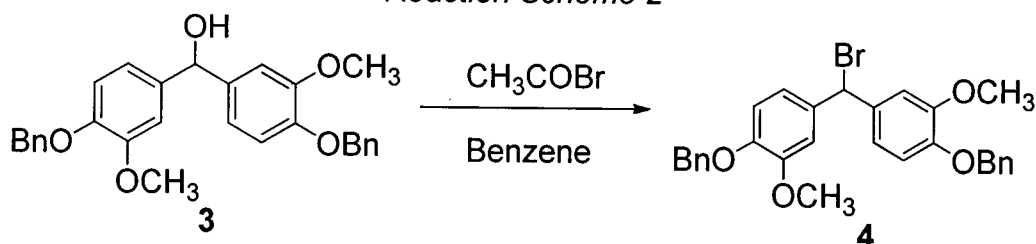
Reaction scheme 1



[00225] To a stirred solution of 1-(benzyloxy)-4-bromo-2-methoxybenzene (1) (10.23 mmol) in THF (25 mL) at -78 °C slowly is added *n*-butyllithium (*n*-BuLi, 10.74 mmol, 1.6 M solution in hexane) under N₂. The mixture is stirred for 30 min at the same temperature. 4-(Benzyloxy)-3-methoxybenzaldehyde (2, 11.26 mmol) in THF (25 mL) is slowly added to the solution over a period of 5 min. Then the solution is stirred for 30 min at -78 °C. TLC indicated complete conversion of 1 to the product. The reaction mixture is allowed to reach 0 °C, quenched with saturated NH₄Cl solution, and extracted with ethyl acetate. The organic layer is dried over anhydrous Na₂SO₄ and evaporated under reduced pressure. The crude compound is purified by column chromatography (ethyl acetate:hexane 35/65 v/v) to afford 3 as a white solid (yield 55%). Figs. 4A and B show ¹H NMR spectrum and MS spectrum for compound 3, respectively.

[00226] Synthesis of 4,4'-(Bromomethylene)bis(1-(benzyloxy)-2-methoxybenzene) (4)

Reaction Scheme 2



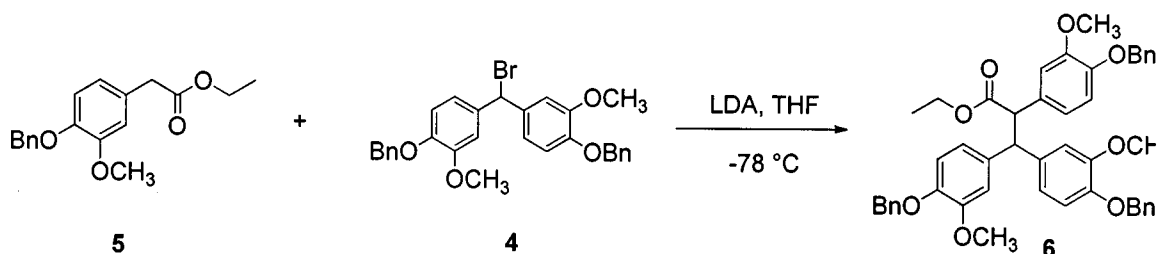
[00227] To a slurry of compound 3 (4.82 mmol) in benzene (30 mL) is added acetyl bromide (14.47 mmol) at ambient temperature (~30 °C) under N₂. The solution is stirred for 5 h. After completion of the reaction, the solvent is

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evaporated, and the residue is azeotroped with toluene (2 times). The crude compound is washed with hexane (2 times) to remove traces of acetic acid and then dried to yield brominated compound **4** as light pink colour sticky solid, which is used for the next step without further purification (yield 56%).

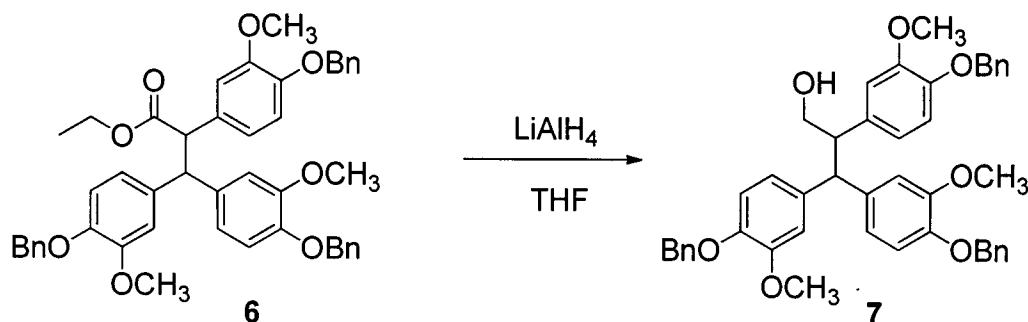
[00228] Synthesis of Ethyl 2,3,3-tris(4-(benzyloxy)-3-methoxyphenyl)propanoate (6)

Reaction Scheme 3



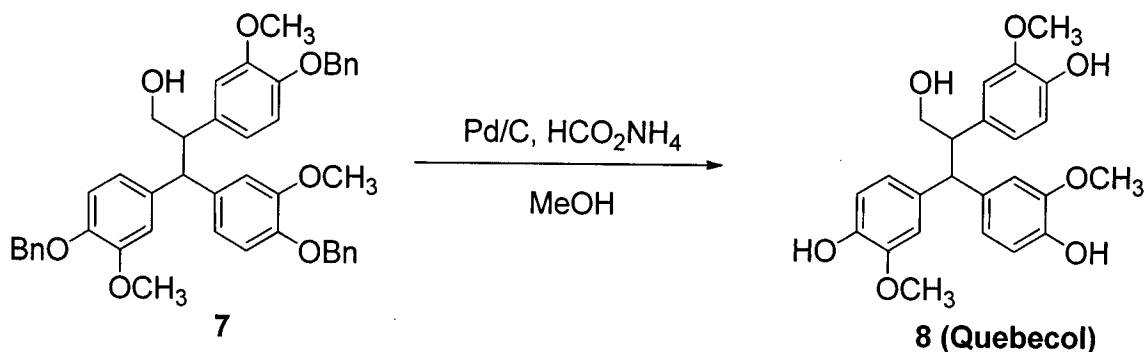
[00229] To an oven dried 50 mL RB flask, is added diisopropylamine (3.85 mmol) and THF (10 mL) under N₂ and then is cooled to 0 °C. *n*-BuLi (3.85 mmol, 1.6 M solution in hexane) is added slowly to the above solution under N₂ atmosphere. The solution is then stirred for 30 min at the same temperature. The solution is cooled to -78 °C. Ethyl 2-(4-(benzyloxy)-3-methoxyphenyl)acetate **5** (3.08 mmol) in THF (5 mL) is slowly added to the reaction mixture. The stirring is continued for 15 min at the same temperature. Then, freshly prepared brominated compound **4** (0.77 mmol) in THF (5 mL) is added to the reaction mixture and the solution is stirred at the same temperature for 30 min. TLC analysis indicated complete conversion of compound **4**. The reaction mixture is allowed to reach 0 °C, quenched with cold water, and extracted into ethyl acetate. The organic layer is dried over anhydrous Na₂SO₄ and evaporated under reduced pressure. The crude compound is purified by column chromatography (ethyl acetate:hexane 30/70 v/v) to afford **6** as pale yellow liquid (yield 30%). Figs. 5A to C show ¹H NMR spectrum and Fig. 5D shows MS spectrum for compound **6**.

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[00230] Synthesis of 2,3,3-Tris(4-(benzyloxy)-3-methoxyphenyl)propan-1-ol (7)*Reaction scheme 4*

[00231] To an oven dried 10 mL RB flask, is added lithium aluminum hydride (0.26 mmol) under N_2 atmosphere. The flask is cooled to 0 °C. THF (2 mL) is slowly added followed by ester **6** in THF (2 mL) to the flask. The solution is stirred at ambient temperature (~30 °C) for 1 h. TLC analysis indicated complete conversion of ester **6**. The reaction mixture is quenched with saturated NH_4Cl solution and extracted with ethyl acetate. The organic layer is dried over anhydrous Na_2SO_4 and evaporated the volatiles. The crude compound is purified by column chromatography (ethyl acetate:hexane 50/50 v/v) to yield **7** as a colorless liquid (yield 76%). Figs. 6A to C show ^1H NMR spectrum and Fig. 6D shows MS spectrum for compound **7**.

File No. P1777PC00

[00232] Synthesis of 4,4',4''-(3-Hydroxypropane-1,1,2-triyl)tris(2-methoxyphenol) (8, Quebecol)*Reaction scheme 5*

[00233] To a stirred solution of **7** in methanol is added ammonium formate and Pd/C. The reaction mixture is stirred for 16 h at ambient temperature (~ 30 °C). TLC analysis indicated complete conversion of **7**. The reaction mixture is filtered through celite pad and the bed is washed with ethyl acetate. The filtrate is evaporated to dryness under reduced pressure. The crude compound is purified by column chromatography (ethyl acetate:hexane 70:30 v/v) to produce Quebecol (**8**) as off-white solid (yield 67%).

EXAMPLE 10**Cytotoxicity of Quebecol and 19 Quebecol analogs against breast cancer cells**

[00234] MTS salt [3-(4,5-dimethylthiazol-2-yl)-5-(3-carboxymethoxyphenyl)-2-(4-sulfenyl)-2H-tetrazolium salt] and etoposide standard are obtained from Sigma-Aldrich. Quebecol is previously isolated in our laboratory as reported (Li and Seeram, 2011) and several analogs are synthesized (see Fig. 9A and B for codes and structures of the compounds).

[00235] Human breast cancer cell lines MCF-7 (estrogen receptor (ER) positive) and MDA-MB-231 (ER negative) are obtained from American Type Culture Collection (Rockville, USA). MCF-7 cells are grown in EMEM medium supplemented with 10% v/v fetal bovine serum, 2% v/v HEPES, 1% v/v

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nonessential amino acids, 1% v/v L-glutamine and 1% v/v antibiotic solution (Sigma). MDA-MB-231 cells are grown in EMEM medium supplemented with 10% v/v fetal bovine serum, and 1% v/v antibiotic solution. Cells are maintained at 37 °C in an incubator under a 5% CO₂/95% air atmosphere at constant humidity. The pH of the culture medium is determined using pH indicator paper (pHydrion™ Brilliant, pH 5.5-9.0, Micro Essential Laboratory, NY, USA) inside the incubator. Cells are counted using a hemacytometer and are plated at 5000 cells per well, in a 96-well format for 24 h prior to compounds addition. All of the test samples are solubilized in DMSO (<0.5 % in the culture medium) and are filter sterilized (0.2 µm) prior to addition to the culture media. Control cells are also run in parallel and subjected to the same changes in medium with a 0.5 % DMSO. In addition, cells are treated as indicated above for 24, 48 or 72 h.

[00236] At the end of each day of treatment with serially diluted test samples (ranging from 1-200 µg/mL concentrations), 20 µL of the MTS reagent, in combination with the electron coupling agent, phenazine methosulfate, is added to the wells and cells are incubated at 37°C in a humidified incubator for 3 h. Absorbance at 490 nm (OD₄₉₀) is monitored with a spectrophotometer (SpectraMax M2, Molecular Devices Corp., operated by SoftmaxPro v.4.6 software, Sunnyvale, CA, USA), to obtain the number of cells relative to control populations. The results are expressed as the concentration that inhibit growth of cell by 50% vs. control cells (control medium used as negative control), IC₅₀. Data are presented as the mean ± S.D. of three separated experiments on each cell line (n = 2 plates per experiment; 2 wells per treatment per time point). Tamoxifen is used as positive control and provided consistent IC₅₀ values of 16.4 ± 1.1 µg/mL for MCF-7 cells and 10.0 ± 1.4 µg/mL for MDA-MB 231 cells at 72 h of treatment.

[00237] Quebecol and its analogs are evaluated for antiproliferative activity in both concentration (ranging from 1-200 µg/mL) and time (at 24, 48, 72, and 96 h) dependent manners by MTS assay. Overall, a clear dose-antiproliferative

File No. P1777PC00

response is observed in most of compounds. The attached Tables 2 and 3 show the IC₅₀ values of all compounds on breast cancer cell lines at different times (Tables 3A and B: concentrations in µg/mL and Tables 4A and B: concentrations in µM). Most of analogs inhibited proliferation of MCF-7 and MDA-MB 231 cell lines compared to the control cells (0.5 % DMSO) in time-dependent manner suggesting that these analogs may have a potential as chemopreventive and chemotherapeutic agents on breast cancer. It should be noted that both the compounds showed similar effects to both MCF-7 and MDA-MB 231 cells.

[00238] As shown in Table 2, TRD8 and TRD7 exhibited the highest antiproliferative activities with IC₅₀ values ranging from 10.6-24.8 µg/mL against MCF-7 cells and 17.47-24.0 µg/mL against MDA-MB 231 cells after 72 h of treatment, respectively. These analogs showed better activity on cancer cell lines when compared to Quebecol (46.3±2.1 and 50.7±2.4 µg/mL against the MCF-7 and MDA-MB 231 cells, respectively). Moreover, these two analogs showed IC₅₀ values similar to Tamoxifen used as positive control (Table 2).

[00239] Moderate activity, close to the values of Quebecol is showed by other analogs such as QB46, TRD6, QB12, TRD5, and TRD10 with IC₅₀ values ranging from 44.8-78.9 and 62.9-77.4 µg/mL against the MCF-7 and MDA-MB 231 cells, respectively (Table 2).

[00240] Finally, analogs such as TRD1, TRD9, QB57, and QB56 showed slight cytotoxicity with IC₅₀ values >80 µg/mL.

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Table 2. Cytotoxic effects of Quebecol & Quebecol analogs against human breast cancer cell lines after 72 h treatment.

Compounds (g/mL)	MCF-7	MDA-MB-231
	IC ₅₀ ^a	IC ₅₀ ^a
Tamoxifen	16.4 ± 1.1	10.0 ± 1.4
TRD1	84.1 ± 3.4	93.9 ± 2.4
TRD2	n.d.	n.d.
TRD3	n.d.	n.d.
TRD4	n.d.	n.d.
TRD5	75.9 ± 1.9	81.6 ± 2.6
TRD6	60.4 ± 1.9	75.2 ± 1.8
TRD7	24.8 ± 1.5	24.0 ± 2.1
TRD8	10.6 ± 2.1	17.4 ± 1.7
TRD9	91.4 ± 3.4	97.8 ± 3.5
TRD10	73.5 ± 1.8	77.4 ± 2.7
TRD11	n.d.	n.d.
QB0	n.d.	n.d.
QB12	65.0 ± 1.4	67.2 ± 1.6
QB39	n.d.	n.d.
QB46	44.8 ± 1.7	62.9 ± 2.1
QB48	n.d.	n.d.
QB49	n.d.	n.d.
QB56	138.1 ± 3.1	136.0 ± 3.7
QB57	n.d.	85.1 ± 1.8
QUEBECOL Natural	46.3 ± 2.1	50.7 ± 2.4
QUEBECOL Synthetic	44.4 ± 1.8	48.5 ± 1.9

^a IC₅₀ (in µg/mL) is defined as the concentration required to achieve 50% inhibition over control cells (DMSO 0.5%); IC₅₀ values are shown as mean ± S.D. from three independent experiments; n.d. = not detected.

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Table 3A. IC₅₀ concentrations in µg/mL

Code	MCF-7 24 h	SD	MCF-7 48 h	SD	MCF-7 72 h	SD	MCF-7 96 h	SD
Tamoxifen	24.6	1.4	17.9	1.5	16.4	1.1	14.2	1.3
TRD1	94.2	2.6	89.1	2.3	84.1	3.4	78.3	1.9
TRD2	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
TRD3	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
TRD4	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
TRD5	n.d.	n.d.	80.9	2.0	75.9	1.9	66.5	2.3
TRD6	78.7	1.9	66.1	2.7	60.4	1.9	43.6	2.7
TRD7	42.3	1.9	32.7	1.7	24.8	1.5	23.2	1.9
TRD8	42.0	1.5	21.8	1.7	10.6	2.1	7.0	1.9
TRD9	110.7	4.3	102.5	2.0	91.4	3.4	82.3	2.2
TRD10	87.0	2.4	81.9	2.4	73.5	1.8	54.4	2.5
TRD11	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
QB0	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
QB12	n.d.	n.d.	n.d.	n.d.	65.0	1.4	59.6	1.3
QB39	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
QB46	73.2	2.7	59.5	1.9	44.8	1.7	36.8	1.9
QB48	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
QB49	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
QUEBECOL	65.9	2.0	53.5	2.2	46.3	2.1	37.4	1.7
QB56	n.d.	n.d.	n.d.	n.d.	138.1	3.1	124.9	3.8
QB57	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
QUEBECOL SYNTHETIC	67.2	2.2	54.2	1.8	44.4	1.8	38.0	1.9

IC₅₀ (in µg/mL) is defined as the concentration required to achieve 50% inhibition over control cells (DMSO 0.5%); IC₅₀ values are shown as mean ± S.D. from three independent experiments; n.d. = not detected.

File No. P1777PC00

Table 3B. IC₅₀ concentrations in µg/mL

Code	MDA-MB-231 24 h		MDA-MB-231 48 h		MDA-MB-231 72 h		MDA-MB-231 96 h	
	Mean	SD	Mean	SD	Mean	SD	Mean	SD
Tamoxifen	26.0	1.0	19.7	1.5	10.0	1.4	7.1	1.5
TRD1	n.d.	n.d.	n.d.	n.d.	93.9	2.4	89.4	2.3
TRD2	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
TRD3	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
TRD4	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
TRD5	n.d.	n.d.	n.d.	n.d.	81.6	2.6	78.2	2.2
TRD6	n.d.	n.d.	n.d.	n.d.	75.2	1.8	71.5	1.6
TRD7	36.4	1.6	30.4	1.9	24.0	2.1	15.6	1.9
TRD8	43.5	2.0	27.8	1.8	17.4	1.7	11.2	2.0
TRD9	114.9	2.3	107.5	2.5	97.8	3.5	89.9	2.5
TRD10	n.d.	n.d.	85.8	2.1	77.4	2.7	68.7	2.0
TRD11	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
QB0	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	24.0	0.7
QB12	n.d.	n.d.	n.d.	n.d.	67.2	1.6	58.0	2.1
QB39	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
QB46	78.3	1.3	72.4	2.8	62.9	2.1	53.0	1.7
QB48	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	51.2	1.7
QB49	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
QUEBECOL	70.9	2.3	57.0	1.8	50.7	2.4	43.3	1.6
QB56	n.d.	n.d.	143.3	2.8	136.0	3.7	129.3	3.9
QB57	n.d.	n.d.	n.d.	n.d.	85.1	1.8	73.8	2.9
QUEBECOL SYNTHETIC	72.5	2.0	55.6	2.2	48.5	1.9	43.8	1.7

IC₅₀ (in µg/mL) is defined as the concentration required to achieve 50% inhibition over control cells (DMSO 0.5%); IC₅₀ values are shown as mean ± S.D. from three independent experiments; n.d. = not detected.

File No. P1777PC00

Table 4A. IC₅₀ concentrations in μM

Code	MCF-7		MCF-7		MCF-7		MCF-7	
	24 h	SD	48 h	SD	72 h	SD	96 h	SD
Tamoxifen	66.2	3.8	48.3	4.1	44.1	2.9	38.2	3.5
TRD1	195.2	5.3	184.7	4.8	174.3	7.1	162.4	3.9
TRD2	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
TRD3	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
TRD4	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
TRD5	n.d.	n.d.	195.2	4.8	183	4.7	160.4	5.5
TRD6	197.5	4.8	165.8	6.7	151.6	4.8	109.4	6.7
TRD7	104.6	4.7	80.9	4.3	61.3	3.8	57.3	4.6
TRD8	96.7	3.5	50.1	3.9	24.3	4.9	16.2	4.3
TRD9	188.6	7.3	174.6	3.4	155.7	5.8	140.2	3.8
TRD10	191.7	5.3	180.4	5.2	162	3.9	119.9	5.6
TRD11	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
QB0	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
QB12	n.d.	n.d.	n.d.	n.d.	185.4	4.1	170.2	3.7
QB39	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
QB46	160.4	5.9	130.4	4.2	98.1	3.7	80.7	4.1
QB48	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
QB49	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
QUEBECOL Natural	154.6	4.6	125.5	5.2	108.8	4.9	87.9	3.9
QB56	n.d.	n.d.	n.d.	n.d.	188.1	4.2	170.1	5.2
QB57	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
QUEBECOL SYNTHETIC	157.7	5.1	127.2	4.2	104.3	4.2	89.1	4.5

IC₅₀ (in μM) is defined as the concentration required to achieve 50% inhibition over control cells (DMSO 0.5%); IC₅₀ values are shown as mean \pm S.D. from three independent experiments; n.d. = not detected.

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Table 4B. IC₅₀ concentrations in μM

Code	MDA-MB-231		MDA-MB-231		MDA-MB-231		MDA-MB-231
	24 h	SD	48 h	SD	72 h	SD	96 h
Tamoxifen	70.1	2.6	53.1	4.1	26.8	3.9	19
TRD1	n.d.	n.d.	n.d.	n.d.	194.6	5	185.4
TRD2	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
TRD3	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
TRD4	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
TRD5	n.d.	n.d.	n.d.	n.d.	196.8	6.2	188.7
TRD6	n.d.	n.d.	n.d.	n.d.	188.7	4.6	179.4
TRD7	90.0	3.9	75.3	4.8	59.4	5.1	38.7
TRD8	100.2	4.6	63.9	4.2	40	3.8	25.8
TRD9	195.7	3.9	183.1	4.3	166.6	6	153.2
TRD10	n.d.	n.d.	189.2	4.6	170.5	5.9	151.3
TRD11	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
QB0	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	193.4
QB12	n.d.	n.d.	n.d.	n.d.	191.8	4.6	165.4
QB39	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
QB46	171.5	2.9	158.6	6.1	137.8	4.6	116.2
QB48	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	185.3
QB49	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
QUEBECOL Natural	166.4	5.3	133.8	4.2	119.1	5.7	101.6
QB56	n.d.	n.d.	195.2	3.8	185.3	5	176.1
QB57	n.d.	n.d.	n.d.	n.d.	181.9	3.9	157.6
QUEBECOL SYNTHETIC							
	170.1	4.7	130.4	5.1	113.9	4.5	102.8

IC₅₀ (in μM) is defined as the concentration required to achieve 50% inhibition over control cells (DMSO 0.5%); IC₅₀ values are shown as mean \pm S.D. from three independent experiments; n.d. = not detected.

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Table 5A.

Code	MCF-7		MCF-7		MCF-7		MCF-7	
	24 h	SD	48 h	SD	72 h	SD	96 h	SD
Tamoxifen	66.2	3.8	48.3	4.1	44.1	2.9	38.2	3.5
QUEBECOL	154.6	4.6	125.5	5.2	108.8	4.9	87.9	3.9
T+Q (1:1)	141.5	3.8	123.4	2.9	108.1	3.8	77.1	4.7

Table 5B.

Code	MDA-MB-231		MDA-MB-231		MDA-MB-231		MDA-MB-231	
	24 h	SD	48 h	SD	72 h	SD	96 h	SD
Tamoxifen	70.1	2.6	53.1	4.1	26.8	3.9	19	4.1
QUEBECOL	166.4	5.3	133.8	4.2	119.1	5.7	101.6	3.7
T+Q (1:1)	152.9	4.2	125.7	2.7	105.3	4.1	78.1	3.2

IC₅₀ (in μM) is defined as the concentration required to achieve 50% inhibition over control cells (DMSO 0.5%); IC₅₀ values are shown as mean \pm S.D. from three independent experiments; n.d. = not detected. T (starting concentration with 50 μM) and Q (start concentration with 200 μM).

[00241] As a last part of this example, the possible synergistic effects of Quebecol and tamoxifen are evaluated. Tables 5 A and B shows the IC₅₀ values of these compounds of the combination (1:1) of both compounds. The data did not show any significant enhanced effects of the combination when compared to the compounds alone.

[00242] Cancer is a leading cause of death worldwide. The current study investigated the *in vitro* anticancer activities of a process-derived phenolic compound, Quebecol, present in maple syrup and 19 different analogs. It should be noted that both natural and synthetic Quebecol showed similar activity.

[00243] Given that these compounds have never been investigated for their anticancer potential, their cytotoxic effects against breast cancer lines (MCF-7 and MDA-MB 321) is investigated. The compounds are evaluated for both time and concentration dependent effects.

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[00244] Two analogs, TRD8 and TRD7, exerted the highest antiproliferative activities against MCF-7 cells and MDA-MB 231 cells after 72 h of treatment, respectively. Notably, this cytotoxic activity on both breast cancer cell lines is higher than exerted by Quebecol, and very similar to the activity exerted by Tamoxifen, used as positive control. Other analogs such as QB46, TRD6, QB12, TRD5, and TRD10 showed a moderate cytotoxic activity.

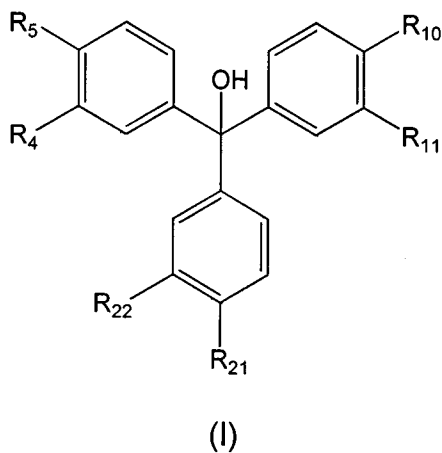
[00245] The results indicate, for the first time, that Quebecol and some of its analogs exert cytotoxic effects on breast cancer cell lines in both time and concentration dependent manners, suggesting that they may have potential as cancer chemopreventive and/or chemotherapeutic agents. Quebecol has previously been shown to have cytotoxic effect on colon cancer cell lines (Gonzales-Sarrias et al. *F. Func. Food.* 4, 1, 185-196, 2011). The highest activity is exerted by two analogs TRD8 and TRD7.

[00246] While preferred embodiments have been described above and illustrated in the accompanying drawings, it will be evident to those skilled in the art that modifications may be made without departing from this disclosure. Such modifications are considered as possible variants comprised in the scope of the disclosure.

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CLAIMS:

1. A compound of formula (I):

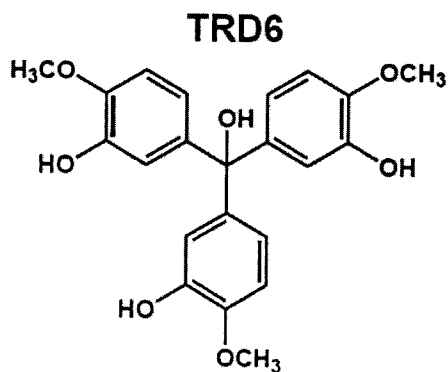


wherein

 R_5 , R_{10} , and R_{21} are OCH_3 , R_4 , R_{11} , and R_{22} are independently chosen from OH, Cl, Br, F, CF_3 , CH_3 and CHO,

pharmaceutically acceptable salt, racemic mixture, enantiomer, diastereoisomer, isomer, and tautomer thereof.

2. A compound of formula TRD6:

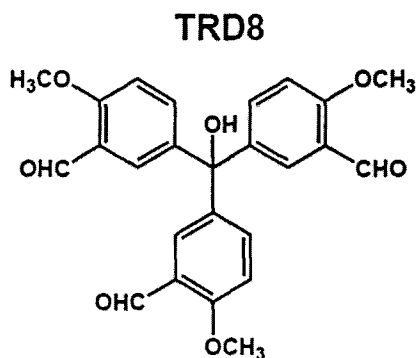


5,5',5''-(hydroxymethanetriyl)tris(2-methoxyphenol)

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pharmaceutically acceptable salt, racemic mixture, enantiomer, diastereoisomer, isomer, and tautomer thereof.

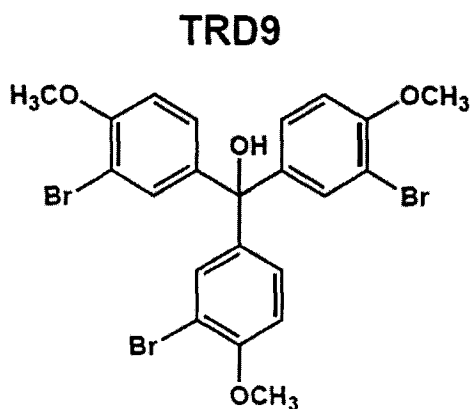
3. A compound of formula TRD8:



5,5'-(hydroxymethanetriyl)tris(2-methoxybenzaldehyde) ,

pharmaceutically acceptable salt, racemic mixture, enantiomer, diastereoisomer, isomer, and tautomer thereof.

4. A compound of formula TRD9:



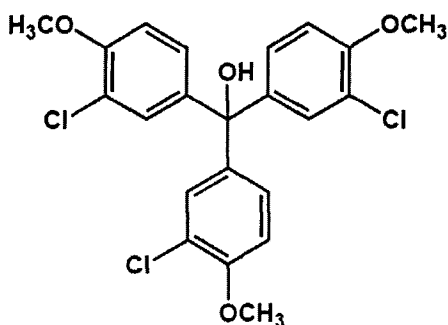
tris(3-bromo-4-methoxyphenyl)methanol ,

pharmaceutically acceptable salt, racemic mixture, enantiomer, diastereoisomer, isomer, and tautomer thereof.

5. A compound of formula TRD10:

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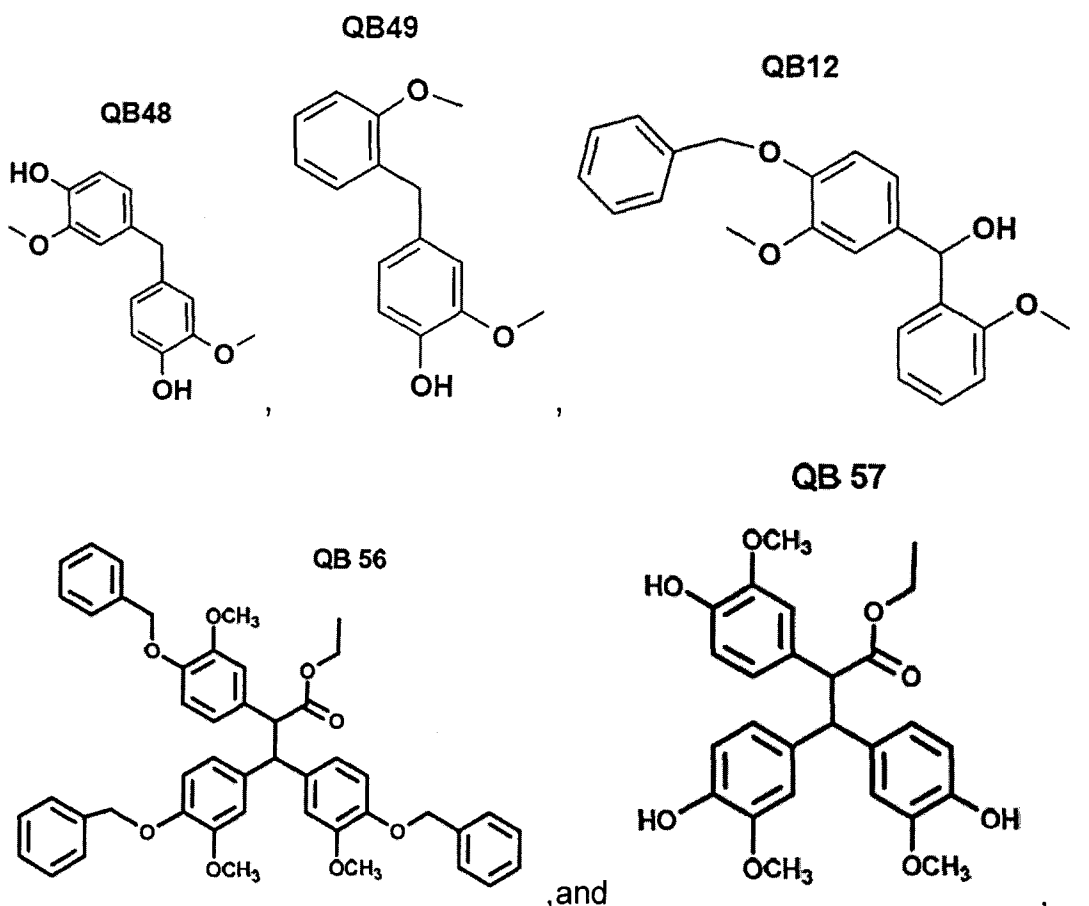
TRD10



tris(3-chloro-4-methoxyphenyl)methanol ,

pharmaceutically acceptable salt, racemic mixture, enantiomer, diastereoisomer, isomer, and tautomer thereof.

6. A compound of formula QB12, QB48, QB49, QB56, and QB57:



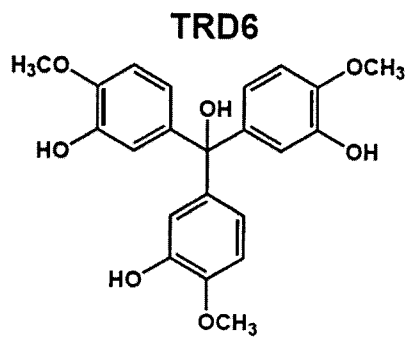
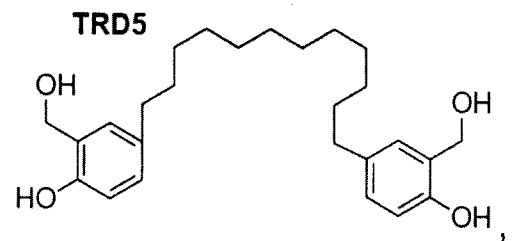
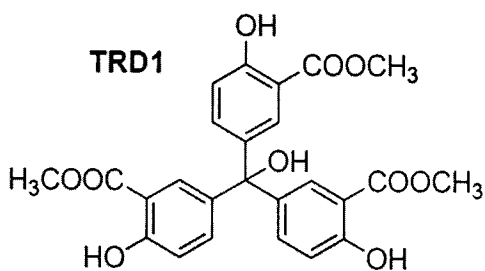
pharmaceutically acceptable salt, racemic mixture, enantiomer, diastereoisomer, isomer, and tautomer thereof.

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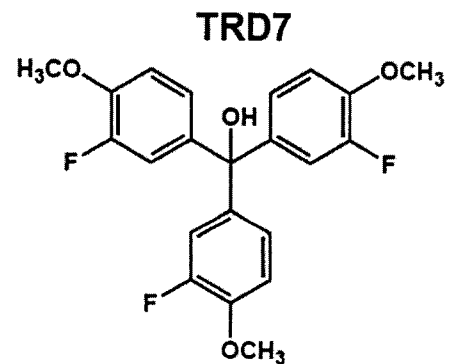
7. A pharmaceutical composition comprising a therapeutically effective amount of a compound according to any one of claims 1 - 6.

8. A method to inhibit tumor growth in a subject, which comprises administering a composition according to claim 7.

9. A method to inhibit tumor growth in a subject, which comprises administering an anticancer amount of a compound chosen from TRD1, TRD5, TRD6, TRD7, TRD8, TRD9, TRD10, QB12, QB39, QB46, QB56, and QB57:



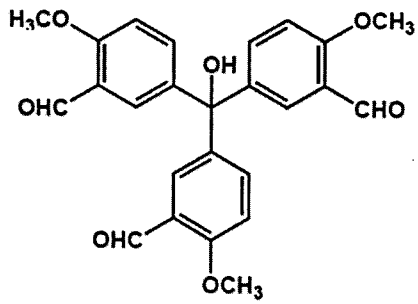
5,5',5''-(hydroxymethanetriyl)tris(2-methoxyphenol) ,



tris(3-fluoro-4-methoxyphenyl)methanol ,

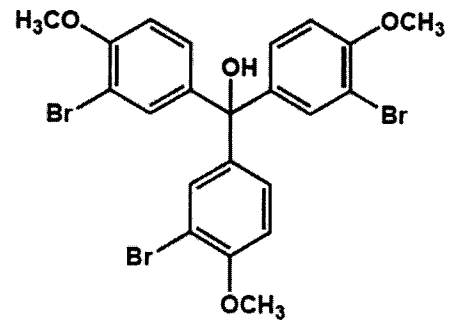
File No. P1777PC00

TRD8



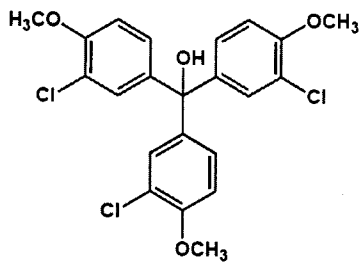
5,5'-(hydroxymethanetriyl)tris(2-methoxybenzaldehyde) ,

TRD9



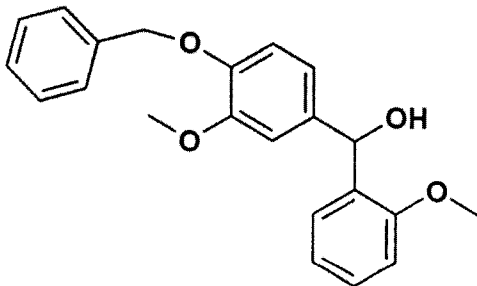
tris(3-bromo-4-methoxyphenyl)methanol ,

TRD10

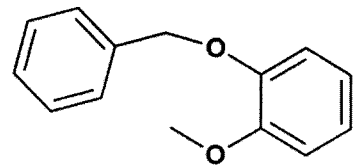


tris(3-chloro-4-methoxyphenyl)methanol ,

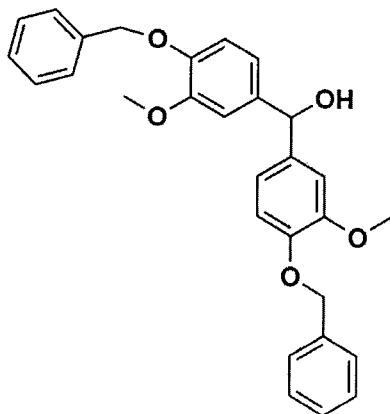
QB12



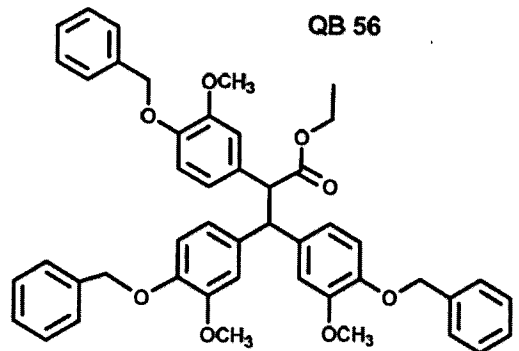
QB39



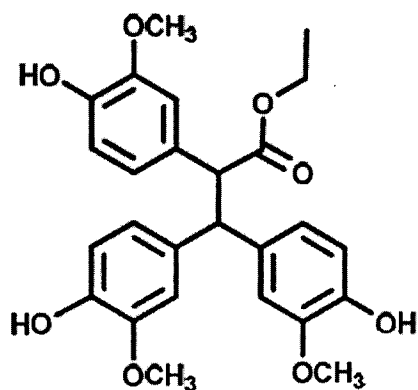
QB46



QB 56



File No. P1777PC00

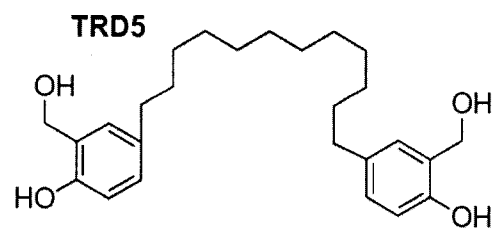
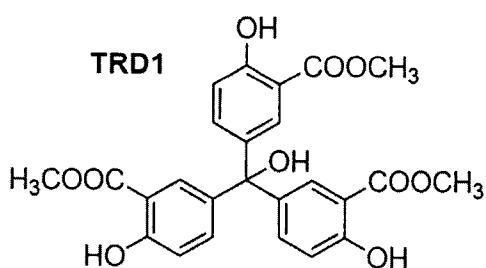
QB 57

and

10. Use of a compound as claimed in any one of claims 1 - 6, for the preparation of a medicament for the treatment of cancer.

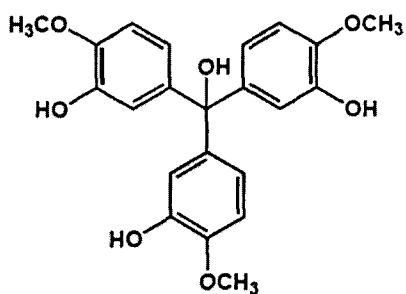
11. Use of a compound as claimed in any one of claims 1 - 6, for the treatment of cancer.

12. Use of a compound chosen from TRD1, TRD5, TRD6, TRD7, TRD8, TRD9, TRD10, QB12, QB39, QB46, QB56, and QB57 to inhibit tumor growth in a subject:



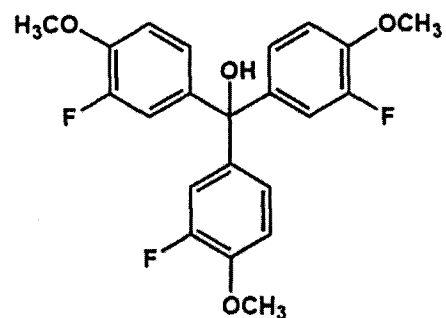
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TRD6



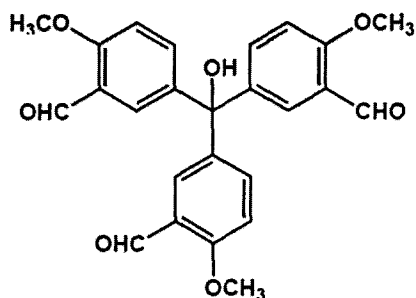
5,5',5''-(hydroxymethanetriyl)tris(2-methoxyphenol) ,

TRD7



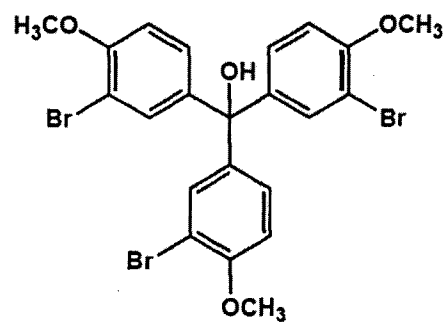
tris(3-fluoro-4-methoxyphenyl)methanol ,

TRD8



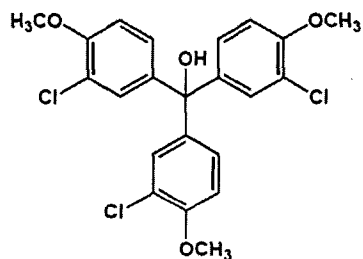
5,5',5''-(hydroxymethanetriyl)tris(2-methoxybenzaldehyde) ,

TRD9



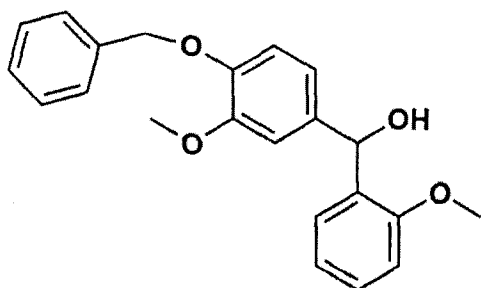
tris(3-bromo-4-methoxyphenyl)methanol ,

TRD10

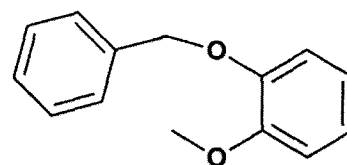


tris(3-chloro-4-methoxyphenyl)methanol ,

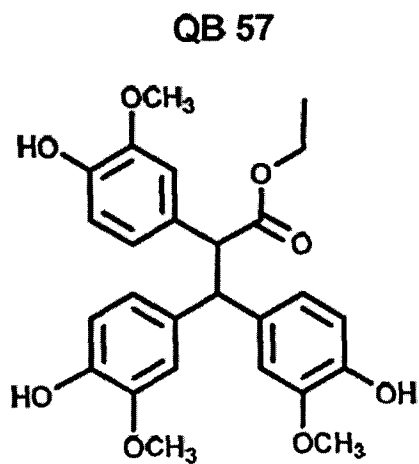
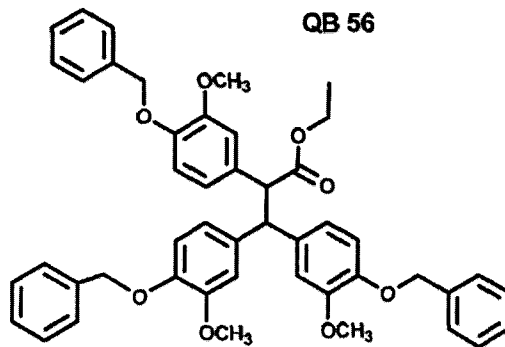
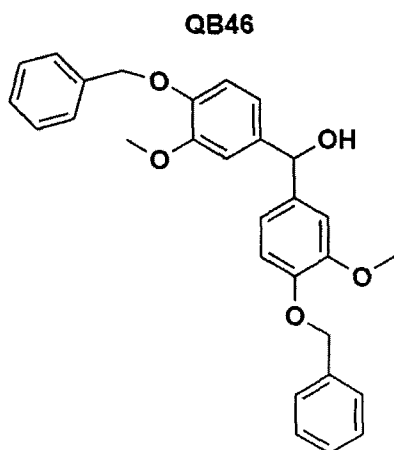
QB12



QB39



File No. P1777PC00

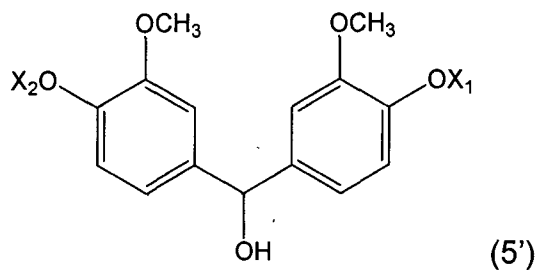


and

13. The use as claimed in claim 12, wherein said tumor is a breast tumor, a prostate tumor, a lung tumor, a colon tumor, a liver tumor and a testes tumor.

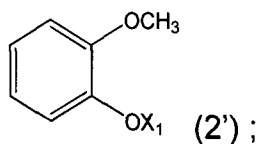
File No. P1777PC00

14. A process for the synthesis of a compound of formula (5'):

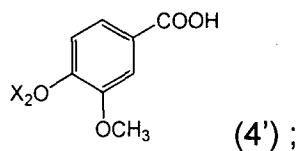


said process comprising the step of :

i. reacting a compound of formula (2')



with a compound of formula (4')

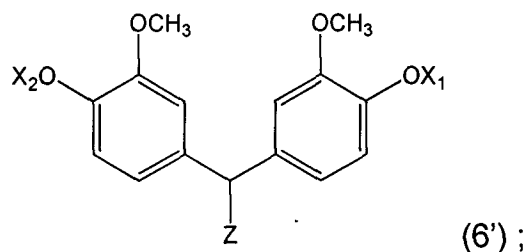


to obtain said compound of formula (5')

wherein X₁, and X₂ is a suitable protecting group for a hydroxyl group.

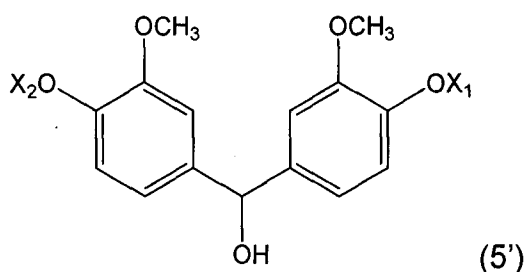
File No. P1777PC00

15. A process for the synthesis of a compound of formula (6')



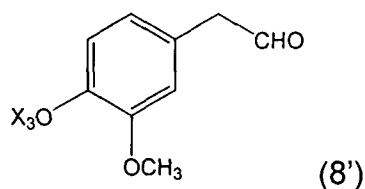
said process comprising the step of :

i. reacting a compound of formula (5')



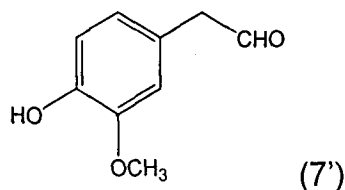
with a halogenating agent to obtain said compound of formula (6'), wherein X_1 , and X_2 is a suitable protecting group for a hydroxyl group, and wherein Z is a halogen atom.

16. A process for the synthesis of a compound of formula (8')



said process comprising the step of :

i. reacting a compound of formula (7')



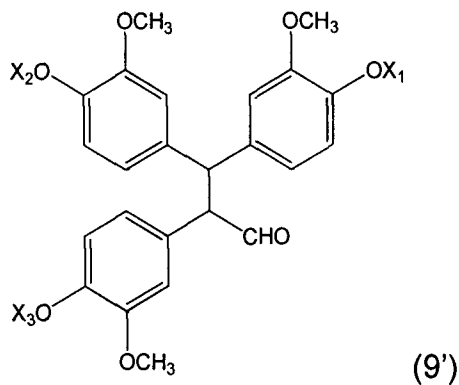
with a suitable hydroxyl protecting group, to obtain said compound of formula (8')

wherein X_3 is a suitable protecting group for a hydroxyl group; and

File No. P1777PC00

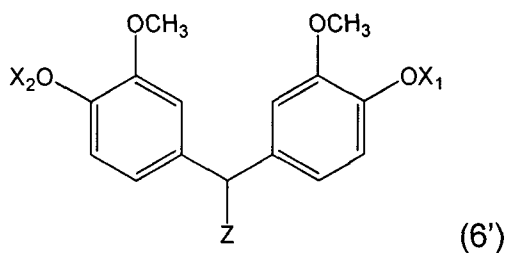
wherein Z is a halogen atom.

17. A process for the synthesis of a compound of formula (9')

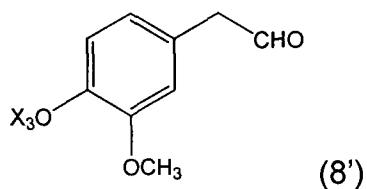


said process comprising the step of :

i. reacting a compound of formula (6')



with a compound of formula (8')

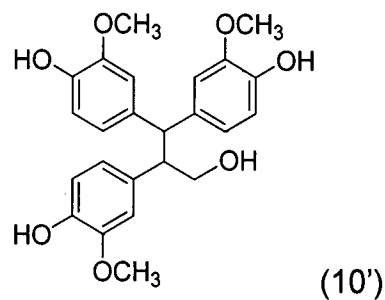


to obtain said compound of formula (9'),

wherein X_1 , X_2 and X_3 is a suitable protecting group for a hydroxyl group; and
 wherein Z is a halogen atom.

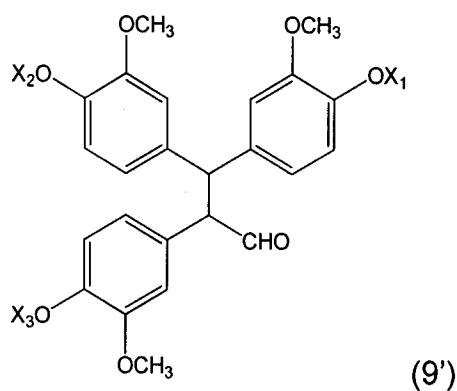
File No. P1777PC00

18. A process for the synthesis of a compound of formula (10') (Quebecol)



said process comprising the steps of:

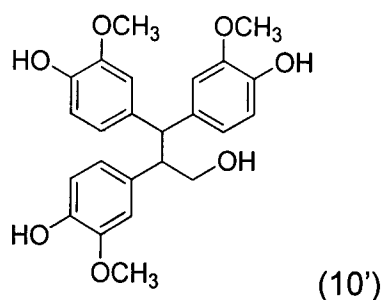
- i. reducing and deprotecting a compound of formula (9')



to obtain said compound of formula (10') (Quebecol);

wherein X₁, X₂ and X₃ is a suitable protecting group for a hydroxyl group; and
 wherein Z is a halogen atom.

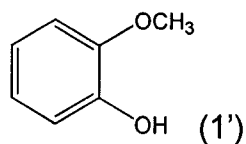
19. A process for the synthesis of a compound of formula (10') (Quebecol)



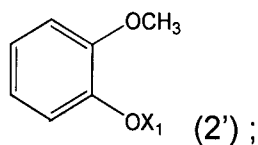
said process comprising the steps of:

- i. reacting a compound of formula (1')

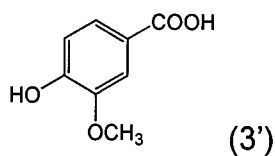
File No. P1777PC00



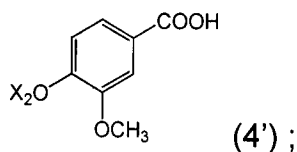
with a suitable hydroxyl protecting group, to obtain a compound of formula (2)



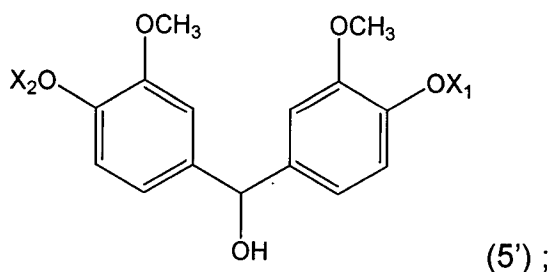
ii. reacting a compound of formula (3')



with a suitable hydroxyl protecting group, to obtain a compound of formula (4')

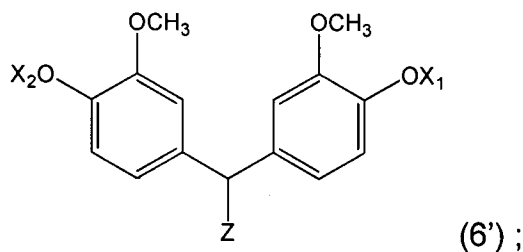


iii. reacting said compound of formula (2') with said compound of formula (4') to obtain a compound of formula (5')

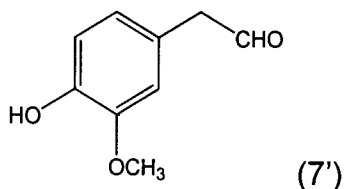


iv. reacting said compound of formula (5') with an halogenating agent to obtain a compound of formula (6')

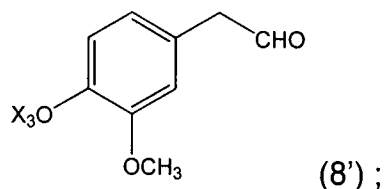
File No. P1777PC00



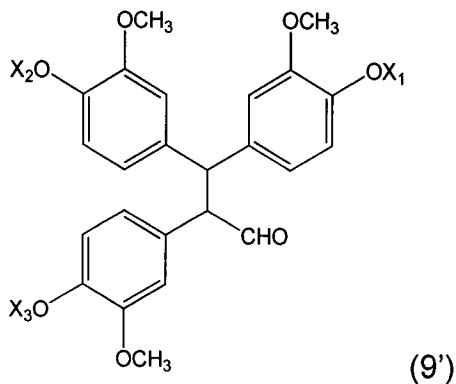
v. reacting a compound of formula (7')



with a suitable hydroxyl protecting group, to obtain a compound of formula (8)



vi. reacting said compound of formula (6') with said compound of formula (8') to obtain a compound of formula (9')



vii. reducing and deprotecting said compound of formula (9') to obtain a compound of formula (10') (Quebecol);
 wherein X₁, X₂ and X₃ is a suitable protecting group for a hydroxyl group; and
 wherein Z is a halogen atom.

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20. The process according to any one of claims 16 to 19, wherein said X_3 is chosen from Fluorenylmethyloxycarbonyl chloride (Fmoc), Triphenylmethyl chloride, and a silyl ether.

21. The process according to any one of claims 16 to 19, wherein said X_3 is a silyl ether.

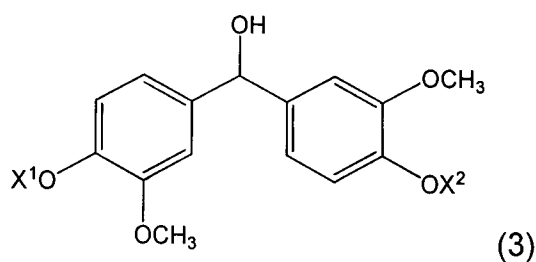
22. The process according to any one of claims 18 - 19, wherein said reducing is by reacting said compound (9') with NaBH_4 .

23. The process according to any one of claims 18 - 19, wherein said deprotection is by reacting said compound of formula (9') with one of tetra-*n*-butylammonium fluoride (TBAF) or trifluoroacetic acid (TFA).

24. The process according to claim 19, wherein in step iv), said halogenating agent is a trihalide of phosphorus.

25. The process according to claim 19, wherein said trihalide of phosphorus is chosen from PBr_3 , and PCl_3 .

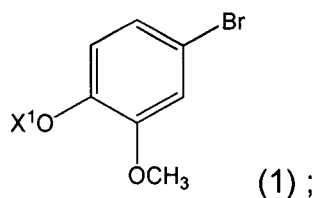
26. A process for the synthesis of a compound of formula (3)



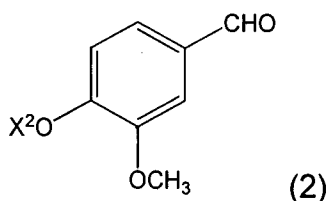
said process comprising the step of:

- i. reacting a compound of formula (1)

File No. P1777PC00



with a compound of formula (2)

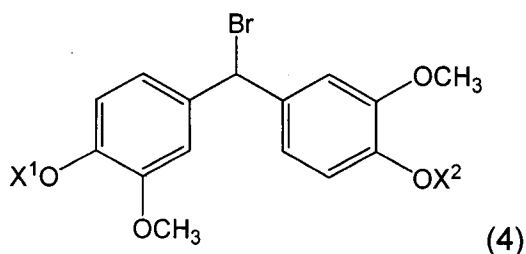


in presence of a strong base, to obtain a compound of formula (3)
wherein X¹ and X² is a suitable protecting group for a hydroxyl group.

27. The process of claim 26, wherein said strong base is n-butyllithium (n-BuLi).

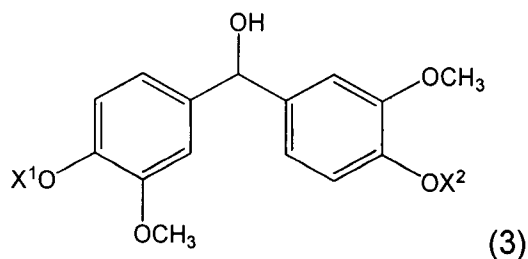
28. The process of claim 27, wherein reacting is in tetrahydrofuran (THF) at -78°C.

29. A process for the synthesis of a compound of formula (4)



said process comprising the step of:

i. brominating a compound of formula (3)



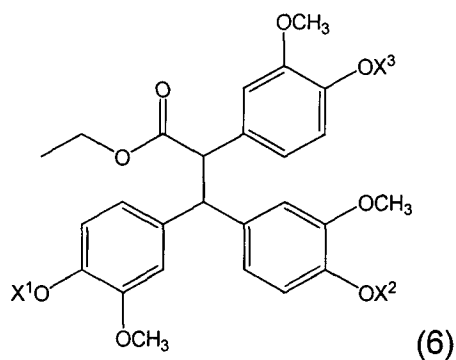
File No. P1777PC00

to obtain said compound of formula (4)
 wherein X^1 and X^2 is a suitable protecting group for a hydroxyl group.

30. The process of claim 29, wherein brominating is with acetyl bromide (CH_3COBr).

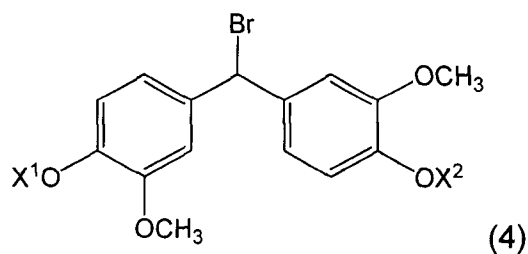
31. The process of claim 29, wherein brominating is with acetyl bromide (CH_3COBr) in benzene.

32. A process for the synthesis of a compound of formula (6)

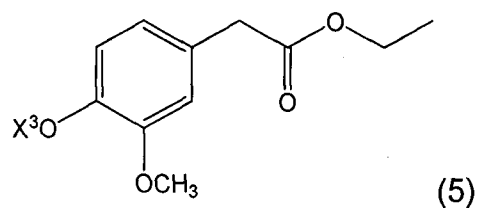


said process comprising the step of:

i. reacting a compound of formula (4)



with a compound of formula (5)



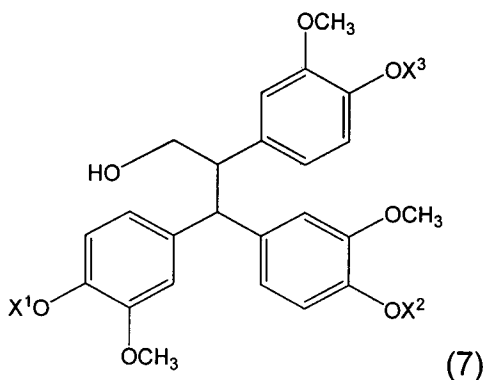
File No. P1777PC00

in the presence of a strong base, to obtain said compound of formula (6),
 wherein X^1 , X^2 , and X^3 is a suitable protecting group for a hydroxyl group.

33. The process of claim 32, wherein said strong base is Lithium diisopropylamide (LDA).

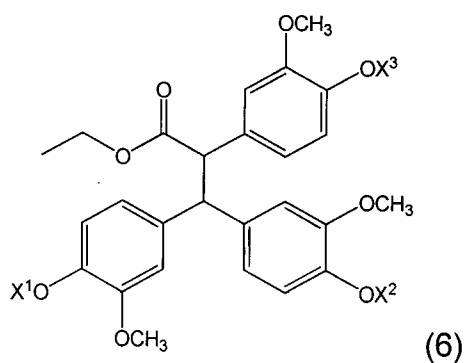
34. The process of claim 32, wherein said strong base is Lithium diisopropylamide (LDA) in tetrahydrofuran at tetrahydrofuran (THF) at -78°C .

35. A process for the synthesis of a compound of formula (7)



said process comprising the step of:

- i. reducing a compound of formula (6)



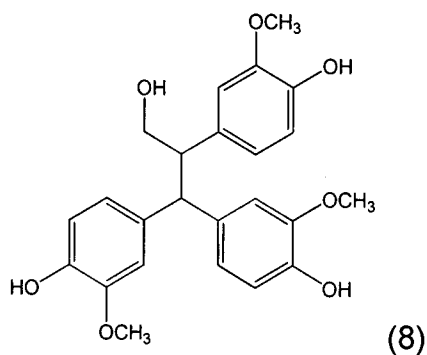
to obtain said compound of formula (7),
 wherein X^1 , X^2 , and X^3 is a suitable protecting group for a hydroxyl group.

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36. The process of claim 35, wherein reducing is with lithium aluminum hydride (LiAlH₄).

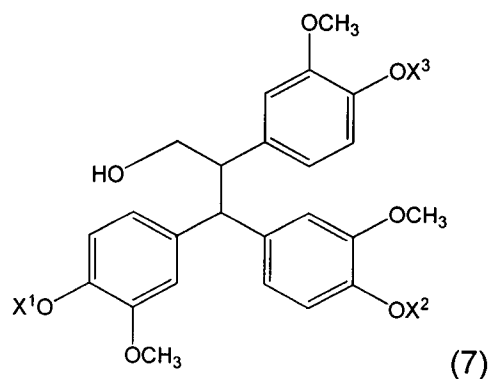
37. The process of claim 35, wherein reducing is with lithium aluminum hydride (LiAlH₄) in tetrahydrofuran (THF).

38. A process for the synthesis of a compound of formula (8) (Quebecol)



said process comprising the step of:

- i. deprotecting a compound of formula (7)



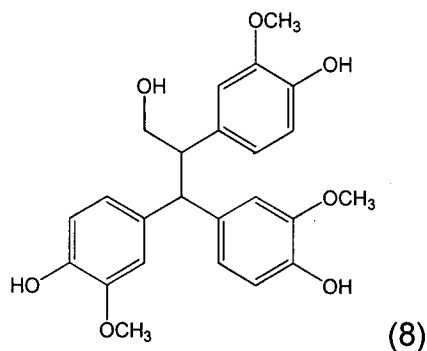
to obtain said compound of formula (8),
wherein X¹, X², and X³ is a suitable protecting group for a hydroxyl group.

39. The process of claim 38, wherein deprotecting is with ammonium formate (HCO₂NH₄) and palladium on carbon (Pd/C).

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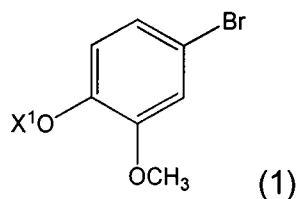
40. The process of claim 38, wherein deprotecting is with ammonium formate (HCO_2NH_4) and palladium on carbon (Pd/C) in methanol (MeOH).

41. A process for the synthesis of a compound of formula (8) (Quebecol)

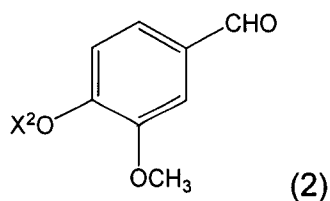


said process comprising the step of:

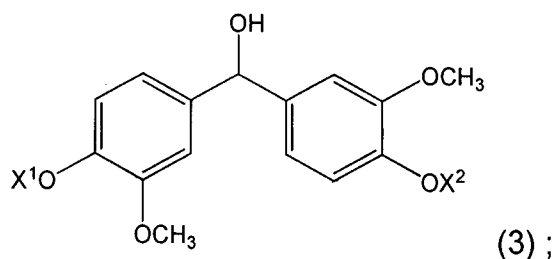
i. reacting a compound of formula (1)



with a compound of formula (2)

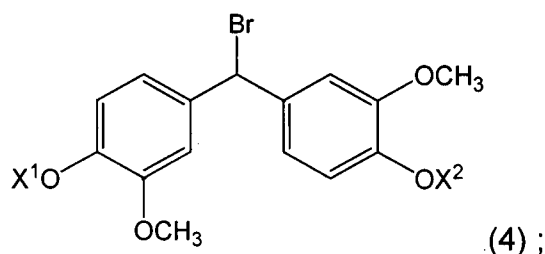


in presence of a strong base, to obtain a compound of formula (3)

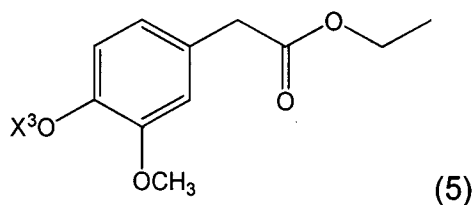


ii. brominating said compound of formula (3), to obtain a compound of formula (4)

File No. P1777PC00

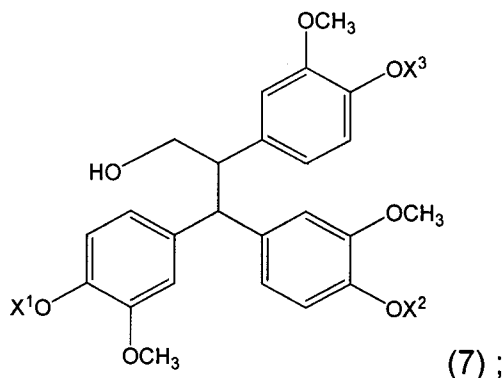


- iii. reacting said compound of formula (4) with a compound of formula (5)



in the presence of a strong base, to obtain a compound of formula (6) ;

- iv. reducing said compound of formula (6) to obtain a compound of formula (7)



- v. deprotecting said compound of formula (7) to obtain said compound of formula (8) (Quebecol),

wherein X^1 , X^2 , and X^3 is a suitable protecting group for a hydroxyl group.

42. The process of any one of claims 14 to 41, wherein said suitable protecting group for a hydroxyl group is chosen from C_1 - C_{25} ethers, C_1 - C_{25} substituted methyl ethers, C_1 - C_{25} substituted ethyl ethers, C_1 - C_{25} acyl groups, C_1 - C_{25} halogenated acyl groups, C_1 - C_{25} substituted benzyl ethers, C_1 - C_{25} silyl ethers, C_1 - C_{25} esters, C_1 - C_{25} carbonates, and C_1 - C_{25} sulfonates.

File No. P1777PC00

43. The process of any one of claims 14 to 42, wherein said suitable protecting group for a hydroxyl group is chosen from diphenylmethylchlorosilane (DPMS), Tosyl, methyl, methoxymethyl, benzyloxymethyl, tetrahydropyranyl, tetrahydrofuranyl, 2-(trimethylsilyl)ethoxymethyl, dioxanyl, 1-ethoxyethyl, 1-(2-chloroethoxy)ethyl, 2,2,2-trichloroethyl, t-butyl, allyl, propargyl, benzyl, p-methoxybenzyl, diphenylmethyl, triphenylmethyl, trimethylsilyl, triethylsilyl, triisopropylsilyl, dimethylisopropylsilyl, diethylisopropylsilyl, dimethylhexylsilyl, t-butyldimethylsilyl, t-butyldiphenylsilyl, tribenzylsilyl, triphenylsilyl, triisopropylsilyl, diphenylmethylsilyl, benzylformate, methylcarbonyl, ethylcarbonyl, methoxymethyl carbonyl, trichloroethoxycarbonyl, benzylcarbonyl, benzyloxycarbonyl, allylsulfonyl, methanesulfonyl, and p-toluenesulfonyl.

44. The process of any one of claims 14 to 42, wherein said suitable protecting group for a hydroxyl group is benzyl (Bn).

45. The process of claim 41, wherein in step i), strong base is n-butyllithium (n-BuLi).

46. The process of claim 45, wherein reacting is in tetrahydrofuran (THF) at -78°C.

47. The process of claim 41, wherein in step ii) brominating is with acetyl bromide (CH₃COBr).

48. The process of claim 41, wherein in step ii) brominating is with acetyl bromide (CH₃COBr) in benzene.

File No. P1777PC00

49. The process of claim 41, wherein in step iii) said strong base is Lithium diisopropylamide (LDA).

50. The process of claim 41, wherein in step iii) said strong base is Lithium diisopropylamide (LDA) in tetrahydrofuran at tetrahydrofuran (THF) at -78°C .

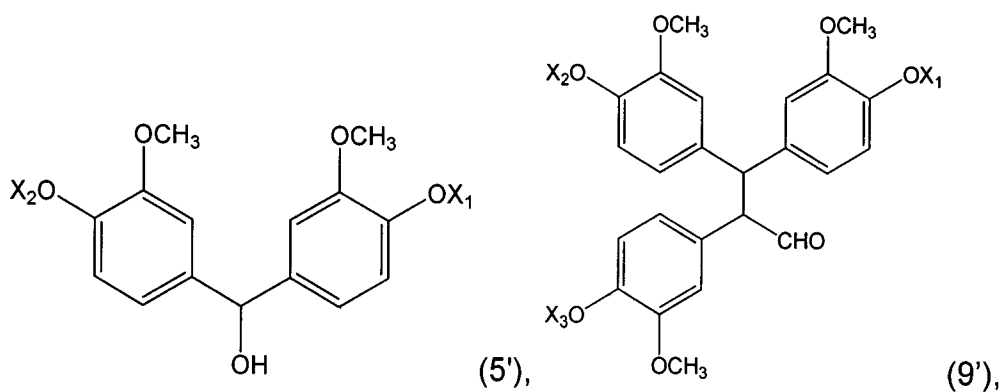
51. The process of claim 41, wherein in step iv) reducing is with lithium aluminum hydride (LiAlH_4).

52. The process of claim 41, wherein in step iv) reducing is with lithium aluminum hydride (LiAlH_4) in tetrahydrofuran (THF).

53. The process of claim 41, wherein in step v) deprotecting is with ammonium formate (HCO_2NH_4) and palladium on carbon (Pd/C).

54. The process of claim 41, wherein deprotecting is with ammonium formate (HCO_2NH_4) and palladium on carbon (Pd/C) in methanol (MeOH).

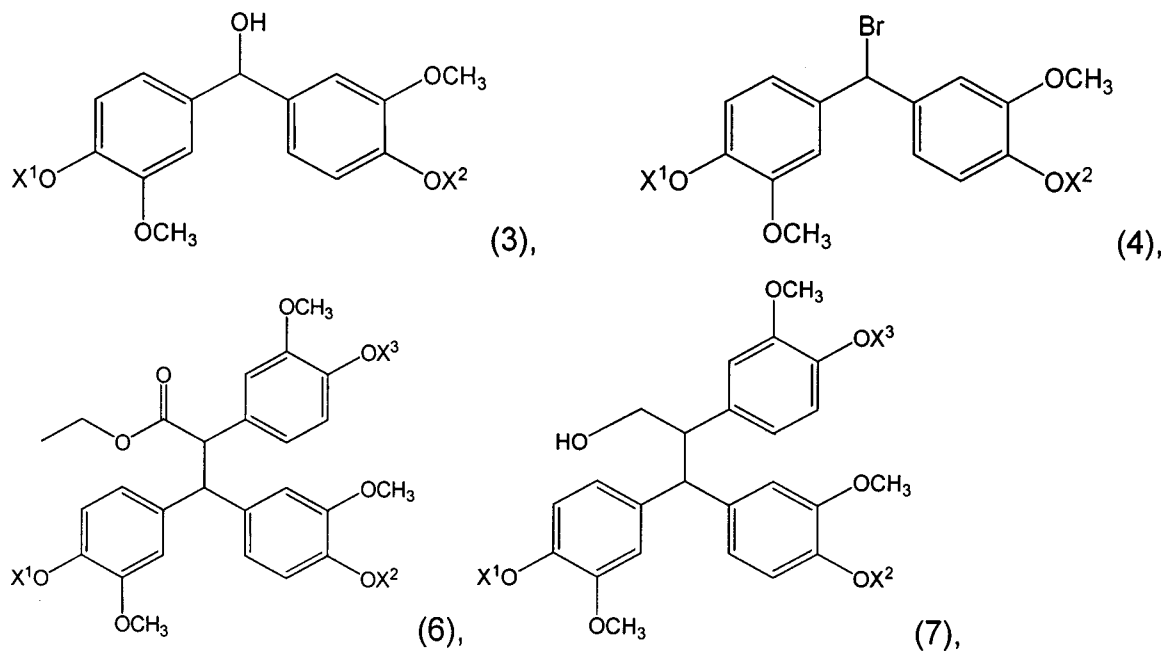
55. A compound of formula (5') and (9'):



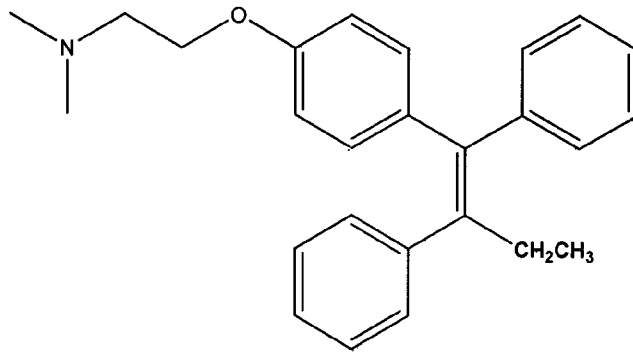
File No. P1777PC00

wherein X₁, X₂ and X₃ are as defined above.

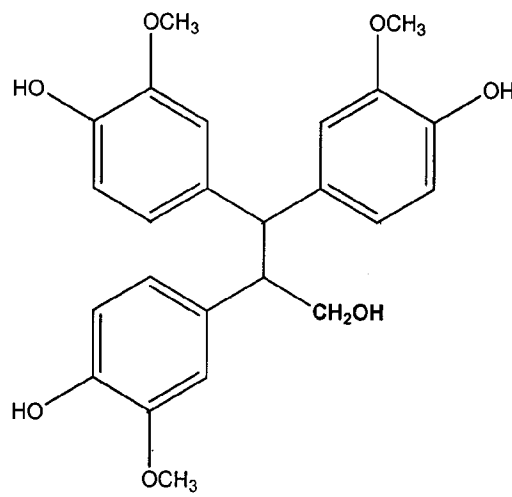
56. A compound of formula (3), (4), (6) and (7):



wherein X¹, X² and X³ are as defined above.



Tamoxifen



Quebecol

Fig. 1

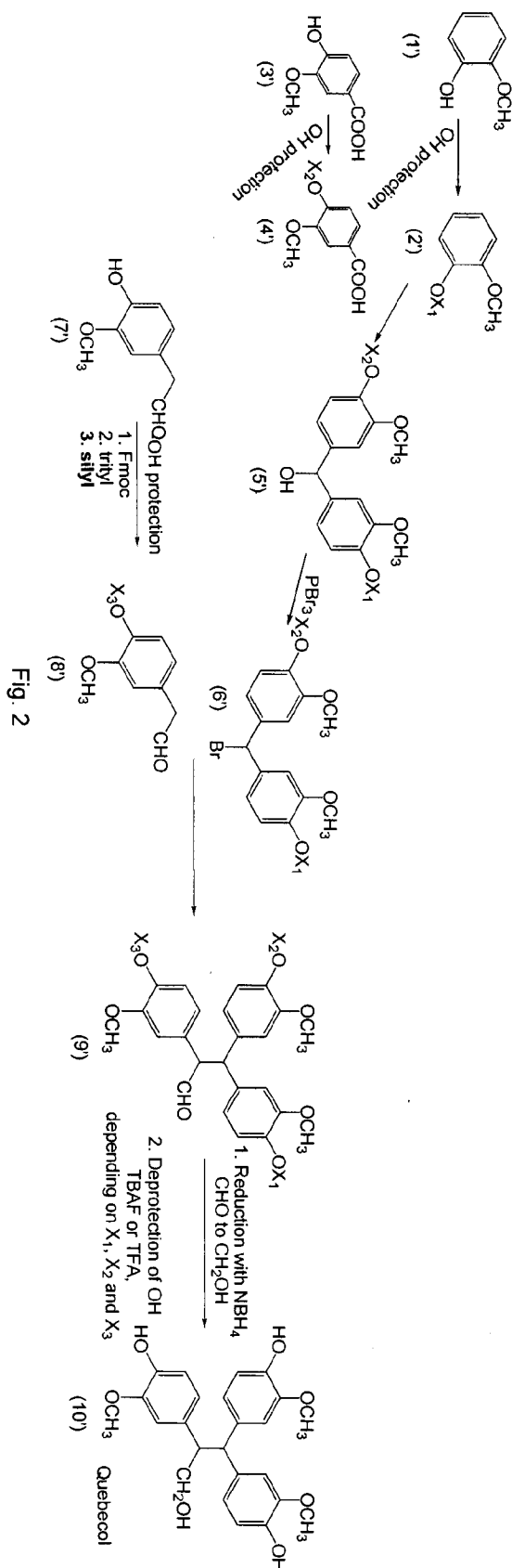


Fig. 2

3/13

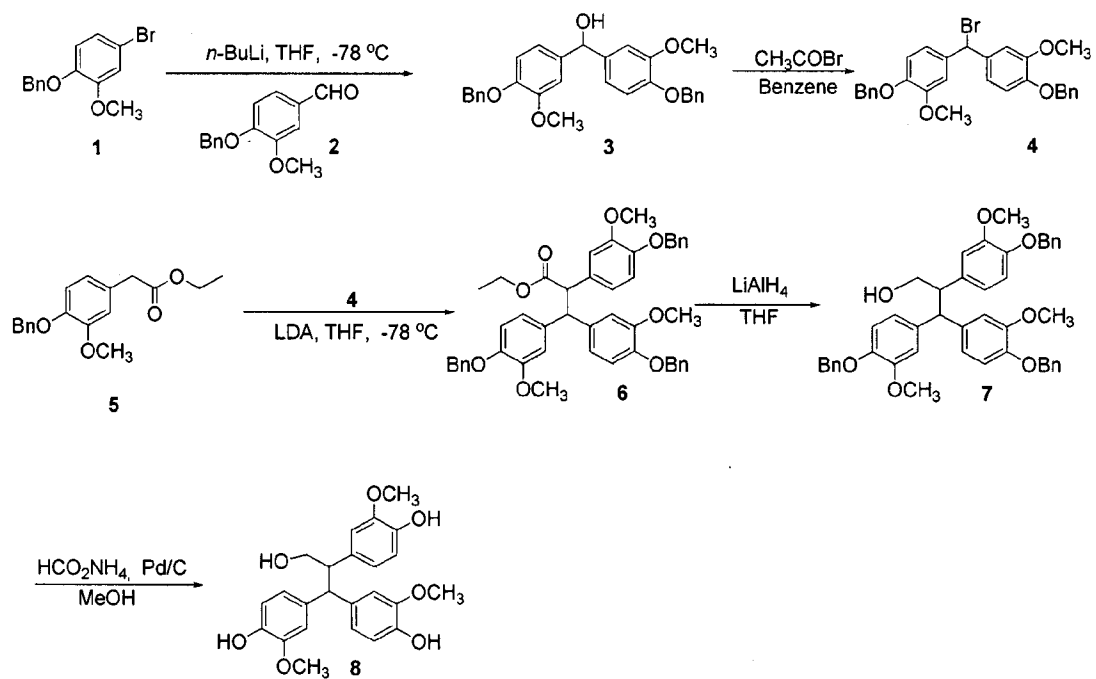


Fig. 3

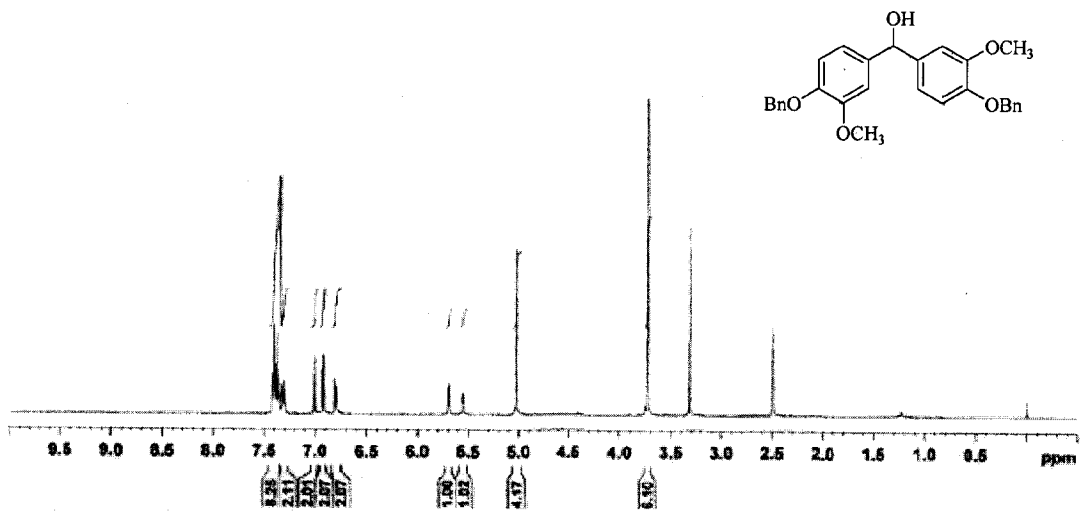


Fig. 4A

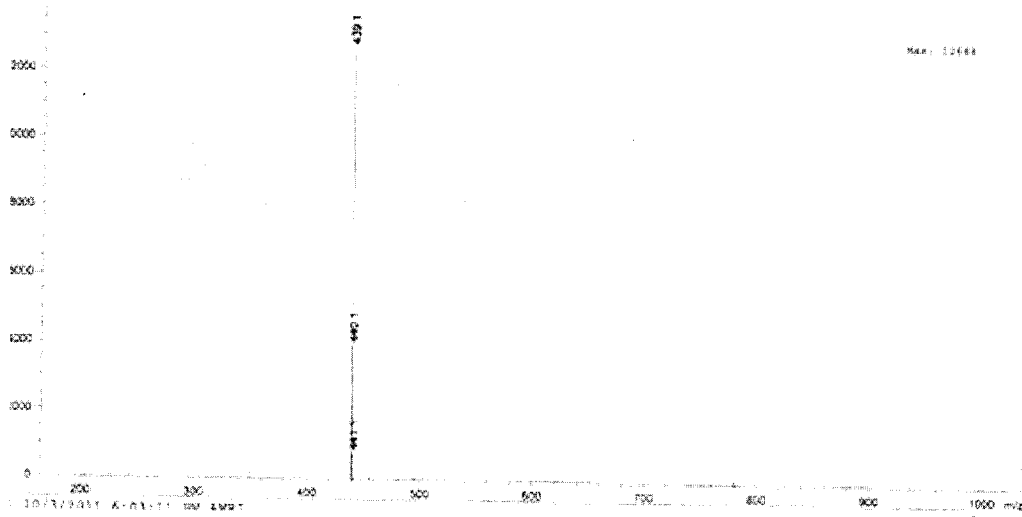


Fig. 4B

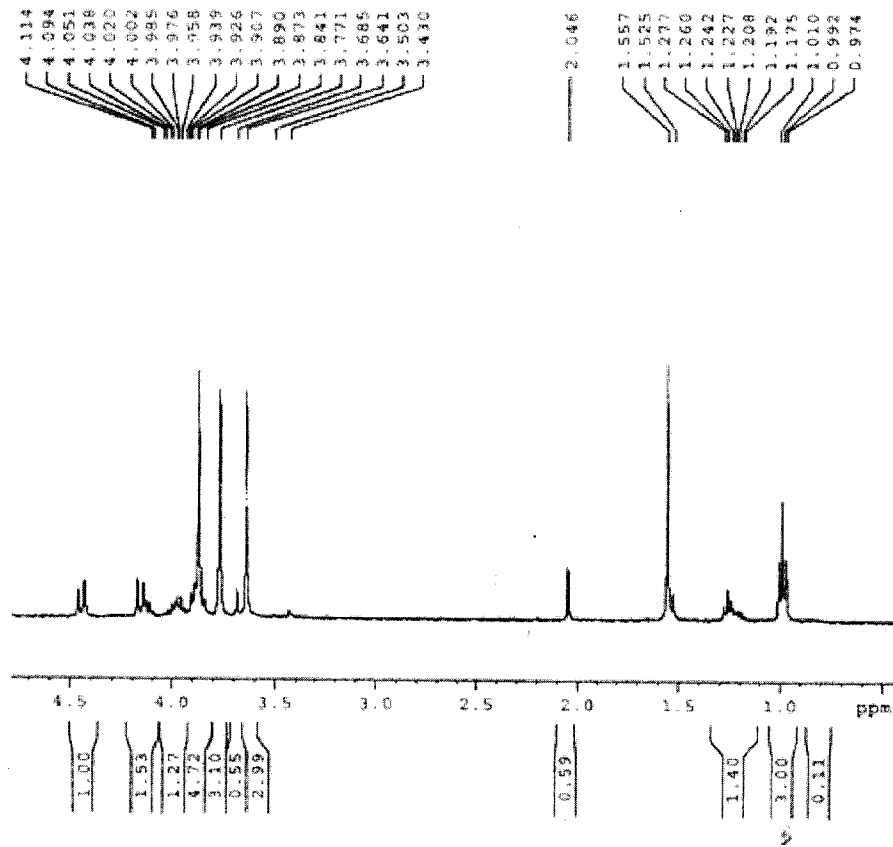


Fig. 5C

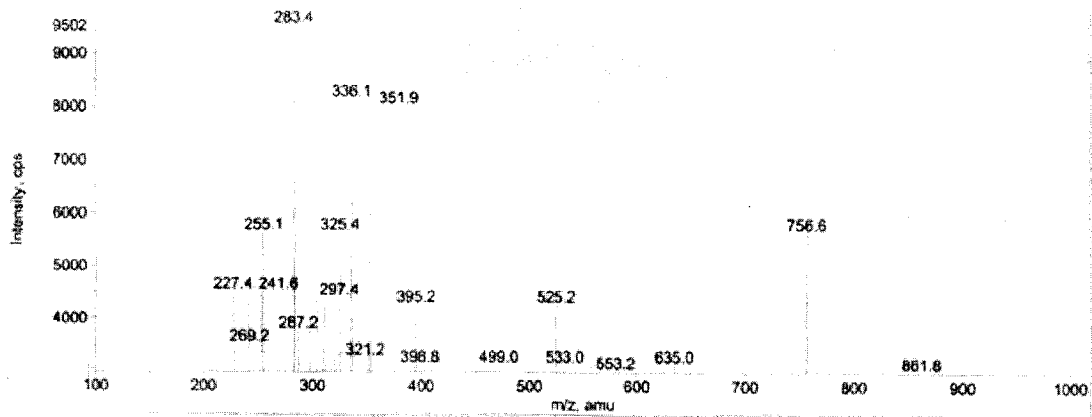


Fig. 5D

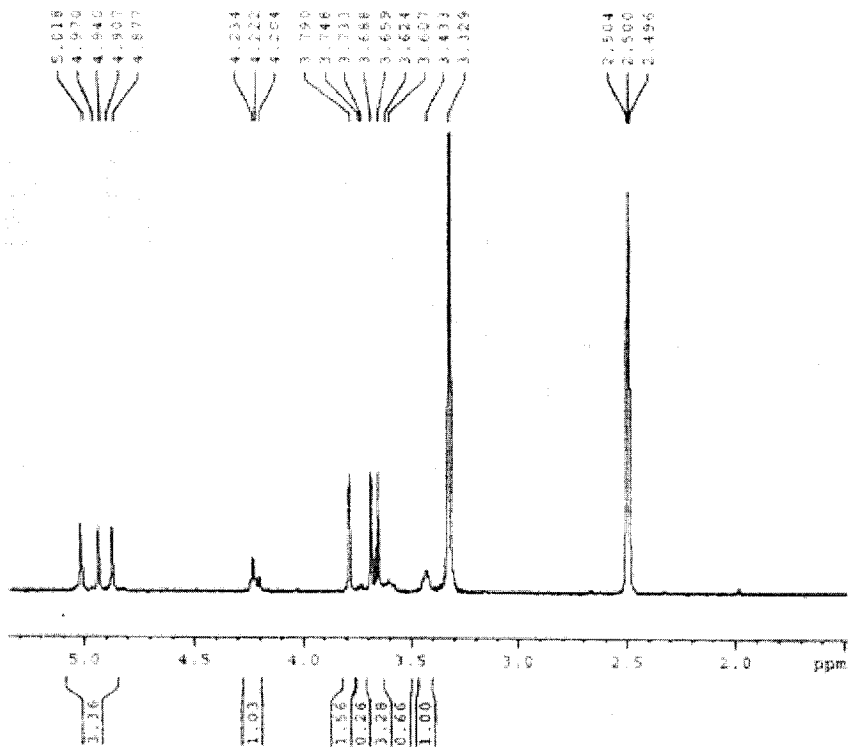


Fig. 6C

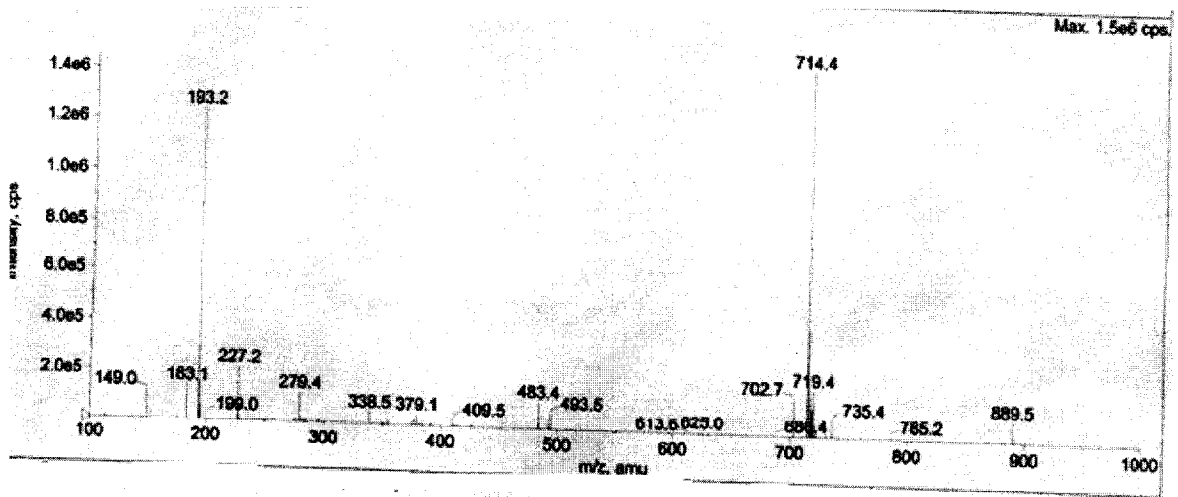


Fig. 6D

PQ-46

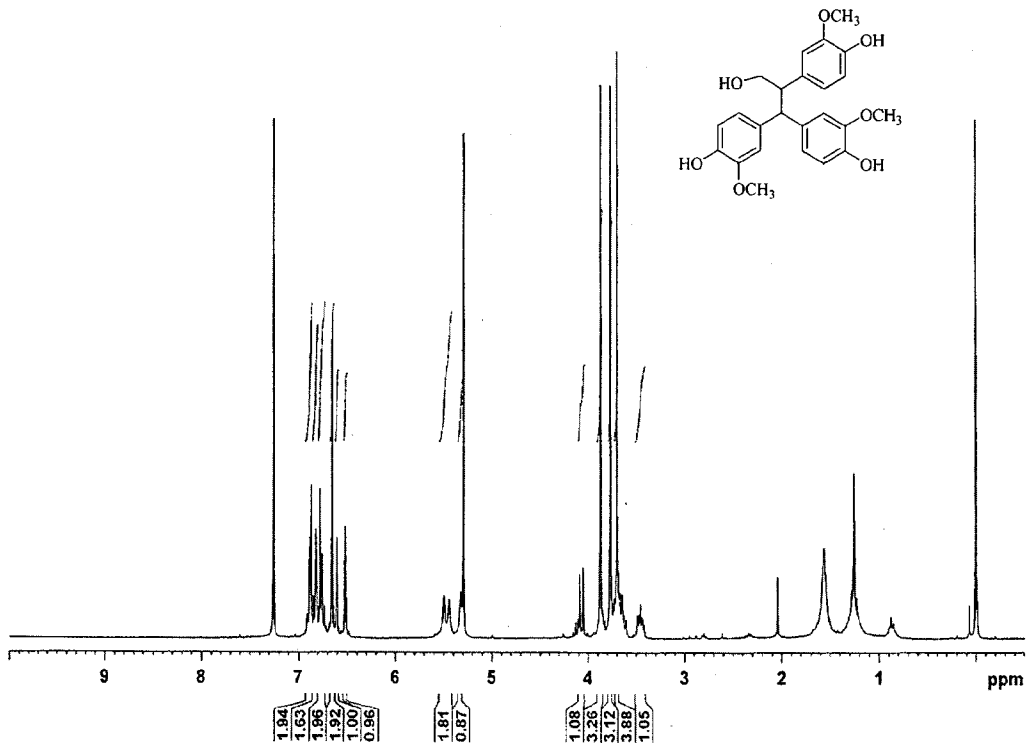


Fig. 7A

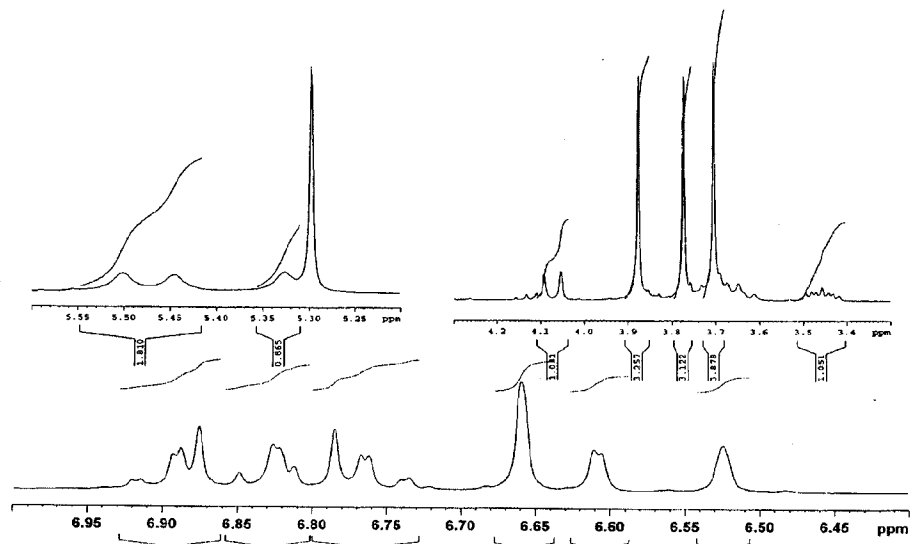


Fig. 7B

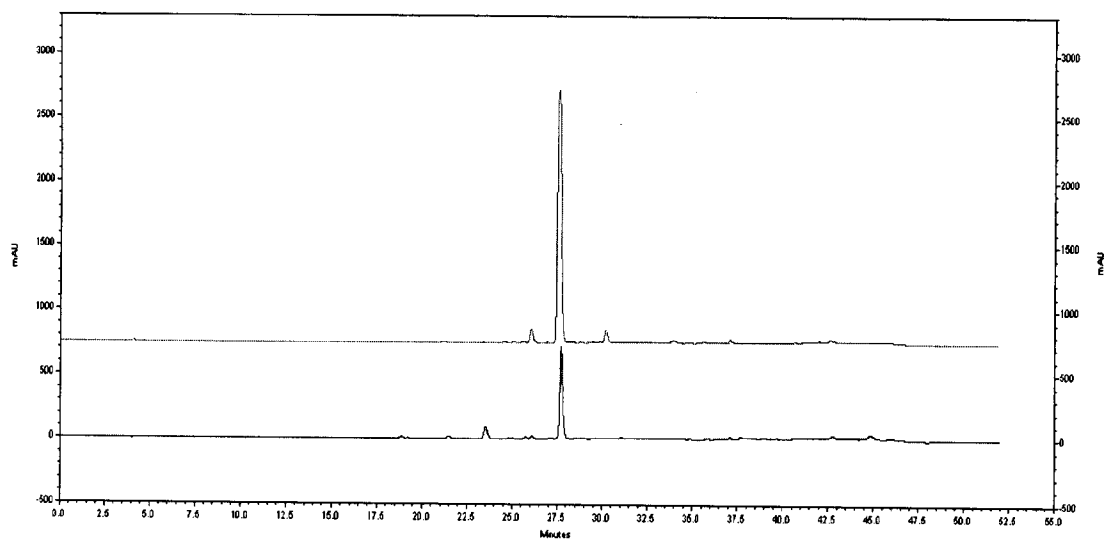
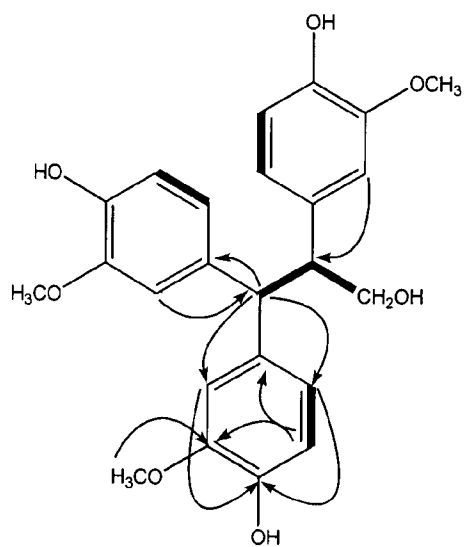


Fig. 7C

A



B

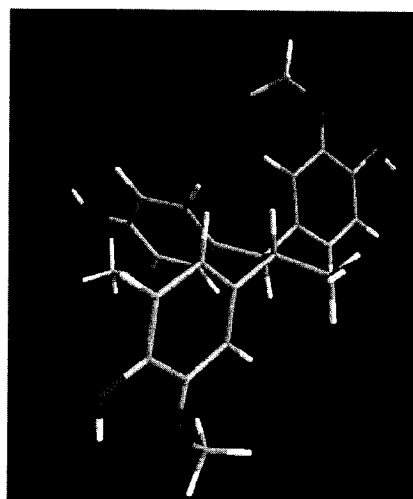
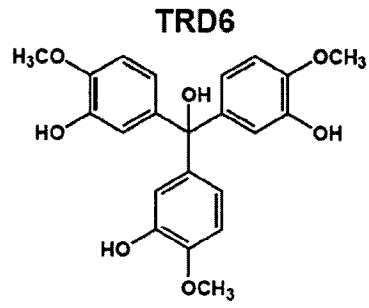
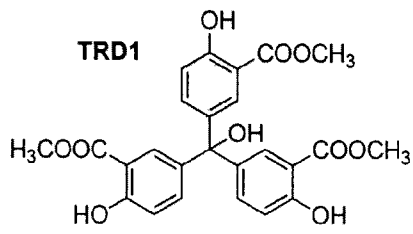
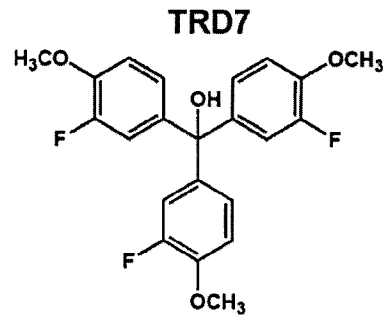
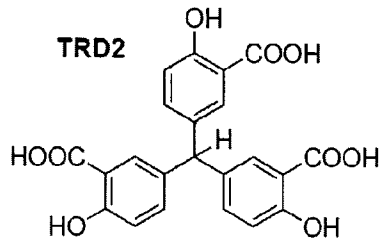


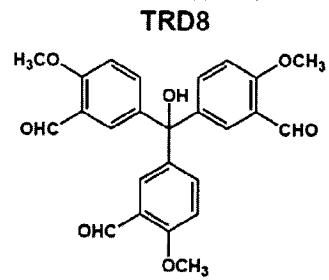
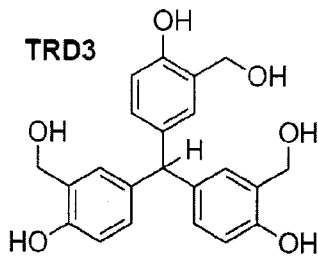
Fig. 8



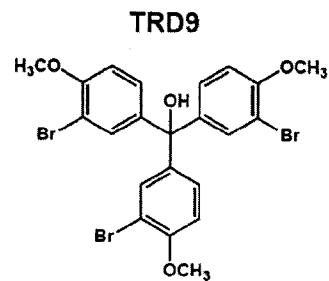
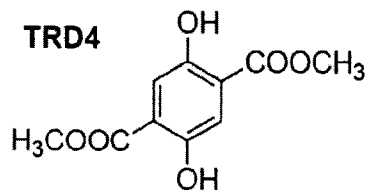
5,5'-(hydroxymethanetriyl)tris(2-methoxyphenol)



tris(3-fluoro-4-methoxyphenyl)methanol

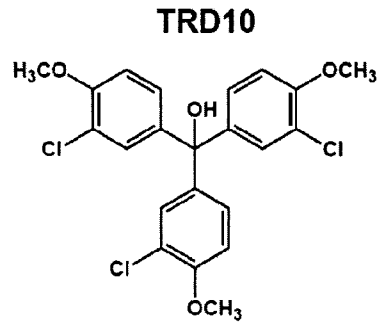
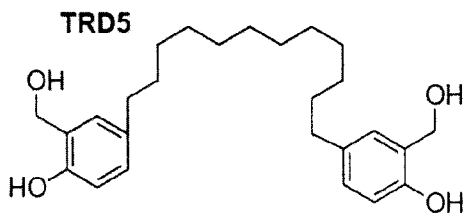


5,5'-(hydroxymethanetriyl)tris(2-methoxybenzaldehyde)



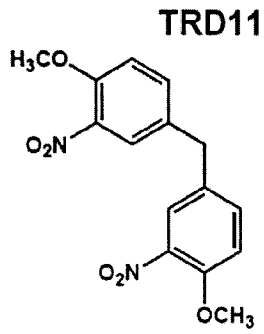
tris(3-bromo-4-methoxyphenyl)methanol

Fig. 9A



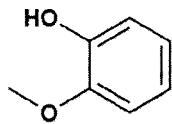
tris(3-chloro-4-methoxyphenyl)methanol

ng

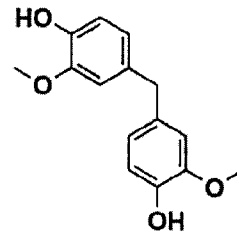


bis(4-methoxy-3-nitrophenyl)methane

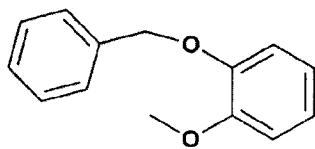
QB0



QB48



QB39



QB49

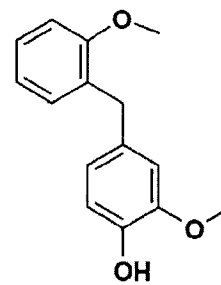


Fig 9B

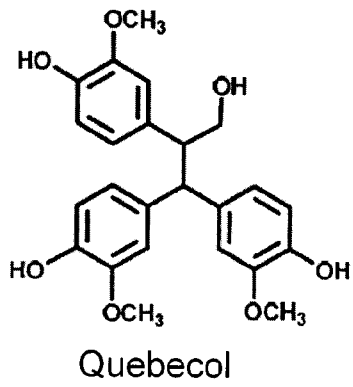
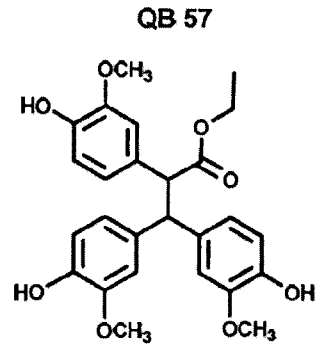
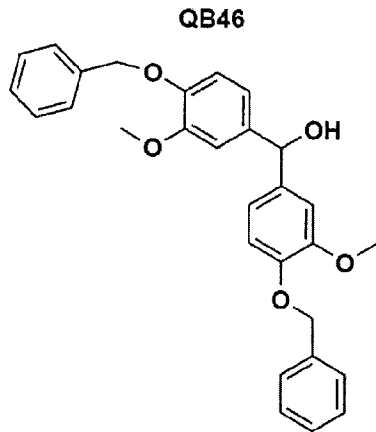
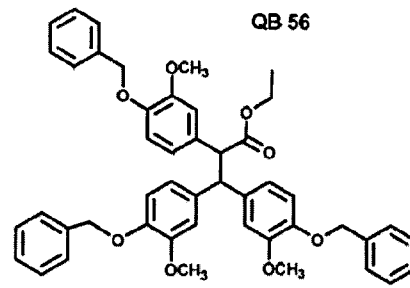
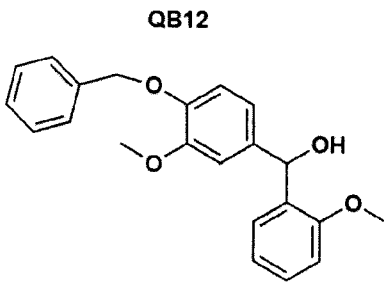


Fig. 9C

INTERNATIONAL SEARCH REPORT

International application No.
PCT/CA2012/000556

<p>A. CLASSIFICATION OF SUBJECT MATTER IPC: <i>C07C 43/23</i> (2006.01) , <i>A61K 31/05</i> (2006.01) , <i>A61K 31/085</i> (2006.01) , <i>A61K 31/216</i> (2006.01) , <i>A61P 35/00</i> (2006.01) , <i>C07C 41/18</i> (2006.01) (more IPCs on the last page) According to International Patent Classification (IPC) or to both national classification and IPC</p>																
<p>B. FIELDS SEARCHED</p> <p>Minimum documentation searched (classification system followed by classification symbols) <i>C07C 43/23</i> (2006.01) , <i>A61K 31/05</i> (2006.01) , <i>A61K 31/085</i> (2006.01) , <i>A61K 31/216</i> (2006.01) , <i>A61P 35/00</i> (2006.01) , <i>C07C 41/18</i> (2006.01),<i>C07C 45/61</i> (2006.01) , <i>C07C 47/575</i> (2006.01) , <i>C07C 67/30</i> (2006.01) , <i>C07C 69/734</i> (2006.01)</p> <p>Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched</p> <p>Electronic database(s) consulted during the international search (name of database(s) and, where practicable, search terms used) Canadian Patent Database, STN, Total Patent, PubMed, Scopus (maple, syrup, sugar maple, quebecol, red maple, cancer, antioxidant, phenolic, polyphenolic)</p>																
<p>C. DOCUMENTS CONSIDERED TO BE RELEVANT</p> <table border="1"> <thead> <tr> <th>Category*</th> <th>Citation of document, with indication, where appropriate, of the relevant passages</th> <th>Relevant to claim No.</th> </tr> </thead> <tbody> <tr> <td>X Y</td> <td>US 4,412,057 (KANEGAFUCHI KAGAKU KOGYO KABUSHIKI KAISHA JP) 25 October 1983 (25-10-1983) column 3, lines 55-56</td> <td>6 7</td> </tr> <tr> <td>X Y</td> <td>CS 91870 (CTVRTNIK J., MAYER J., CS) 15 September 1959 (15-09-1959) page 2, lines 40-46</td> <td>15, 29, 55, 56 30, 31</td> </tr> <tr> <td>X Y</td> <td>GLASS "Cooperative Chemical Sensing with Bis-tritylacetylenes: Pinwheel Receptors with Metal Ion Recognition Properties" J.Am. Chem. Soc. 2000, vol. 122, pages 4522-4523 S-1, compound 5</td> <td>1 7</td> </tr> <tr> <td>X Y</td> <td>SMITH ET AL. "The Basicity of Hydroxyde Ion at 170°C" IPC Technical Paper Series 1982, Number 131, pages 1-16 page 10, compound 25</td> <td>6 7</td> </tr> </tbody> </table>		Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.	X Y	US 4,412,057 (KANEGAFUCHI KAGAKU KOGYO KABUSHIKI KAISHA JP) 25 October 1983 (25-10-1983) column 3, lines 55-56	6 7	X Y	CS 91870 (CTVRTNIK J., MAYER J., CS) 15 September 1959 (15-09-1959) page 2, lines 40-46	15, 29, 55, 56 30, 31	X Y	GLASS "Cooperative Chemical Sensing with Bis-tritylacetylenes: Pinwheel Receptors with Metal Ion Recognition Properties" J.Am. Chem. Soc. 2000, vol. 122, pages 4522-4523 S-1, compound 5	1 7	X Y	SMITH ET AL. "The Basicity of Hydroxyde Ion at 170°C" IPC Technical Paper Series 1982, Number 131, pages 1-16 page 10, compound 25	6 7
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.														
X Y	US 4,412,057 (KANEGAFUCHI KAGAKU KOGYO KABUSHIKI KAISHA JP) 25 October 1983 (25-10-1983) column 3, lines 55-56	6 7														
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X Y	SMITH ET AL. "The Basicity of Hydroxyde Ion at 170°C" IPC Technical Paper Series 1982, Number 131, pages 1-16 page 10, compound 25	6 7														
<p><input checked="" type="checkbox"/> Further documents are listed in the continuation of Box C. <input checked="" type="checkbox"/> See patent family annex.</p> <table border="1"> <tbody> <tr> <td>* Special categories of cited documents :</td> <td>"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention</td> </tr> <tr> <td>"A" document defining the general state of the art which is not considered to be of particular relevance</td> <td>"X" document of particular relevance: the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone</td> </tr> <tr> <td>"E" earlier application or patent but published on or after the international filing date</td> <td>"Y" document of particular relevance: the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art</td> </tr> <tr> <td>"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</td> <td>"&" document member of the same patent family</td> </tr> <tr> <td>"O" document referring to an oral disclosure, use, exhibition or other means</td> <td></td> </tr> <tr> <td>"P" document published prior to the international filing date but later than the priority date claimed</td> <td></td> </tr> </tbody> </table>		* Special categories of cited documents :	"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention	"A" document defining the general state of the art which is not considered to be of particular relevance	"X" document of particular relevance: the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone	"E" earlier application or patent but published on or after the international filing date	"Y" document of particular relevance: the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art	"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)	"&" document member of the same patent family	"O" document referring to an oral disclosure, use, exhibition or other means		"P" document published prior to the international filing date but later than the priority date claimed				
* Special categories of cited documents :	"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention															
"A" document defining the general state of the art which is not considered to be of particular relevance	"X" document of particular relevance: the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone															
"E" earlier application or patent but published on or after the international filing date	"Y" document of particular relevance: the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art															
"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)	"&" document member of the same patent family															
"O" document referring to an oral disclosure, use, exhibition or other means																
"P" document published prior to the international filing date but later than the priority date claimed																
<p>Date of the actual completion of the international search</p> <p>13 August 2012 (13-08-2012)</p>	<p>Date of mailing of the international search report</p> <p>01 October 2012 (01-10-2012)</p>															
<p>Name and mailing address of the ISA/CA</p> <p>Canadian Intellectual Property Office Place du Portage I, C114 - 1st Floor, Box PCT 50 Victoria Street Gatineau, Quebec K1A 0C9 Facsimile No.: 001-819-953-2476</p>	<p>Authorized officer</p> <p>Alessandra Mezzetti (819) 934-6736</p>															

INTERNATIONAL SEARCH REPORTInternational application No.
PCT/CA2012/000556**Box No. II Observations where certain claims were found unsearchable (Continuation of item 2 of the first sheet)**

This international search report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons :

1. Claim Nos. : 8-9
because they relate to subject matter not required to be searched by this Authority, namely :

Claims 8-9 are directed to a method for treatment of the human or animal body by surgery or therapy which the International Search Authority is not required to search. However, this Authority has carried out a search based on the alleged effects or purposes/uses of the product defined in claims 10-12.
2. Claim Nos. :
because they relate to parts of the international application that do not comply with the prescribed requirements to such an extent that no meaningful international search can be carried out, specifically :
3. Claim Nos. :
because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).

Box No. III Observations where unity of invention is lacking (Continuation of item 3 of first sheet)

This International Searching Authority found multiple inventions in this international application, as follows :

1. As all required additional search fees were timely paid by the applicant, this international search report covers all searchable claims.
2. As all searchable claims could be searched without effort justifying additional fees, this Authority did not invite payment of additional fees.
3. As only some of the required additional search fees were timely paid by the applicant, this international search report covers only those claims for which fees were paid, specifically claim Nos. :
4. No required additional search fees were timely paid by the applicant. Consequently, this international search report is restricted to the invention first mentioned in the claims; it is covered by claim Nos. :

Remark on Protest The additional search fees were accompanied by the applicant's protest and, where applicable, the payment of a protest fee.

The additional search fees were accompanied by the applicant's protest but the applicable protest fee was not paid within the time limit specified in the invitation.

No protest accompanied the payment of additional search fees.

INTERNATIONAL SEARCH REPORT

International application No.
PCT/CA2012/000556

C (Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	HARIG ET AL. "2,3,6,7,10,11-Hexamethoxytribenzotriquinacene: Synthesis, Solid-State Structure, and Functionalization of a Rigid Analogue of Cyclotriveratrylene" Eur. J. Org. Chem. 2004, pages 2381-2397 scheme 3	14, 55, 56
X	WOOD ET AL. "Chiral Aromatase and Dual Aromatase - Steroid Sulfatase Inhibitors from Letrozole Template: Synthesis, Absolute Configuration, and In Vitro Activity" J. Med. Chem. 2008, vol. 51, pages 4226-4238 scheme 3	55, 56

INTERNATIONAL SEARCH REPORT
Information on patent family members

International application No.
PCT/CA2012/000556

Patent Document Cited in Search Report	PublicationPatent Date	Family Member(s)	Publication Date
US4412057	25 October 1983 (25-10-1983)	None	
CS91870	15 September 1959 (15-09-1959)	None	

INTERNATIONAL SEARCH REPORT

International application No.
PCT/CA2012/000556

C07C 45/61 (2006.01) , *C07C 47/575* (2006.01) , *C07C 67/30* (2006.01) , *C07C 69/734* (2006.01)