

# One-pot tandem cyclization of enantiopure asymmetric *cis*-2,5-disubstituted pyrrolidines: Facile access to chiral 10-heteroazatriquinanes

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## Full Research Paper

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## Abstract

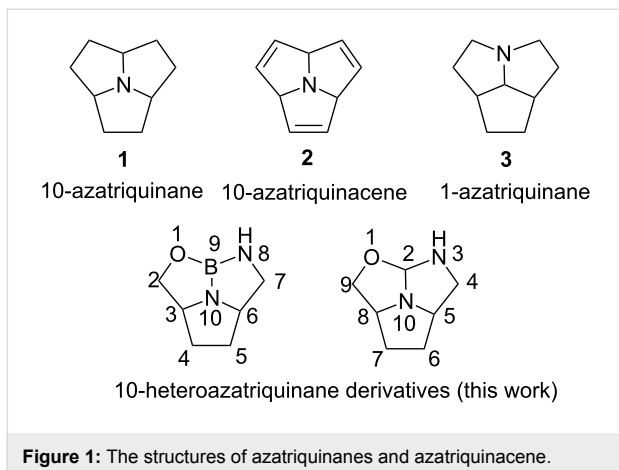
A series of chiral 10-heteroazatriquinanes were synthesized from enantiopure asymmetric *cis*-2,5-disubstituted pyrrolidines through a one-pot tandem cyclization procedure. The structures and configurations of these new chiral 10-heteroazatriquinanes are confirmed by X-ray single-crystal diffraction analysis.

## Introduction

The azatriquinane derivatives are an important substance class in organic chemistry containing nitrogen and three fused five-membered rings [1-3]. Due to the unique rigid bowl-shaped structure with one noninvertible electron lone pair at the bottom of the central nitrogen (“*centro*-N”) atom, 10-azatriquinane analogues are used as efficient chelation reagents of metal cations [4,5]. Mascal and colleagues [6] described the first synthesis of 10-azatriquinane (**1**) from dimethyl 3,3'-(1*H*-pyrrole-2,5-diyl)dipropionic acid in five steps, and the reactivity of **1** was also investigated. 10-azatriquinane (**1**) is very active because of its high basicity. The X-ray structure of **1**·HBF<sub>4</sub> revealed that the *centro*-N is pyramidalized. 10-azatriquinacene (**2**), an unsaturated analogue of **1**, was attractive due to its high

proton affinity [7]. Recently, Mascal [8] and colleagues have developed a series of 10-azatriquinanes as a C<sub>3v</sub>-symmetric platform for tripodal metal complexes and calixiform scaffolds (Figure 1).

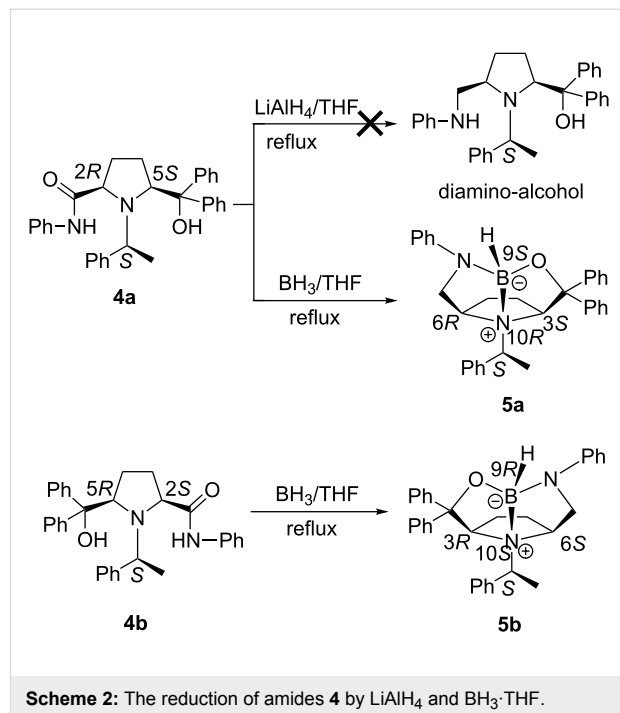
Previously, we established a facile access to enantiopure asymmetric *cis*-2,5-disubstituted pyrrolidines **4** from commercially available starting materials diethyl *meso*-2,5-dibromoadipate and (*S*)-(-)-1-phenylethylamine (Figure 2) [9]. The preparations of compounds **5a**, **5b**, **6a** and **6b** were also reported in [9], but our previously published structures for compounds **5a** and **5b** were not completely correct, because the B–N dative bonds were missing. The aim of the following procedures was to



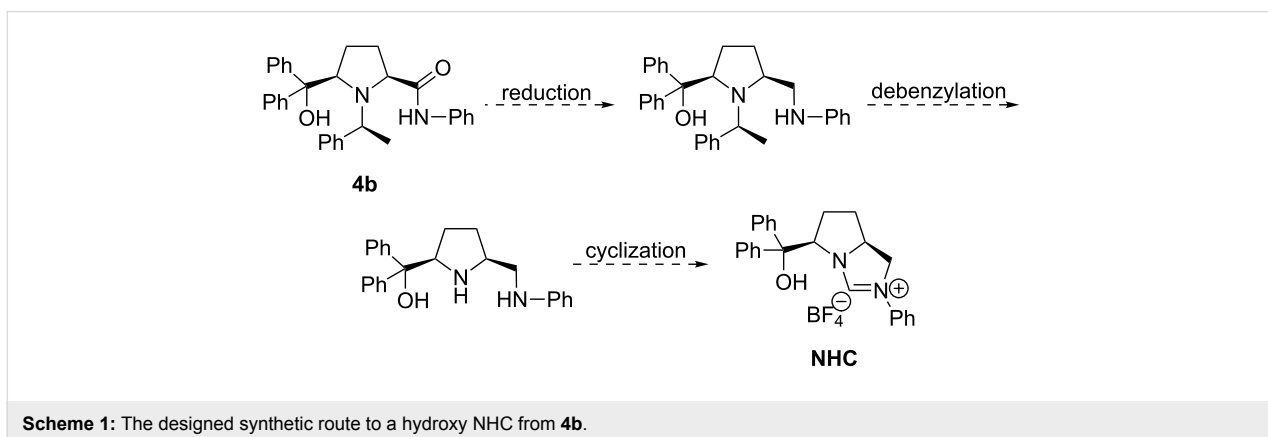
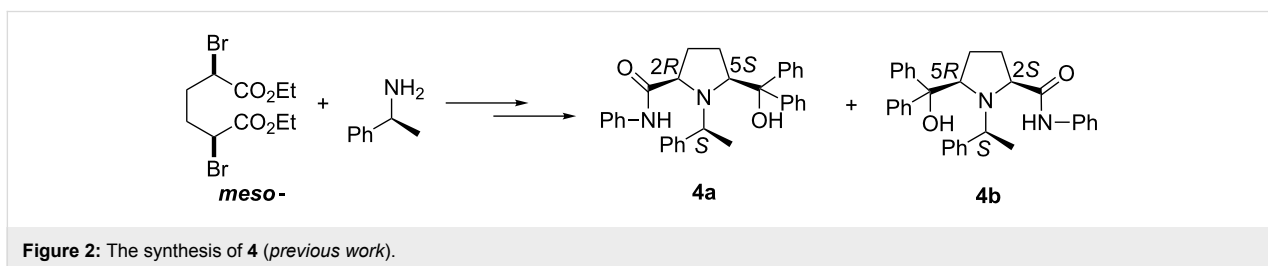
obtain several novel bifunctional *N*-heterocyclic carbenes (NHCs) [10–14] from compounds **4** through three steps, including reduction, debenzoylation and cyclization (Scheme 1), but this failed. However, we have found a novel one-pot tandem cyclization of these enantiopure asymmetric *cis*-2,5-disubstituted pyrrolidines to produce chiral trisubstituted 10-heteroazatriquinane derivatives. To the best of our knowledge, there is very little research on the synthesis of 10-heteroazatriquinanes, and the chiral 10-heteroazatriquinanes are unknown up to now.

## Results and Discussion

For the reduction of highly hindered amides **4**, the system of  $\text{LiAlH}_4$  in anhydrous THF was found to be useless even upon



heating under reflux for 18 h under an inert atmosphere, and no desired diamino-alcohols were obtained after workup. However, **4a** reacted smoothly with  $\text{BH}_3$  to give white crystals after chromatographic purification (Scheme 2). The X-ray single-crystal diffraction analysis established that the reduction product **5a** (Figure 3) was formed with a rigid 10-heteroaza-



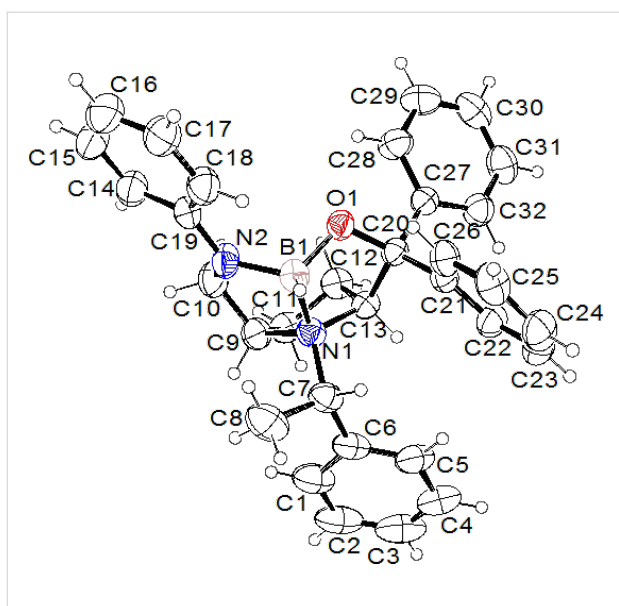


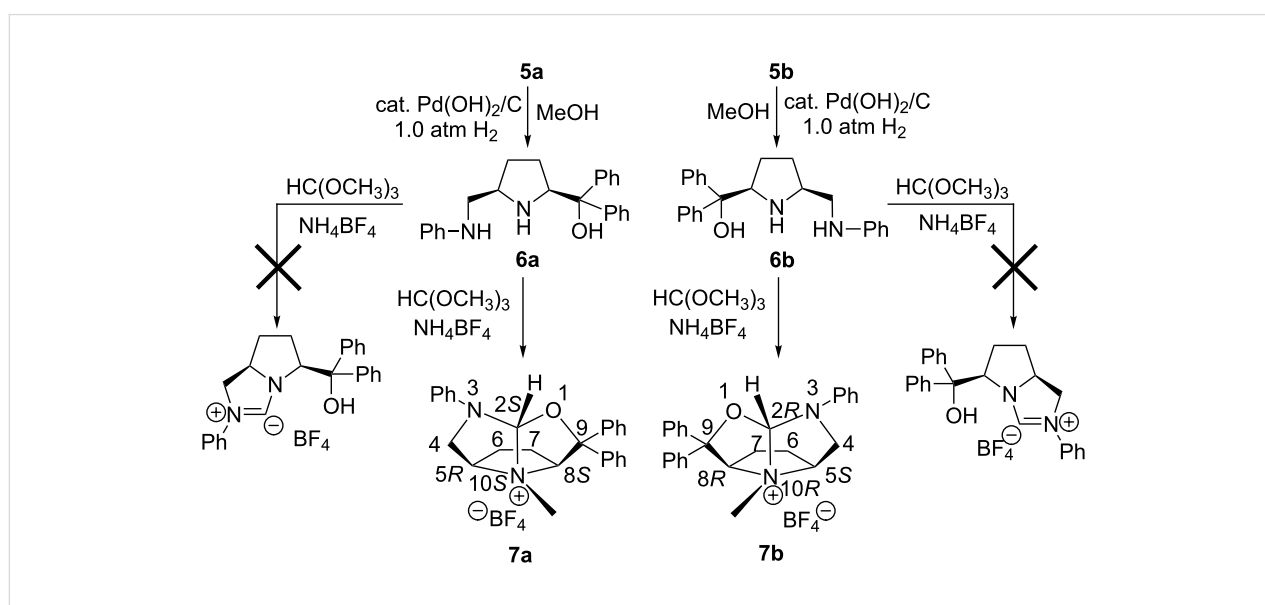
Figure 3: X-ray crystal structure of compound **5a**.

quinane skeleton through an intramolecular Lewis acid–base pair interaction [15–18]. Three five-membered rings are fused to give a very stable bowl-shaped tricyclic system with five stereogenic centers, especially for one chiral nitrogen center and one chiral boron center. The configurations of the stereogenic centers in **5a** are deduced from the configuration of the chiral auxiliary (*S*)-(-)-1-phenylethylamine to be *S*, 3*S*, 6*R*, 9*S* (chiral B atom) and 10*R* (chiral N atom), respectively. This novel tandem reduction/cyclization was made possible by the *cis*-configuration of the starting 2,5-disubstituted pyrrolidine **4a**. Following the same procedure, 10-heteroazaquinane **5b** was

obtained in good yield (87%), and the configurations of **5b** were assigned to be *S*, 3*R*, 6*S*, 9*R* (chiral B atom) and 10*S* (chiral N atom), respectively. Compounds **5a** and **5b** are diastereomers.

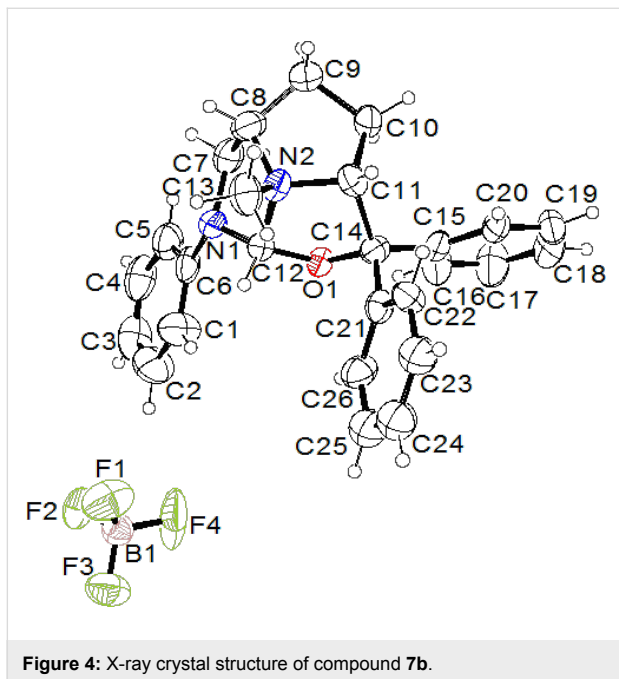
In the presence of a catalytic amount of Pd(OH)<sub>2</sub>/C, under 1.0 atm of hydrogen, the above 10-heteroazaquinane derivatives **5** were converted into enantiopure asymmetric *cis*-2,5-disubstituted pyrrolidines **6** in good yields with a diamino-alcohol skeleton. In this process, both *N*-debenzylation and a ring-opening reaction occurred. Diamino-alcohols **6a** and **6b** are enantiomers, and they can serve as precursors for the synthesis of hydroxy *N*-heterocyclic carbenes. Following the reported procedure for the preparation of *N*-heterocyclic carbenes [19,20], the enantiopure pyrrolidine **6b** and NH<sub>4</sub>BF<sub>4</sub> in HC(OCH<sub>3</sub>)<sub>3</sub> is heated to 80 °C for 2 h, the light yellow crystals were obtained in good yield after workup (Scheme 3). The same result was obtained by heating the mixture of **6b**, NH<sub>4</sub>BF<sub>4</sub> and HC(OCH<sub>3</sub>)<sub>3</sub> in anhydrous toluene under reflux.

To our delight, a single crystal was grown from CH<sub>2</sub>Cl<sub>2</sub>, suitable for X-ray diffraction analysis. It was found that the ring-closing reaction took place during the heating process following *N*-methylation to provide the rigid 1-oxo-3-aza-10-azaquinane skeleton **7b** as its ammonium salt. Compound **7b** contains four stereogenic centers, and their configurations are assigned to be 2*R*, 5*S*, 8*R* and 10*R* (chiral N atom) based on its starting material **4b** (Figure 4). Actually, HC(OCH<sub>3</sub>)<sub>3</sub> as an efficient reagent for *C*-, *N*-, and *O*-methylation has been reported [21–24], but the mechanism of these methylations is elusive. The other chiral ammonium salt **7a** was obtained under the same



Scheme 3: One-pot tandem cyclization of **6** in the presence of HC(OCH<sub>3</sub>)<sub>3</sub>.

conditions as those for the preparation of **7b**. The chiral ammonium salts **7a** and **7b** were derived from the enantiomers **6a** and **6b**, therefore, **7a** and **7b** are also enantiomers. The configurations of **7a** are assigned to be 2*S*, 5*R*, 8*S* and 10*S* (chiral N atom).

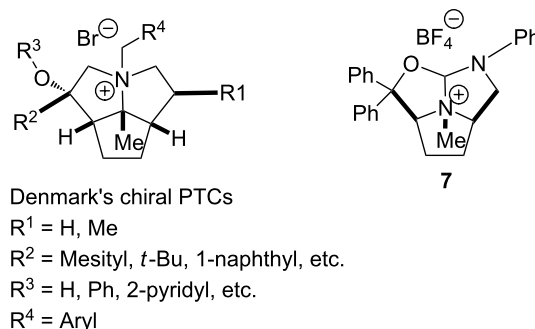


**Figure 4:** X-ray crystal structure of compound **7b**.

As shown in Figure 4, this 10-heteroazatriquinane **7b** possesses a rigid bowl-like molecular scaffold with a quaternary nitrogen site (N2) at the bottom of the cavity. Recently, Denmark and colleagues [25,26] have synthesized a series of chiral phase-transfer catalysts (chiral PTCs) based on a 2-azatriquinane skeleton (Figure 5), and they have investigated their catalytic activities in asymmetric alkylation reactions for producing enantiomerically enriched amino acids. The synthesis of these quaternary ammonium ions follows a diversity-oriented approach wherein the tandem inter-[4 + 2]/intra-[3 + 2] cycloaddition of nitroalkenes serves as the key transformation. The chiral ammonium salts **7** have a structure similar to the chiral PTCs of Denmark et al. The substituents in the chiral ammonium salts **7** are easily tunable, which could open a route to various chiral PTCs for organic synthesis.

## Conclusion

In summary, we provide here a facile access to chiral 10-heteroazatriquinanes from enantiopure asymmetric *cis*-2,5-disubstituted pyrrolidines through one-pot tandem cyclization reactions, and their configurations are confirmed by X-ray single-crystal diffraction analysis. The applications of these novel 10-heteroazatriquinanes are currently being investigated in our laboratory.



**Figure 5:** The structural comparison of chiral ammonium salts such as **7** with the chiral PTCs of Denmark et al.

## Supporting Information

### Supporting Information File 1

Full experimental details, analytical data and crystallographic information.

[<http://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-9-32-S1.pdf>]

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## References

- Mascal, M.; Lera, M.; Blake, A. J. *J. Org. Chem.* **2000**, *65*, 7253–7255. doi:10.1021/jo005571o
- Mascal, M.; Hafezi, N.; Meher, N. K.; Fettingner, J. C. *J. Am. Chem. Soc.* **2008**, *130*, 13532–13533. doi:10.1021/ja805686u
- Mascal, M.; Hext, N. M.; Shishkin, O. V. *Tetrahedron Lett.* **1996**, *37*, 131–134. doi:10.1016/0040-4039(95)02091-8
- Jiao, H. J.; Halet, J.-F.; Gladysz, J. A. *J. Org. Chem.* **2001**, *66*, 3902–3905. doi:10.1021/jo001800v
- Gussenhoven, E. M.; Jevric, M.; Olmstead, M. M.; Fettingner, J. C.; Mascal, M.; Balch, A. L. *Cryst. Growth Des.* **2009**, *9*, 1786–1792. doi:10.1021/cg800906x
- Hext, N. M.; Hansen, J.; Blake, A. J.; Hibbs, D. E.; Hursthouse, M. B.; Shishkin, O. V.; Mascal, M. *J. Org. Chem.* **1998**, *63*, 6016–6020. doi:10.1021/jo980788s
- Mascal, M. *J. Org. Chem.* **2007**, *72*, 4323–4327. doi:10.1021/jo070043z
- Jevric, M.; Zheng, T.; Meher, N. K.; Fettingner, J. C.; Mascal, M. *Angew. Chem., Int. Ed.* **2011**, *50*, 717–719. doi:10.1002/anie.201006470

9. Wang, P.-A.; Xu, Z.-S.; Chen, C.-F.; Gao, X.-G.; Sun, X.-L.; Zhang, S.-Y. *Chirality* **2007**, *19*, 581–588. doi:10.1002/chir.20424
10. He, L.; Zhang, Y. R.; Huang, X. L.; Ye, S. *Synthesis* **2008**, *17*, 2825–2829. doi:10.1055/s-2008-1067216
11. Lv, H.; Chen, X.-Y.; Sun, L.-h.; Ye, S. *J. Org. Chem.* **2010**, *75*, 6973–6976. doi:10.1021/jo101318u
12. Huang, X.-L.; Chen, X.-Y.; Ye, S. *J. Org. Chem.* **2009**, *74*, 7585–7587. doi:10.1021/jo901656q
13. Zhang, Y.-R.; He, L.; Wu, X.; Shao, P.-L.; Ye, S. *Org. Lett.* **2008**, *10*, 277–280. doi:10.1021/ol702759b
14. Wang, X.-N.; Shao, P.-L.; Lv, H.; Ye, S. *Org. Lett.* **2009**, *11*, 4029–4031. doi:10.1021/ol901290z
15. Zhu, L.; Shabbir, S. H.; Gray, M.; Lynch, V. M.; Sorey, S.; Anslyn, E. V. *J. Am. Chem. Soc.* **2006**, *128*, 1222–1232. doi:10.1021/ja055817c
16. Tsurumaki, E.; Saito, S.; Kim, K. S.; Lim, J. M.; Inokuma, Y.; Kim, D.; Osuka, A. *J. Am. Chem. Soc.* **2008**, *130*, 438–439. doi:10.1021/ja078042b
17. Stepanenko, V.; De Jesús, M.; Correa, W.; Bermúdez, L.; Vázquez, C.; Guzmán, I.; Ortiz-Marciales, M. *Tetrahedron: Asymmetry* **2009**, *20*, 2659–2665. doi:10.1016/j.tetasy.2009.11.009
18. Montalbano, F.; Candeias, N. R.; Veiros, L. F.; André, V.; Duarte, M. T.; Bronze, M. R.; Moreira, R.; Gois, P. M. *Org. Lett.* **2012**, *14*, 988–991. doi:10.1021/ol203224n
19. Funk, T. W.; Berlin, J. M.; Grubbs, R. H. *J. Am. Chem. Soc.* **2006**, *128*, 1840–1846. doi:10.1021/ja055994d
20. Van Veldhuizen, J. J.; Gillingham, D. G.; Garber, S. B.; Kataoka, O.; Hoveyda, A. H. *J. Am. Chem. Soc.* **2003**, *125*, 12502–12508. doi:10.1021/ja0302228
21. Janin, Y. L.; Huel, C.; Flad, G.; Thiroit, S. *Eur. J. Org. Chem.* **2002**, 1763–1769. doi:10.1002/1099-0690(200206)2002:11<1763::AID-EJOC1763>3.0.CO;2-Q
22. Selva, M.; Tundo, P. *J. Org. Chem.* **1998**, *63*, 9540–9544. doi:10.1021/jo980914s
23. Padmanabhan, S.; Reddy, N. L.; Durant, G. J. *Synth. Commun.* **1997**, *27*, 691–699. doi:10.1080/00397919708003343
24. Kumar, H. M. S.; Reddy, B. V. S.; Mohanty, P. K.; Yadav, J. S. *Tetrahedron Lett.* **1997**, *38*, 3619–3622. doi:10.1016/S0040-4039(97)00684-9
25. Denmark, S. E.; Gould, N. D.; Wolf, L. M. *J. Org. Chem.* **2011**, *76*, 4260–4336. doi:10.1021/jo2005445
26. Denmark, S. E.; Gould, N. D.; Wolf, L. M. *J. Org. Chem.* **2011**, *76*, 4337–4357. doi:10.1021/jo2005457

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