

# An easy synthesis of 5-functionally substituted ethyl 4-amino-1-aryl- pyrazolo-3-carboxylates: interesting precursors to sildenafil analogues

Said A. S. Ghozlan<sup>\*1</sup>, Khadija O. Badahdah<sup>2</sup> and Ismail A. Abdelhamid<sup>\*1</sup>

## Full Research Paper

Open Access

### Address:

<sup>1</sup>Department of Chemistry, Faculty of Science, Cairo University, Giza, A. R. Egypt and <sup>2</sup>Department of Chemistry, Faculty of Science, King AbdulAziz University, Jeddah-21411. P.O. Box 154, Saudi Arabia

### Email:

Said A. S. Ghozlan<sup>\*</sup> - s\_ghozlan@yahoo.com; Khadija O. Badahdah - kbadahdah@yahoo.com; Ismail A. Abdelhamid<sup>\*</sup> - ismail\_shafy@yahoo.com

\* Corresponding author

*Beilstein Journal of Organic Chemistry* **2007**, 3, No. 15.

doi:10.1186/1860-5397-3-15

Received: 28 January 2007

Accepted: 01 May 2007

Published: 01 May 2007

© 2007 Ghozlan et al; licensee Beilstein-Institut.

License and terms: see end of document.

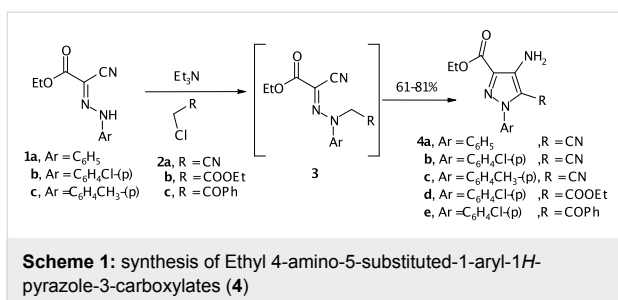
## Abstract

3-Oxo-2-arylhydrazononitriles **1a-c** react readily with chloroacetonitrile, ethyl chloroacetate, and with phenacyl chloride to give 4-aminopyrazoles **4a-e**. The pyrazolo[4,3-*d*]pyrimidine derivatives **7** and **10** are synthesized via reaction of the aminopyrazole **4b** with phenylisothiocyanate and DMFDMA/NH<sub>4</sub>OAc respectively.

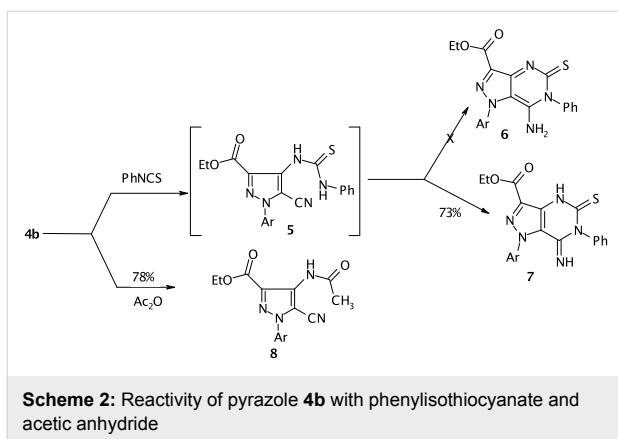
## Background

Interest in the chemistry of 4-aminopyrazole carboxylic acid derivatives has recently been recognized as their derivatives are ideal precursors for the synthesis of biologically active pyrazolo[4,3-*d*]pyrimidine ring systems [1-6]. The reported synthetic approaches to these derivatives are also multistep, non atom economical and non eco friendly [1,5,6]. Recently however a route to 4-aminopyrazole-5-carboxylic acid derivatives via reacting 2-arylhydrazononitriles with  $\alpha$ -haloacid derivatives has been reported by Elnagdi et al [7,8] as well as other researchers [9]. In the present article we report results of our work aimed at exploring this synthetic methodology and adoption of products for the synthesis of pyrazolo[4,3-*d*]pyrimidines.

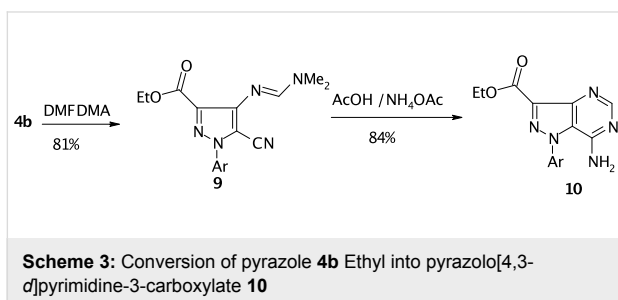
Thus, compounds **1a-c**, were prepared according to literature procedures via coupling of ethyl cyanoacetate with aromatic diazonium salts [10]. It has been found that **1a-c** react with  $\alpha$ -chloroacetonitrile **2a** to yield **4a-c**, most likely via acyclic intermediates **3a-c** that could not be isolated. The structure of **4a-c** was confirmed based on <sup>1</sup>H NMR spectra that revealed the presence of amino signals and also <sup>13</sup>C NMR which revealed the presence of only one CN signal. Similarly reacting **1b** with ethyl chloroacetate **2b** and with phenacyl chloride **2c** afforded **4d,e**. The structure of **4d,e** was also confirmed based on IR and <sup>13</sup>C NMR, which revealed the absence of CN bands and signals (cf. Scheme 1).



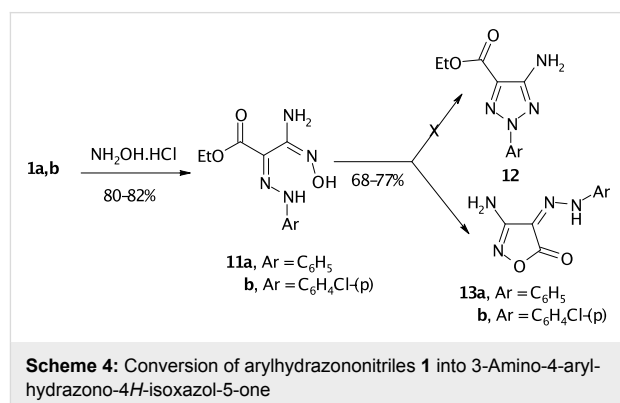
Compound **4b** reacted readily with phenylisothiocyanate to yield a 1:1 adduct. The IR and  $^{13}\text{C}$  NMR spectra of the product revealed the absence of CN bands and signals. Thus structure **6** or **7** is suggested.  $^1\text{H}$  NMR showed two NH signals at  $\delta$  8.33 and 10.3 ppm, thus structure **7** is assigned for the reaction product. Acetylation of **4b** in acetic anhydride afforded mono-acetyl derivative **8**. (cf. Scheme 2)



Compound **4b** condensed with dimethylformamide dimethylacetal (DMFDMA) to yield the enamine **9**. The  $^1\text{H}$  NMR spectrum indicated two distinct singlets at  $\delta$  2.97 and 3.05 ppm for the *N,N*-dimethylamino protons which mean that the two methyl groups are magnetically nonequivalent, as to be expected. Compound **9** could be readily converted into pyrazolo[4,3-*d*]pyrimidine **10** on treatment with AcOH/ $\text{NH}_4\text{OAc}$  mixture. (cf. Scheme 3)



Compound **1** reacted with hydroxylamine hydrochloride in ethanol/sodium acetate solution to yield amidooxime **11** as in the literature [10]. Trials to cyclize the amidooxime into 1,2,3-triazole **12** utilizing the reaction conditions described earlier in literature [11] failed. However, the amidooxime **11** cyclizes smoothly via loss of ethanol in DMF and in presence of anhydrous sodium acetate into isoxazolone **13**. (cf. Scheme 4)



## Conclusion

We could show that arylhydrazononitriles **1a-c** are valuable precursors to 4-amino-5-substituted-1-aryl-1*H*-pyrazole-3-carboxylic acid ethyl ester which can be used for preparation of sildenafil analogues.

## Supporting Information

### Supporting Information File 1

The experimental section. The experimental data and the results of analysis

[<http://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-3-15-S1.doc>]

## References

- Haning, H.; Niewöhner, U.; Schenke, T.; Lampe, T.; Hillisch, A.; Bischoff, E. *Bioorg. Med. Chem. Lett.* **2005**, *15*, 3900–3907. doi:10.1016/j.bmcl.2005.05.090
- Kim, D.-K.; Lee, J. Y.; Lee, N.; Ryu, D. H.; Kim, J.-S.; Lee, S.; Choi, J.-Y.; Ryu, J.-H.; Kim, N.-H.; Im, G.-J.; Choi, W.-S.; Kim, T.-K. *Bioorg. Med. Chem.* **2001**, *9*, 3013–3021. doi:10.1016/S0968-0896(01)00200-0
- El Haddad, M.; Soukri, M.; Lazar, S.; Bennamara, A.; Guillaumet, G.; Akssira, M. *J. Heterocycl. Chem.* **2000**, *37*, 1247–1252.
- Holla, B. S.; Mahalinga, M.; Karthikeyan, M. S.; Akberali, P. M.; Shetty, N. S. *Bioorg. Med. Chem.* **2006**, *14*, 2040–2047. doi:10.1016/j.bmc.2005.10.053
- Zhao, Y.-f.; Zhai, X.; Chen, J.-y.; Guo, S.-c.; Gong, P. *Chem. Res. Chin. Univ.* **2006**, *22*, 468–473. doi:10.1016/S1005-9040(06)60144-X

6. Carpino, P. A.; Griffith, D. A.; Sakya, S.; Dow, R. L.; Black, S. C.; Hadcock, J. R.; Iredale, P. A.; Scott, D. O.; Fichtner, M. W.; Rose, C. R.; Day, R.; Dibrino, J.; Butler, M.; DeBartolo, D. B.; Dutcher, D.; Gautreau, D.; Lizano, J. S.; O'Connor, R. E.; Sands, M. A.; Kelly-Sullivan, D.; Ward, K. M. *Bioorg. Med. Chem.* **2006**, *16*, 731–736. doi:10.1016/j.bmcl.2005.10.019
7. Abdel-Motaleb, R. M.; Makhloof, A.-M. A.; Ibrahim, H. M.; Elnagdi, M. H. *J. Heterocycl. Chem.* **2006**, *43*, 931–934.
8. Abdel-Motaleb, R. M.; Makhloof, A.-M. A.-S.; Ibrahim, H. M.; Elnagdi, M. H. *J. Heterocycl. Chem.* **2007**, *44*, 109–114.
9. Salaheldin, A. M.; Abdallah, T. A.; Radwan, N. F.; Hassaneen, H. M. Z. *Naturforsch.* **2007**, in press.
10. Elnagdi, M. H.; Elmoghayar, M. R. H.; Hafez, E. A. A.; Alnima, H. H. *J. Org. Chem.* **1975**, *40*, 2604–2607. doi:10.1021/jo00906a007
11. Ghozlan, S. A. S.; Abdelhamid, I. A.; Ibrahim, H. M.; Elnagdi, M. H. *ARKIVOC* **2007**, No. xv, 53–60.

## License and Terms

This is an Open Access article under the terms of the Creative Commons Attribution License (<http://creativecommons.org/licenses/by/2.0>), which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.

The license is subject to the Beilstein Journal of Organic Chemistry terms and conditions: (<http://www.beilstein-journals.org/bjoc>)

The definitive version of this article is the electronic one which can be found at:  
doi:10.1186/1860-5397-3-15